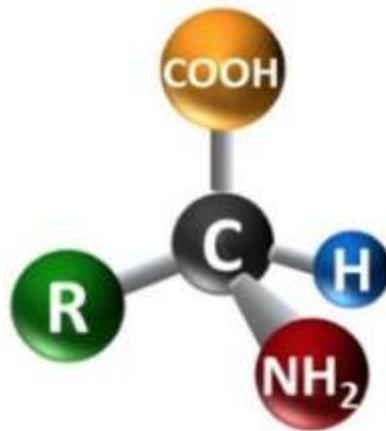




# Amino Acids 1



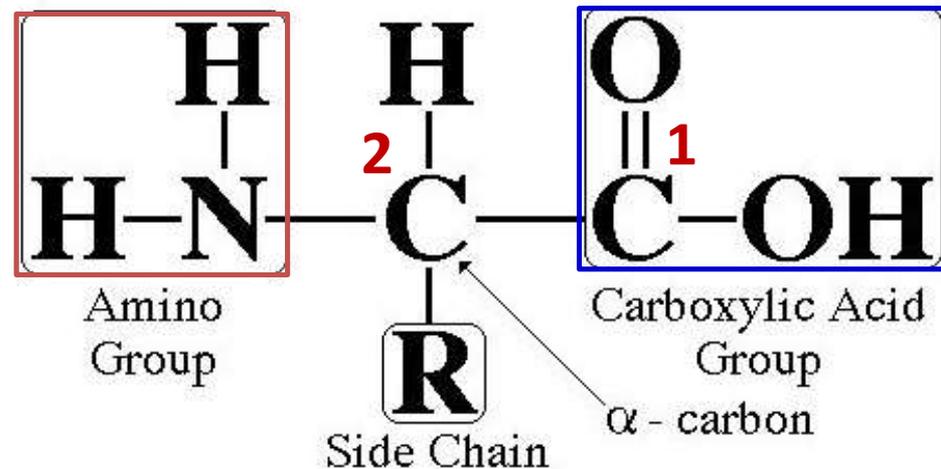
Dr. Nesrin Mwafi

Biochemistry & Molecular Biology Department  
Faculty of Medicine, Mutah University

# Amino Acid Structure



- Amino acids are biologically important organic molecules that contain both **carboxylic acid (-COOH)** as well as **amine (-NH<sub>2</sub>)** groups
- The side-chain also called “**R**” group is specific to each amino acid



- Amino group is attached to  $\alpha$ -carbon (**C2**)
- C, N, O and H are the key elements of amino acids

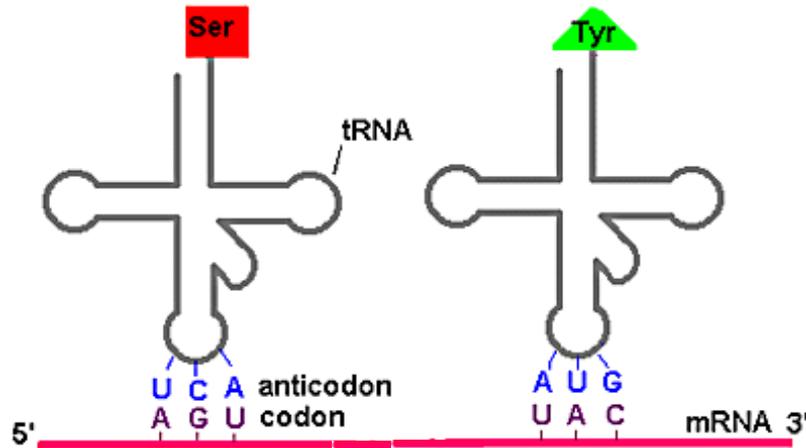
# Biological significance of Amino Acids



1. Amino acids are N-containing molecules
  2. The basic structural building units (monomers) of proteins (**protein role**)
  3. Precursors of many biomolecules like neurotransmitters (**non-protein role**)
  4. They are also utilized as an energy source
- There are **20 standard** (canonical) amino acids which are encoded directly by triplet codons in the universal genetic code during in vivo protein synthesis process (mRNA translation)



# Genetic Code Table



- **The 20 standard** amino acids are known as proteinogenic or natural amino acids

		2nd base in codon					
		U	C	A	G		
1st base in codon	U	Phe Phe Leu Leu	Ser Ser Ser Ser	Tyr Tyr STOP STOP	Cys Cys STOP Trp	U C A G	3rd base in codon
	C	Leu Leu Leu Leu	Pro Pro Pro Pro	His His Gln Gln	Arg Arg Arg Arg	U C A G	
	A	Ile Ile Ile Met	Thr Thr Thr Thr	Asn Asn Lys Lys	Ser Ser Arg Arg	U C A G	
	G	Val Val Val Val	Ala Ala Ala Ala	Asp Asp Glu Glu	Gly Gly Gly Gly	U C A G	

