

رحلة الدواء داخل الجسم

PHARMACOKINETICS

1

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Pharmacology

The science that deals with
drugs.

Drugs

Substances used to prevent and treat diseases.

Drugs

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graph TD; A[Drugs] --> B[Pharmacokinetics]; A --> C[Pharmacodynamics];
```

Pharmacokinetics

**what the
body does
to the
drug?**

Pharmacodynamics

**what the
drug does
in the
body?**

*target
Mechanism
action on body*

Pharmacokinetics



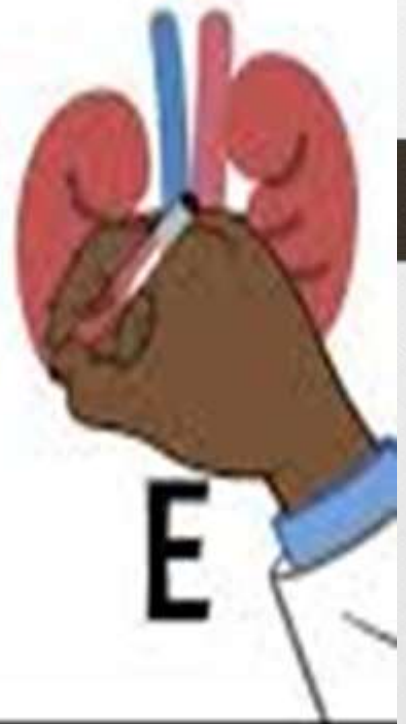
A



D



M



E

Pharmacokinetics

what the body does to the drug?

Absorption *IV* *absorption* *ما عدا* *all drug* *يتوصل الدم*

من الدم للأعضاء Distribution

تغيير الأدوية Metabolism
(استقلاب)

Excretion

elimination

ABSORPTION for any route
except IV

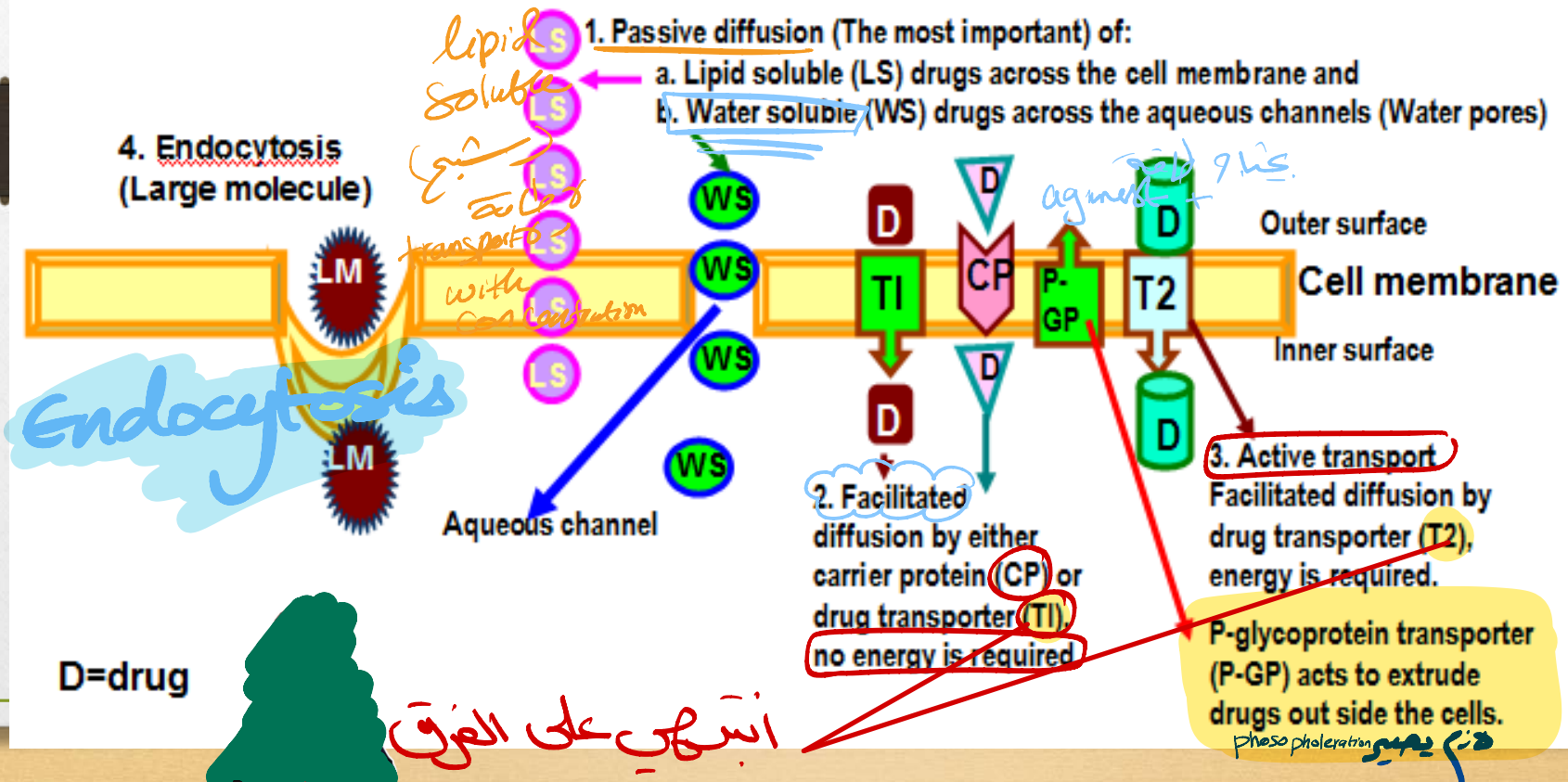
من المكان الذي أعطيت الدواء إلى الدم



محل
شحنة [Cell membrane] is Bilayer of phospholipid

Mechanisms of drug absorption (how drugs cross biological membranes)

Mechanisms of drug movement across the biological membranes



1. Passive diffusion for Both lipid soluble and water soluble

- Rapid movement of lipid soluble drugs across the cell membrane. لا طاقة ولا Carrier وببساطة مع التركيز.
- Movement of the water soluble drugs across the aqueous channels (water pores).
- No energy needed and with concentration gradient.

2. Facilitated diffusion

- The drugs are carried into inside the cell by carrier or transporter.
- No energy is required and according to the concentration gradient

3. Active transport

- The drug movement may be **against the concentration gradient** by drug carrier or transporter.
- Energy is required

energy and carrier are required

4. Endocytosis

example: Heparin

- Drugs of **high molecular weight**, the drug binds to the cell membrane, dips in and **enveloped** by the cell membrane.

Factors affecting absorption:

alveoli have the largest surface area in the body



- Route of **A**dministration
- **A**bsorbing surface
- Co **A**dministration of food or drugs
Shock: when no blood reach tissues] vasoconstriction
