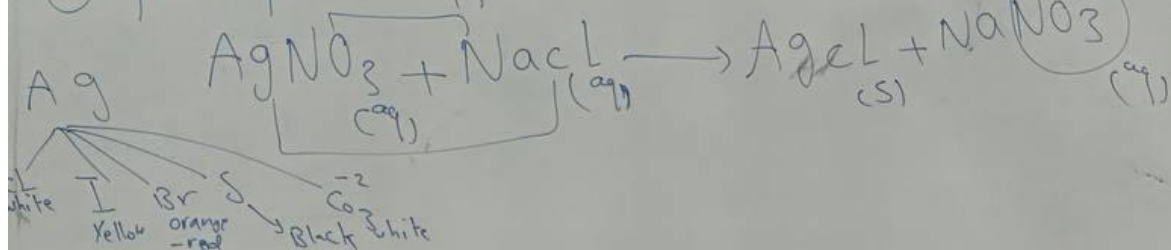


① chemical properties:



evidence for chemical rxns:-

① precipitate appear or disappear.



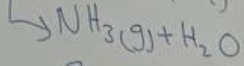
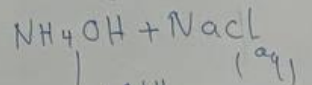
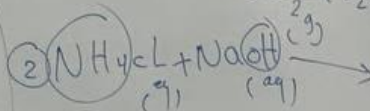
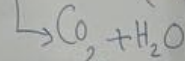
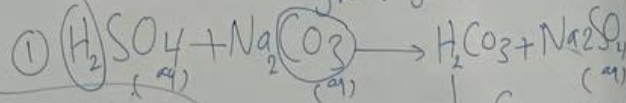
② color change.

③ heat observed

exothermic rxn.
endothermic rxn.

④ gas evolved

g_o: gas with odor
g_{no}: gas no odor.



g_o

Handwritten notes on a piece of paper, partially visible on the left side of the table.

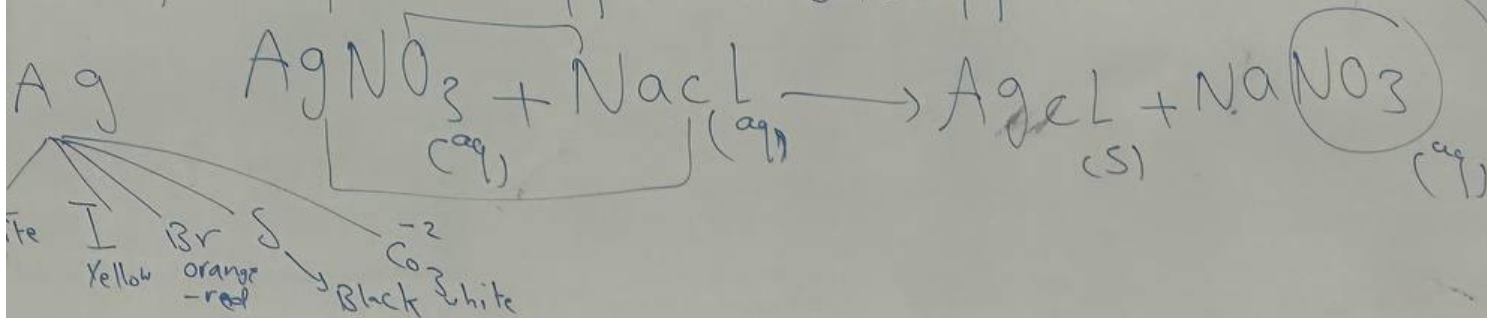


① chemical properties:



evidence for chemical rxns:-

① precipitate appear or disappear.



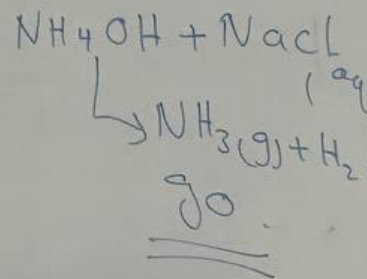
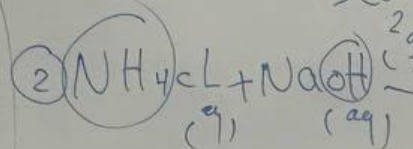
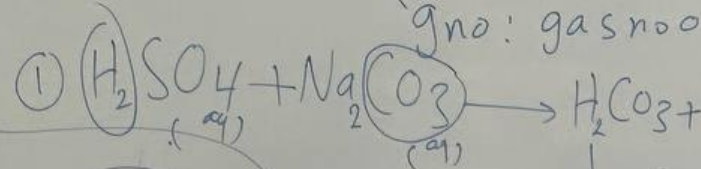
② color change.

③ heat observed

exothermic
endothermic

④ gas evolved

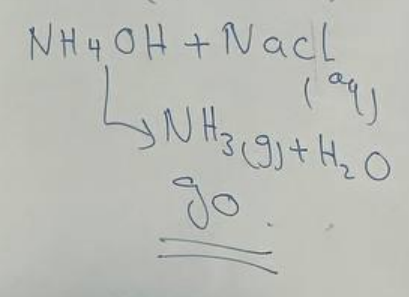
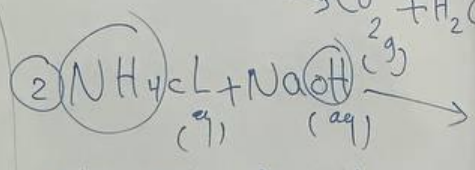
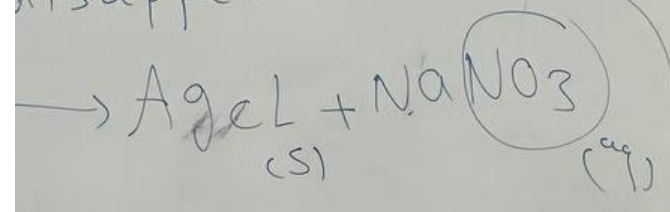
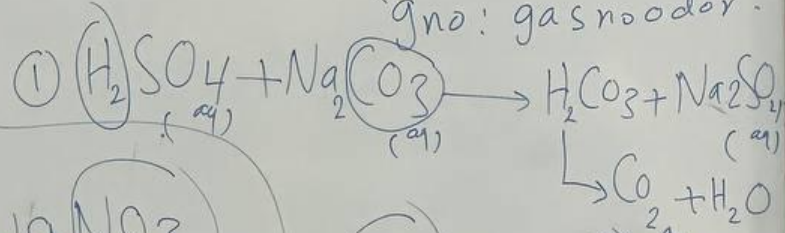
go: gas with
gno: gas no



(3) heat observed ← endothermic rxn.

(4) gas evolved ← go: gas with odor
gno: gas no odor.

s:-
disappear.



$\text{AgNO}_3 \rightarrow \text{Ag}_2\text{SO}_4 + \text{NaCl}$ ppt, mP: milky
 C: cloudy
 gp: granular
 ZnSO_4 unknown NaCl
 go: gas with odor
 gno: gas no odor
 N.R: no reaction
 N.R ppt
 C
 Zn(OH)_2 N.R
 N.R
 N.R

Tech n for odor: -
 Fan The vapor
 gently toward the
 nose