# Standard Amino Acids List

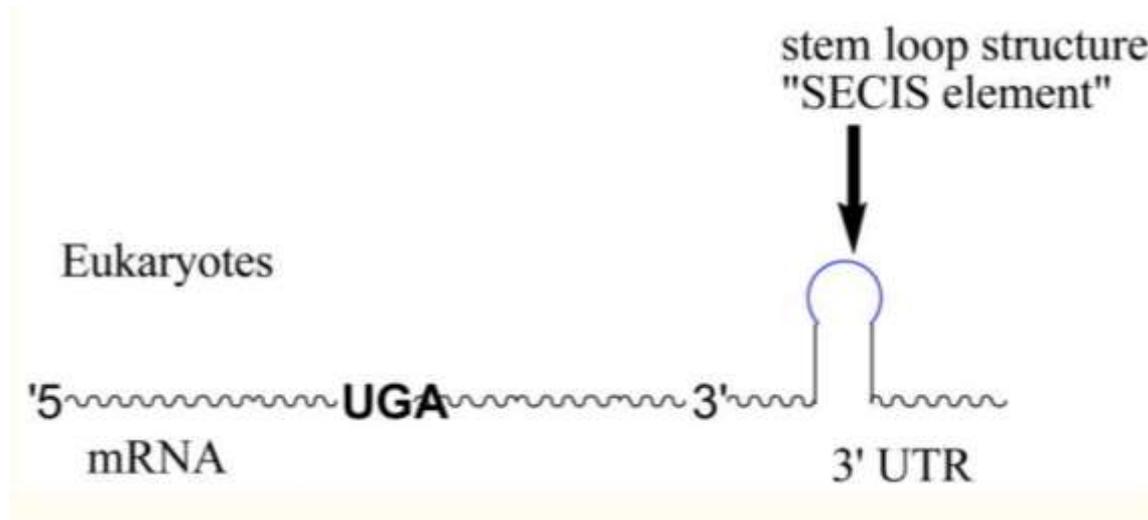


Histidine	Arginine	Alanine
Isoleucine	Asparagine	Asparatate
Leucine	Glutamine	Cysteine
Methionine	Glycine	Glutamate
Phenylalanine	Proline	
Threonine	Serine	
Tryptophan	Tyrosine	
Valine		
Lysine		

# Standard Amino Acids



1. They are proteinogenic and natural amino acids (the other proteinogenic amino acids N-formyl methionine, pyrrolysine and selenocysteine are called non-standard or non-canonical amino acids)

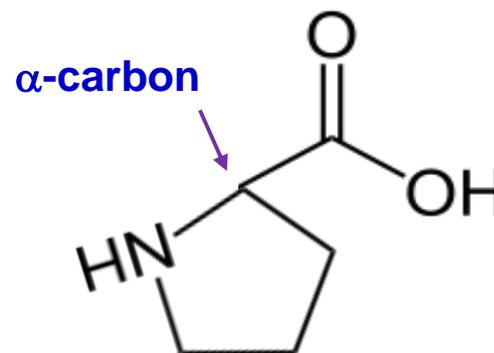
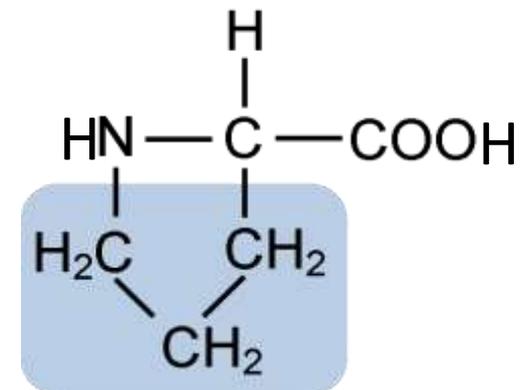
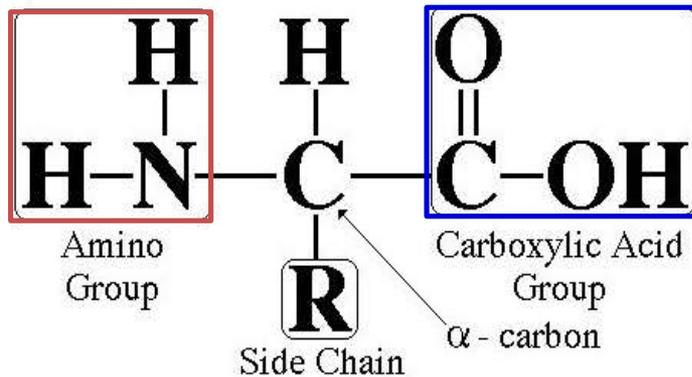


Incorporation of selenocysteine in protein structure by unique mechanism



# Standard Amino Acids

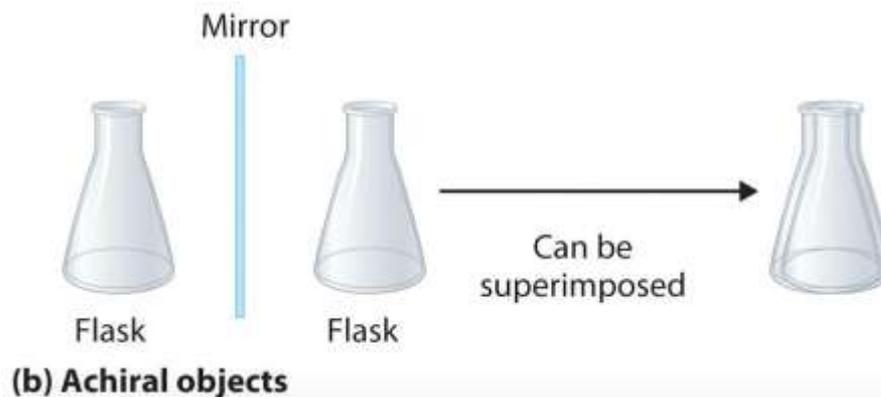
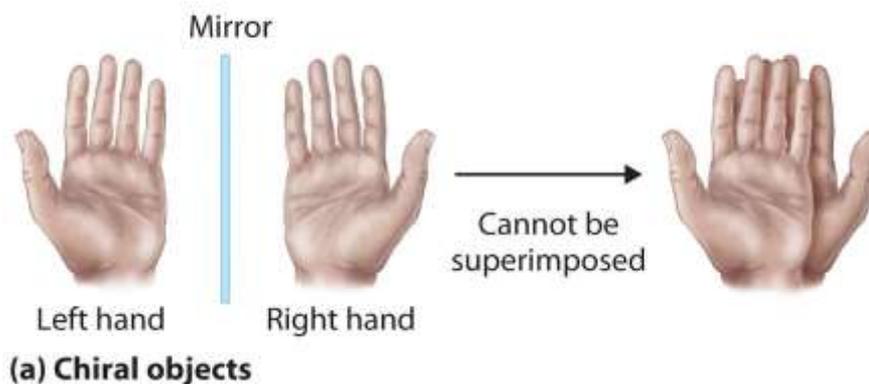
2. Known as 2-, alpha- or  **$\alpha$ -amino acids** as the primary amino group ( $-\text{NH}_2$ ) is attached to  $\alpha$ -carbon (the carbon next to  $-\text{COOH}$  group). Proline is an exception which has a secondary amino group ( $-\text{NH}-$ )