Shock → الازمة
→ oral → ينفس الدواء
→ parenteral or intracastial → يعطى الدواء
→ حقن
- **S**ystemic circulation
- **S**pecific factors



- 1- Water & lipid **s**olubility
- 2- Pharmaceutical **p**reparation
- 3- **I**onization of the drugs

A. Factors related to the patient

Route of Administration

I.V. and inhalation > I.M. > S.C. > Oral > Topical

Absorbing surface

- **Vascularity:** (Alveoli > ^{subcutaneous} S.C. tissue).
- **Surface area:** (Alveoli > Intestine > Stomach).
- **Pathological conditions:** Diarrhea decrease oral absorption

Shock ٥٤٤
+ ٥٤٤ (٤ cases) ٥٤٤

Systemic *circulation*

- **Shock** decrease absorption; oral and subcutaneous routes are not suitable. ✕

Specific factors

Intrinsic factor is essential for vitamin B12 absorption.

بعض حالات استئصال المعدة

Co Administration of other drugs & food

beneficial

▶ S.C. adrenaline (added to local anesthetics) → V.C. absorption of local anesthetics → longer duration of action of local anesthetics.

bad

▶ Ca²⁺ (e.g. in milk) ▼ oral absorption of tetracyclines (antibiotics).

tetracycline and (Ca) → دواء مضاد حيوي مع معدن الكالسيوم

B. Factors related to the drug

1- Water and lipid Solubility

- ▶ **Completely water-soluble compounds** are not absorbed (e.g. barium chloride).
- ▶ **increase lipid solubility** lead to increase absorption (lipid/water partition coefficient)

معامل التوزيع الدهون/الماء (Lipid/Water Partition Coefficient) هو مقياس يحدد مدى ذوبانية مادة معينة في الدهون مقارنة بالماء. يُستخدم هذا المعامل لفهم كيفية امتصاص الأدوية عبر الأغشية الحيوية (biological membranes) التي تحتوي على طبقة دهنية.

2- Pharmaceutical preparation

- **Dosage form:** Solution > Suspension > tablet.
- **Shape, size** of particles and rate of dissolution of tablets.
- **Excipient (filler)** containing Ca^{+2} decreases oral absorption of tetracyclines.

شكليه من الصيدلية
شو كان؟

كل كان أقل كان أفضل

مواد ليس لها تأثير علاجي تغطي لزيادة مفعول الدواء.

صحة صيدلانية
أدوية وقت تدقيق
مركباتهم الأولية

دواء معلق

Cell membrane is charged

صفتون
لوحظت ايشي
في صور تشارف
PH

اطكان

action potential

الأحج

