**Isomerism**

**Molecular formula:** an expression that shows number and types of atoms that formed the compound.

**Examples:** water H2O

 Methane CH4

 Glucose C6H12O6

The molecular formula does not show how the atoms in the compound connected with each other

**Structural formula**: an expression that shows number, types of atoms that formed the compound and how these atoms connected with each other.







**Abbreviated Structural Formulas**

