# Standard Amino Acids



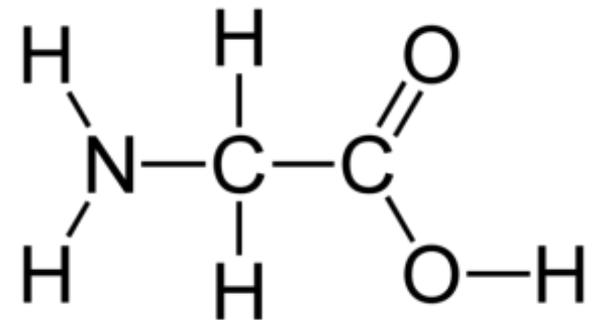
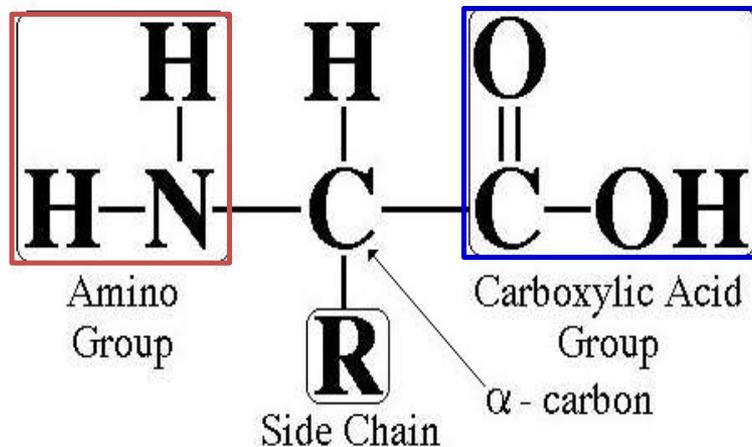
3. They are all chiral molecules (except glycine which has **achiral** C) with **L-** stereochemical configuration (left-handed isomers)



# Standard Amino Acids



- Chiral molecules should contain at least one chiral center (**usually a carbon atom**)
- **Chiral carbon**: asymmetric carbon atom attached to 4 different groups of atoms



**Glycine**

# Isomerization



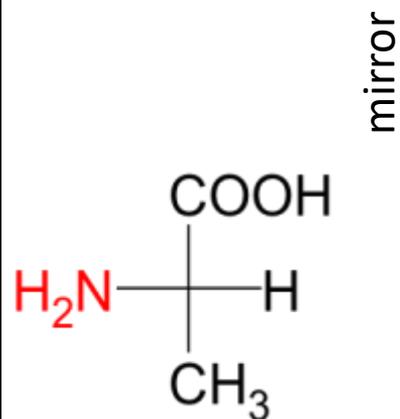
- Isomers: are molecules with same molecular formula but different chemical structures
  1. Constitutional (structural) isomers: atoms and functional groups bind together in different ways
  2. Stereoisomers (spatial isomers): differ in the configuration of atoms rather than the order of atomic connectivity

# D/L Amino Acids

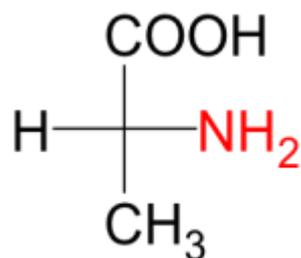


- Enantiomers: are two stereoisomers that are mirror images to each other but not superimposable
- **D-** (dexter)/**L-** (laevus) Nomenclature system: commonly used to assign the configurations in sugars (carbohydrates) and amino acids
- As a rule of thumb: if the amino group is on the right-hand side of  $\alpha$ -carbon at Fisher projection, the configuration is D. If it is on the left-hand side, the configuration is assigned as L.

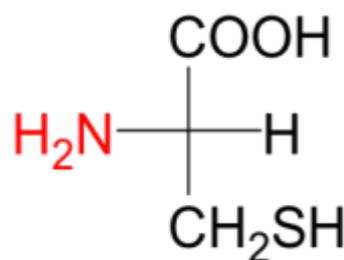
# Fischer Projections of Amino Acids



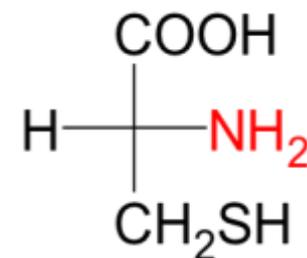
L-alanine



D-alanine



L-cysteine



D-cysteine

**Fisher Projection:** is one way commonly used to represent the structure of chiral molecules like carbohydrates and amino acids