bad
uptake

3- Ionization of the drug:

ionized
عئين
unionized
عئين

Ionization decreases lipid solubility and absorption of drugs.

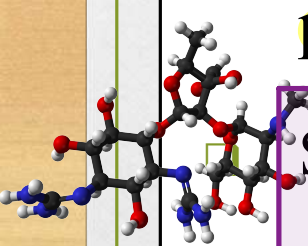
any ionized drug is not absorbed and any nonionized drug is absorbed

Non-ionized (uncharged) → better absorption.

Depends on pKa of the drug and pH of the medium .

NH_4^+
Quaternary ammonium compounds → ionized

poor absorption. بما إنه لا عئين ناأنا أضع الدواء في اهل الذي اريد له ان يصل إليه



Streptomycin has high pKa → always ionized
not absorbed orally.

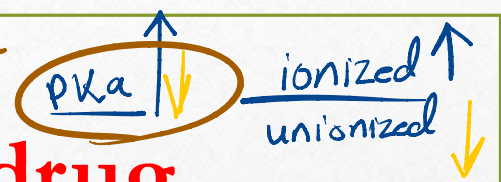
Tertiary amine عئين → adverse in everywhere (and Brain)

يعني دواء pKa = 5.5 بين داخلية في وسط حموضة 3 أفضل من ازا وصحة في الوسط حموضة 4

لو pKa لدواء كانت 5.5 فهذا الوسط ما يفضله من pKa هو الاصل

الدواء الحمضي يحب الوسط القلبي
والدواء القاعدي يحب الوسط القاعدي

كل دواء له
its own
pKa

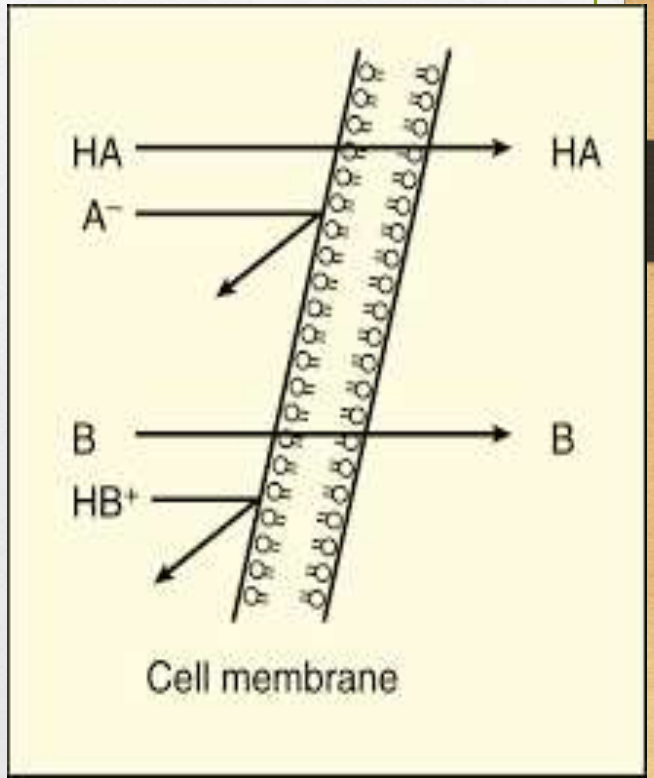


The effect of pH on drug absorption

العقود تحب العقود رحت ان توضع في عقود اعلى منها

When drugs bind hydrogen,

- weak **acids** become **unionized** ($A^- + HA$)
- while weak **base** are **ionized** ($B + BH^+$)





0 1/2 1 1 1/2 2

ال

PKa لَبَن →

pH اَلْمَصْرُوحُو

At low pH weak acids become unionized while the weak bases become ionized.

At high pH weak base drugs become unionized while weak acids become ionized.

- Accordingly, weak acid are more absorbed in acidic media while weak bases are more absorbed in alkaline media.

- The pH at which the concentrations of the ionized and unionized forms of the drug are equal is termed **pKa**.
- Each drug has its own pKa.

$$\frac{\text{ionized}}{\text{unionized}}$$

Clinical importance of pKa

1- GIT:
Aspirin
(acidic drug)
has low pKa.

Drug molecules become unionized in the empty stomach (low pH) and can enter gastric mucosal cells. In gastric mucosal cells (high pH) aspirin becomes ionized and trapped in gastric mucosal cell “peptic ulceration”

2- Kidney: In
drug poisoning,

renal elimination could be enhanced by changing urinary pH to increase ionization of drug and inhibit tubular reabsorption of the drug.

