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Reactions of Organic Compounds (Outlines)

I. Reactions of Hydrocarbons

- A. Substitution Reactions of alkanes to give alkyl halides.
- B. Combustion Reactions
- C. Addition Reactions on alkenes and alkynes.
 - i) <u>Hydrogenation</u> to form <u>alkanes</u>.
 - ii) <u>Halogenation</u> to form <u>alkyl halide</u>.
 - iii) Addition of a halogen acid (HF, HCl, HBr, HI) to produce alkyl halide.
- iv) Addition of water (Hydration) to produce <u>alcohol</u>.
- D. Aromatic substitution reaction.

II. Reactions of organic compounds that contain other functional groups:

- 1. Reactions of Alcohols:
 - A. <u>Oxidation</u> to give aldehydes and ketones.
 - B. <u>Dehydration</u> to give alkenes.
 - C. <u>Substitution Reactions</u> to give alkyl halides.
 - D. <u>Esterification</u> to give esters.
- 2. Reactions of Ethers:

Cleavage reaction to give alkyl halide and/or alcohol.

- 3. Reactions of Aldehydes and Ketones:
 - A. <u>Reduction</u> of aldehydes to give primary alcohol s. <u>Reduction</u> of ketone to give secondary alcohols.
 - B. <u>Oxidation</u> of aldehydes to give carboxylic acids.
 - C. <u>Hydration</u> (addition of Water) to give two hydroxyl groups on the same carbon.
 - D. Addition of Alcohol to produce Acetals and hemiacetals.
 - E. Addition of Hydrogen Cyanide to produce cyanohydrin.
 - F. Addition of ammonia and ammonia derivatives to give imines.
- 4. Reactions of Carboxylic acids:
 - A. Reaction with thionyl chloride to give <u>acid chloride</u>.
 - B. Reaction with acid chloride to produce <u>acid anhydride</u>.
 - C. Reaction with alcohols to give ester.
 - D. Reaction with amine to produce an <u>amide</u>:
- 5. Reactions of Esters:

Cleaved reaction to give carboxylic acid and alcohol.

Introduction to Organic Chemistry Part Three: Preparation of Organic Compounds

Important remarks on the preparation of different organic compounds as inferred from the reactions mentioned in part II

Preparation of alkyl halides:

Alkyl halides are prepared by:

- 1. Substitution reaction of halogens on saturated hydrocarbons.
- 2. Substitution reaction of HX on primary alcohols.
- 3. Addition reaction of hydrogen halides or halogens on unsaturated hydrocarbons.
- 4. Cleavage reaction of ethers in presence of HX.

Preparation of saturated hydrocarbons:

By addition reaction of hydrogen (hydrogenation) on unsaturated hydrocarbons.

Preparation of alcohols:

Alcohols are prepared by:

- 1. Addition reaction of water on unsaturated hydrocarbons.
- 2. Cleavage reaction of ethers in presence of HX, gives primary alcohols.
- 3. Reduction of aldehydes gives primary alcohols.
- 4. Reduction of ketones secondary alcohols.
- 5. Cleavage reactions of esters.

Preparation of aldehydes:

Aldehydes are prepared by oxidation of primary alcohols.

Preparation of ketones:

Ketones are prepared by oxidation of secondary alcohols.

Preparation of alkenes:

Alkenes and alkynes are prepared by dehydration of alcohols.

Preparation of ethers:

Ethers are prepared by dehydration of primary alcohols.

Preparation of carboxylic acids:

Carboxylic acids are prepared by:

- 1. Cleavage reactions of esters.
- 2. Oxidation of primary alcohols

Preparation of esters:

Esters are prepared by reaction of primary alcohols with carboxylic acids.

Saturated, unsaturated hydrocarbons, alkyl halides, aldehydes, ketones, ethers, carboxylic acids and esters all can be prepared from alcohols.