# D/L Amino Acids

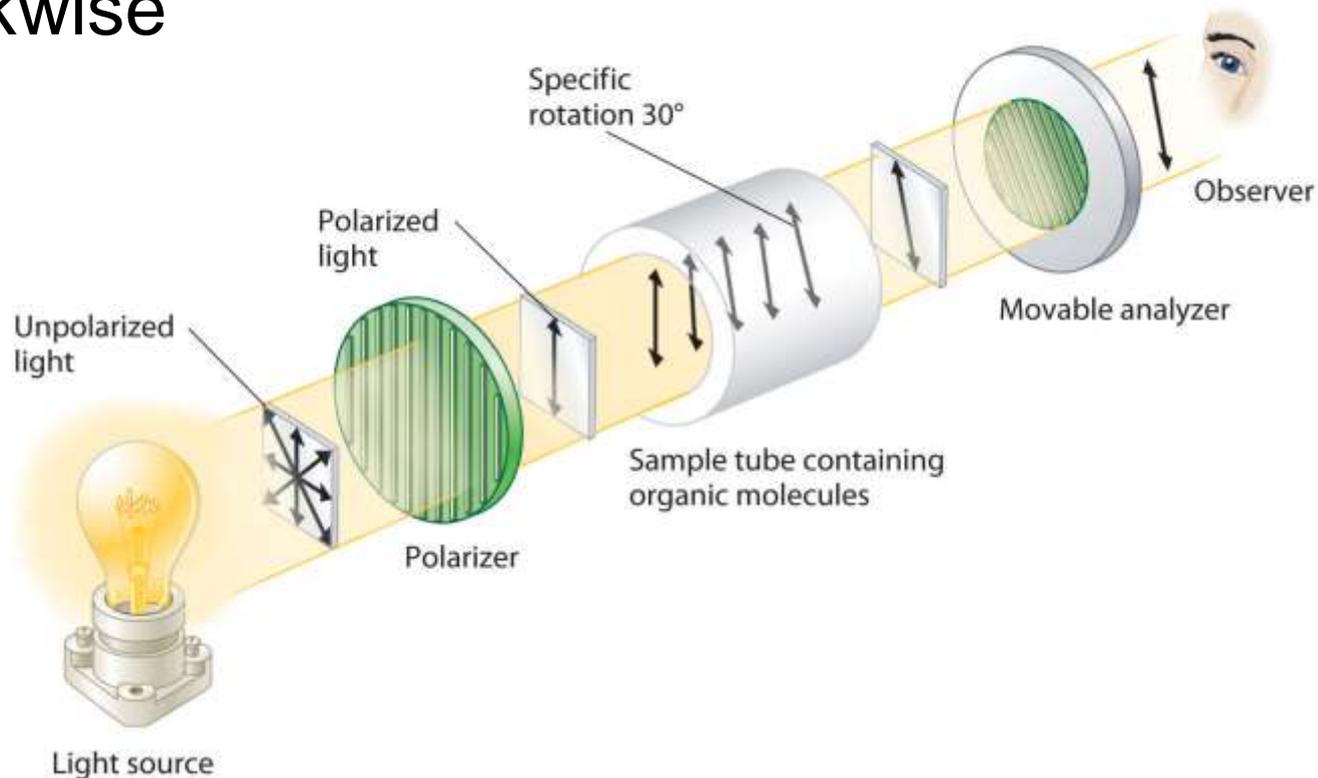


- Most naturally occurring sugars are D-isomers while most naturally occurring amino acids are L-isomers (amino acids of protein)
- D-amino acids polypeptides (right-handed isomers) are components of bacterial cell walls to resist digestion by other organisms

# Optical Activity



- Enantiomers are optically active and can rotate the polarized light plane either clockwise or counterclockwise



**Polarimeter is used to measure optical rotation**

# Optical Activity

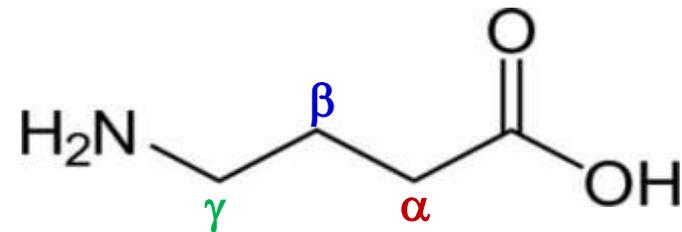


- **(+)/(-) nomenclature system:** if one enantiomer rotates the light clockwise, it is labeled (+) or (*d*) (dextrorotatory). The second mirror image enantiomer is labeled (-) or (*l*) laevorotatory
- D/L system should not be confused with +/- or *d/l* system. For example, D-isomer might be levorotatory
- 9 of 19 L-amino acids commonly found in proteins are dextrorotatory
- **Racemic mixture** contains equal amounts of each enantiomer (net rotation is zero)

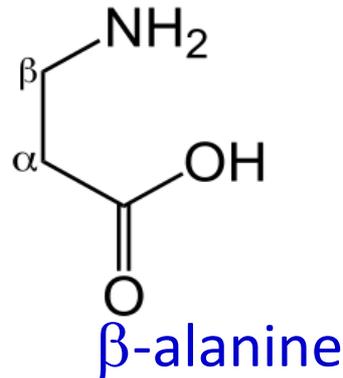
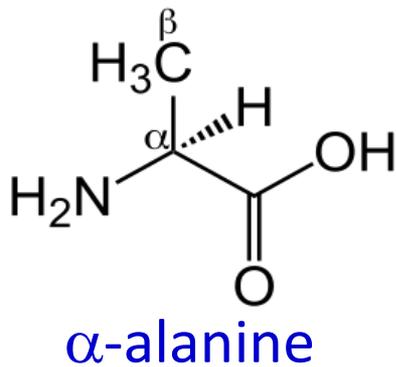
# Classification of Amino Acids



- >300 amino acids classified in many ways:
  - 1) Standard and non-standard amino acids
  - 2)  $\alpha$ ,  $\beta$ ,  $\gamma$  and  $\delta$  amino acids