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[Redacted text block]

- **Alkalinization** of urine by **sodium bicarbonate** (to increase urine pH above drug pKa) is useful in acidic drug poisoning e.g. **Aspirin and phenobarbital**.

صنوف جاربیتال
زیادہ

- **Acidification** of urine by **ascorbic acid** (to decrease urine pH below drug pKa) is used in basic drug poisoning e.g. **amphetamine**.

آم الفاسفامین

first pass
metabolism

من یوتیوں: اکثر ما یوتیوں علیہ صو

BIOAVAILABILITY

- It is the percentage of drug that reaches the systemic circulation and becomes available for biological effect.

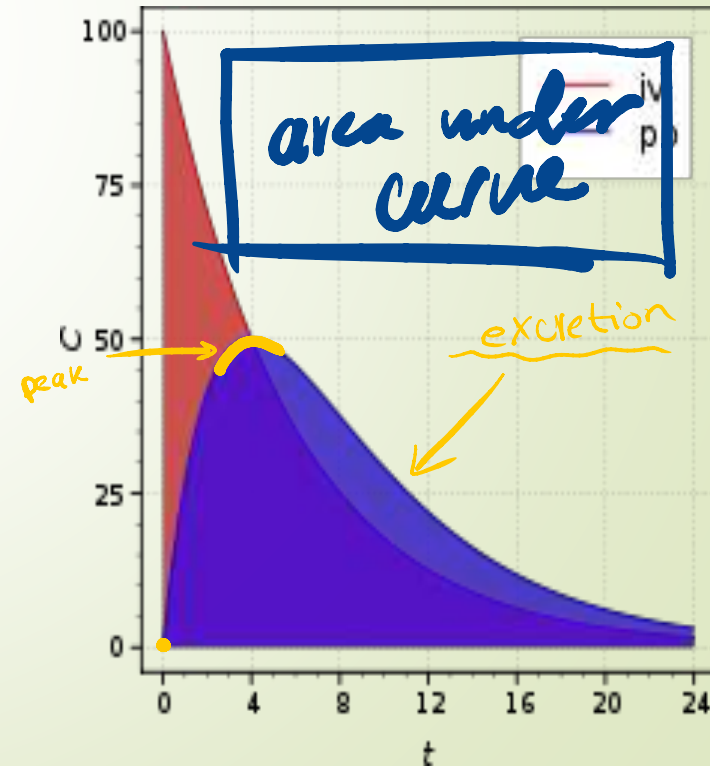
Bioavailability =

Area under the curve (AUC) after oral route

X 100

Area under the curve (AUC) after L.V. route

by IV is always
100%



FACTORS AFFECTING BIOAVAILABILITY:

1- The extent of drug absorption.

مدى الامتصاص

2- 1st pass effect (1st pass metabolism):

التفاعل مع الدواء على انه مادة غريبة لازم يكرها

It is the metabolism of some drugs in a single passage through gut wall, liver or lungs before reaching systemic circulation.

A. Hepatic 1st pass effect:

- Nitroglycerin and propranolol pass from GIT to liver where they are extensively metabolized in their 1st pass through liver before reaching systemic circulation.

B. Intestinal 1st pass effect:

- Estrogens are extensively metabolized in their 1st pass through intestinal wall.

C. Pulmonary metabolism:

- After inhalation, nicotine is partially metabolized in the lung.

A top-down view of a spiral-bound notebook with a white cover and lined pages. The notebook is open to a page with the words "TO BE CONTINUED" written in large, bold, black, sans-serif capital letters. The page is decorated with several small, crumpled pieces of paper in various colors: pink, yellow, green, and orange. A yellow pencil is lying diagonally across the bottom right corner of the notebook. The notebook is placed on a light-colored wooden surface. Two dark grey horizontal bars are visible on the left and right sides of the image, partially overlapping the notebook's edges.

**TO BE
CONTINUED**