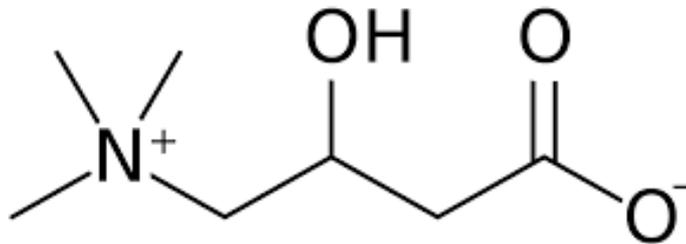
$\gamma$ -aminobutyric acid (GABA) is the inhibitory neurotransmitter in the brain



# Classification of Amino Acids



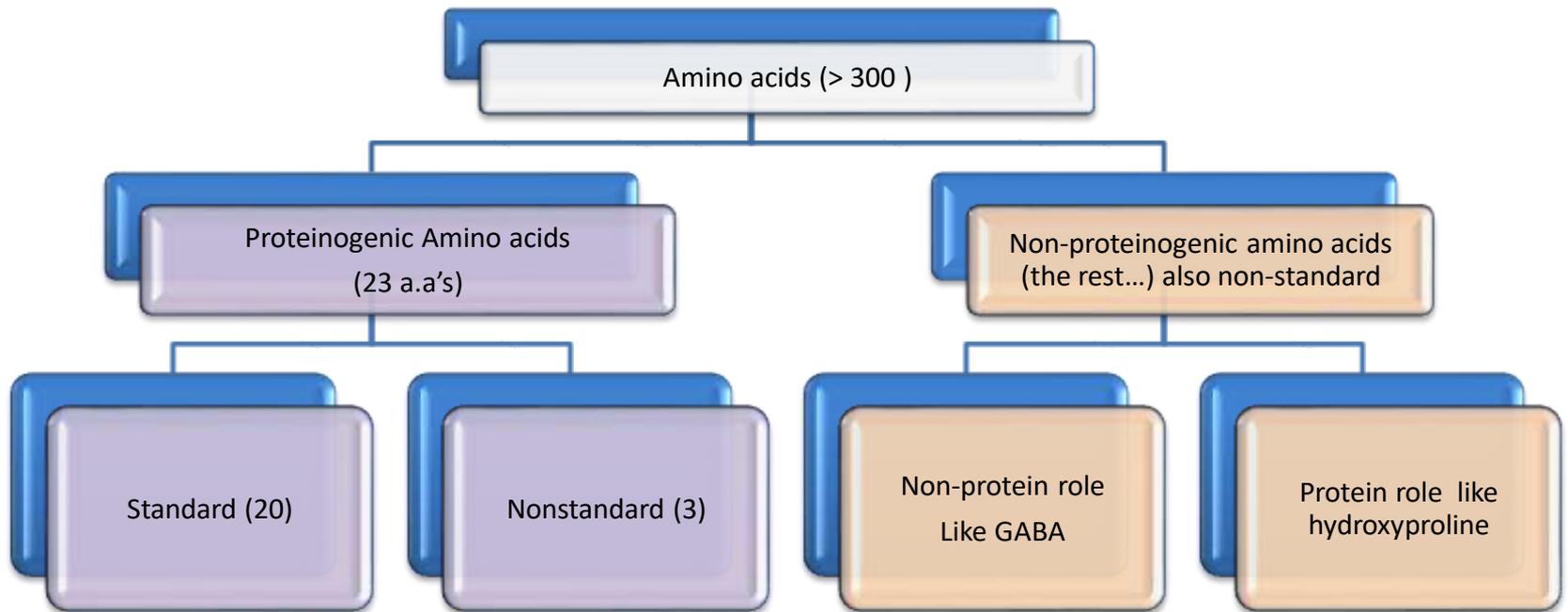
3) Proteinogenic and non-proteinogenic amino acids (non-proteinogenic amino acids either have non-protein role like GABA and carnitine or have a protein role but formed by post-translational modification of protein like hydroxyproline )



**Carnitine has a role in lipid transportation and fat metabolism**



# Classification of Amino Acids



# Categories of Standard Amino Acids

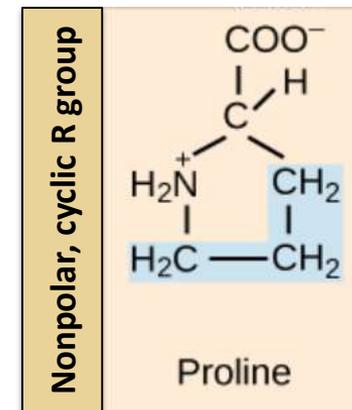
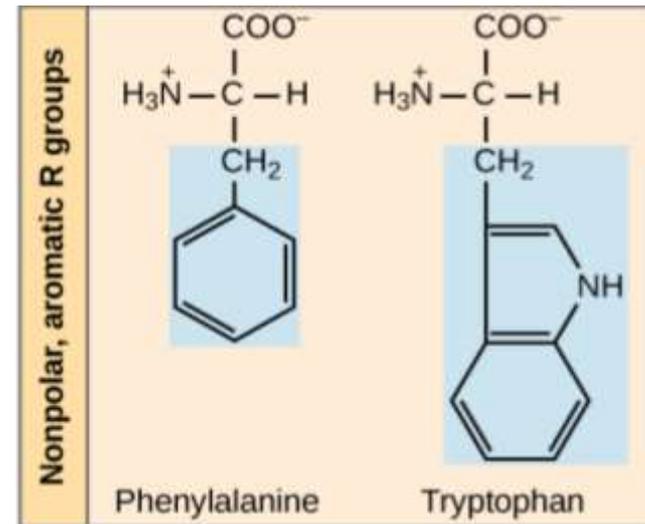
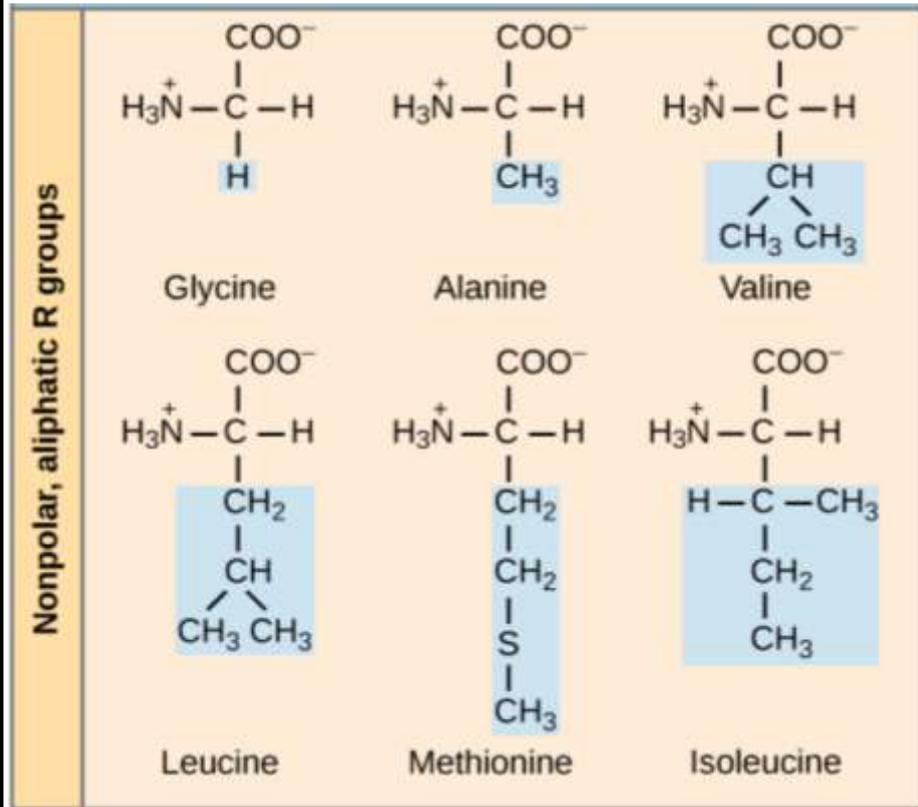


- The 20 standard amino acids are classified into 3 major categories according to the polarities of their “R” groups:
  - 1) **Amino acids with non-polar R groups**
  - 2) **Amino acids with charged polar R groups**
  - 3) **Amino acids with uncharged polar R groups**

# Amino acids with non-polar R groups



- 6 amino acids with aliphatic, 2 with aromatic and one with cyclic side chains



# Amino acids with non-polar R groups

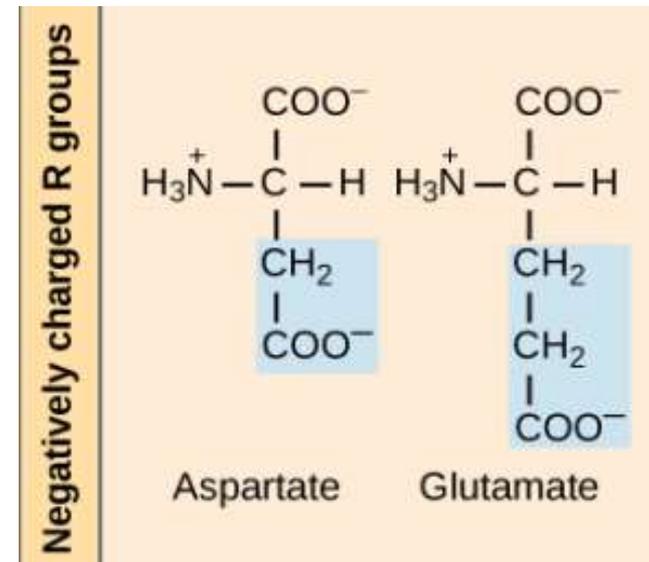
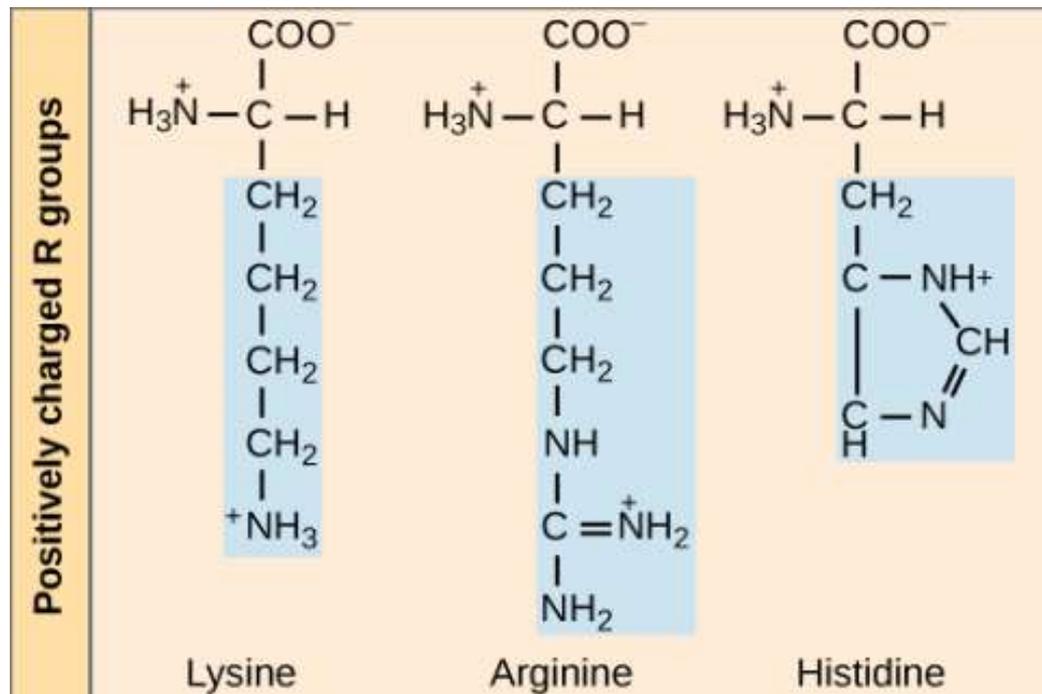


- Glycine has the simplest side chain: H atom
- Alanine, valine, leucine and isoleucine have aliphatic hydrocarbon side chains
- Methionine has a thioether side chain (sulfur atom)
- Proline has a cyclic pyrrolidine side chain
- Phenylalanine has a phenyl moiety
- Tryptophan has an indole group

# Amino acids with charged polar R groups



- 3 amino acids are positively charged (basic) and 2 amino acids are negatively charged (acidic)





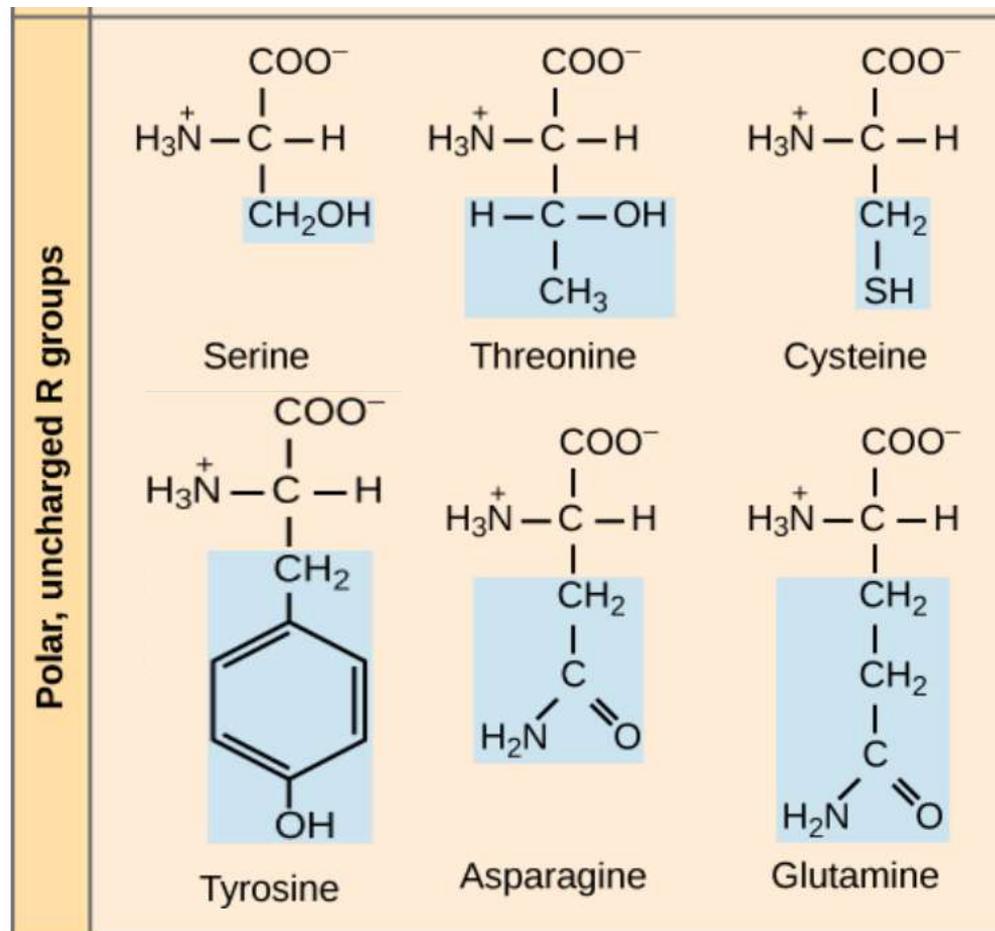
## Amino acids with charged polar R groups

- Arginine has a guanidine group
- Lysine has a butyl ammonium side chain
- Histidine has imidazole group
- Aspartic and glutamic acids in their ionized state are called aspartate and glutamate, respectively

# Amino acids with uncharged polar R groups



- 6 amino acids with hydroxyl, amide or thiol groups



## Amino acids with uncharged polar R groups

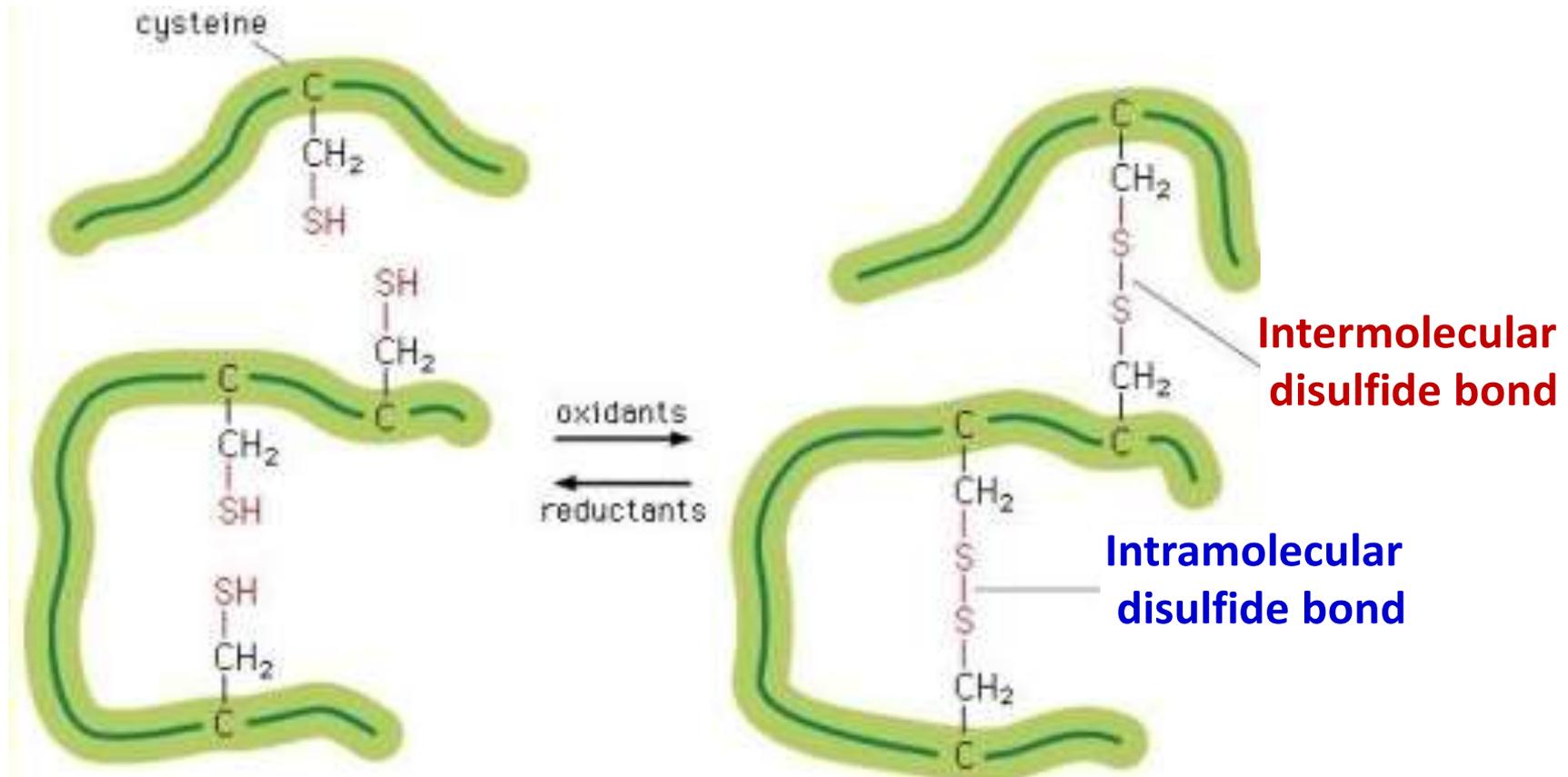


- Serine and threonine bear hydroxyl (-OH) R group and Tyrosine is aromatic and has a phenolic group
- Asparagine and glutamine have amide bearing side chains. They are the amide derivatives of aspartic and glutamic acids (OH is replaced with NH<sub>2</sub>)
- Cysteine is unique because it has free sulfhydryl (**-SH**) group that can form a disulfide bond (-S-S-) with another cysteine through the oxidation of 2 thiol groups (cystine is the oxidized **dimeric** form). The disulfide bridge in proteins contributes to the stability and overall shape of a protein

# Amino acids with uncharged polar R groups



- Disulfide bond** is a covalent linkage formed between the sulfhydryl groups (SH) of **two cysteine residues** (after oxidation) to produce a **cystine** residue



## Amino acids with uncharged polar R groups



- Cysteine residues may be separated from each other by many amino acids in the primary sequence of a polypeptide or may even be located on two different polypeptides. The folding of the polypeptide chain(s) brings the cysteine residues into proximity and permits covalent bonding of their side chains.
- Disulfide bond could be **intramolecular** (2 cysteine residues on the same polypeptide chain) or **intermolecular** (2 cysteine residues on two separate/ different polypeptide chains)

# Amino Acids Abbreviations



<u>3-letters</u>	<u>1-letter</u>	<u>Amino acid</u>
Ala	A	<u>A</u> lanine
Arg	R	<u>A</u> rginine
Asn	N	<u>A</u> sparagine
Asp	D	Aspartic acid (Aspartate)
Cys	C	<u>C</u> ysteine
Gln	Q	Glutamine
Glu	E	Glutamic acid (Glutamate)
Gly	G	<u>G</u> lycine
His	H	<u>H</u> istidine
Ile	I	<u>I</u> soleucine
Leu	L	<u>L</u> eucine
Lys	K	Lysine
Met	M	<u>M</u> ethionine
Phe	F	Phenylalanine
Pro	P	<u>P</u> roline
Ser	S	<u>S</u> erine
Thr	T	<u>T</u> hreonine
Trp	W	Tryptophan
Tyr	Y	<u>T</u> yrosine
Val	V	<u>V</u> aline

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**Transcript: wnt8a-201** ENSDART00000132193.3

**Description** wingless-type MMTV integration site family, member 8a [Source:ZFIN;Acc:[ZDB-GENE-980526-332](#)]

**Gene Synonyms** etID309727.14, wnt8, wnt8 ORF1, wnt8 ORF2, wnt8.1, wu:fa20e02, wu:fe05d07

**Location** [Chromosome 14: 34,490,445-34,494,899](#) forward strand.

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wnt8a-202	<a href="#">ENSDART00000144301.2</a>	884	<a href="#">263aa</a>	Protein coding	<a href="#">B8A6C1</a>	<a href="#">NM_130946</a> <a href="#">NP_571021</a>	CDS 3' incomplete
wnt8a-203	<a href="#">ENSDART00000148044.3</a>	554	<a href="#">161aa</a>	Protein coding	<a href="#">B8A6C0</a>	-	CDS 3' incomplete

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**Protein sequence**

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Exons An exon **Another exon** Residue overlap splice site

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