

# [Amino Acids 1]

~ Biologically important organic molecules that contain both

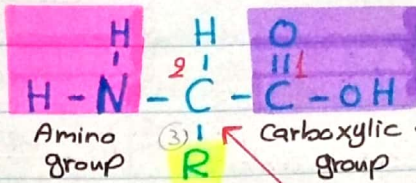
ٻه وظيفتي مجموعتين - two functional groups

- Carboxylic acid (-COOH)
- amino (-NH<sub>2</sub>)

\* we can differentiate between

20 standard amino acids from the

side chain [R-group].



Amino group

R

Carboxylic acid group

side chain

α-carbon

is the carbon which neighbour of the carboxylic group.

\* Carboxylic group is [reference point]

نقطة مرجعية

منه اولى مرتبة جنبها اعطوا (α) والثانية (β) التي

~ Side chain [R group] is specific to each amino acid

~ Amino group is attached to α-carbon (C<sub>2</sub>)

**note** Standard amino acids are called [α-amino acid] because the

Amino group is attached to α-carbon (C<sub>2</sub>).

## - Biological significance of amino acids -

1- Amino acids are N-containing molecules.

2- The basic structural building units - monomers - of proteins.

3- Precursors of many biomolecules like neurotransmitters [non-

Protein role].

4- They are also utilized as an energy source.

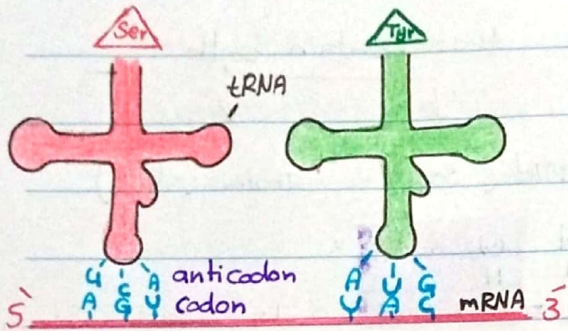
\* اي غذاء نؤكلوا نستخدم واحد من اربعة اقسام [سكريات و بروتينات و اطياف امينية و دهون]

~~20 standard~~ <sup>شعري</sup> [canonical] amino acids which are encoded directly

by triplet codons in the universal genetic code during in vivo protein

synthesis process [mRNA translation].

~ Genetic code table ~



- standard amino acids  
 - basic amino acids

\* The [20 standard] amino acids are known as [proteinogenic] or [neutral] amino acids.

\* Proteinogenic [time of addition] added to the polypeptide chain

[mRNA translation]

\* neutral [automatically]

... من وضع البروتينات

2<sup>nd</sup> base in codon

|   |                   | U   | C   | A    | G    |   |
|---|-------------------|-----|-----|------|------|---|
| U | 1st base in codon | Phe | Ser | Tyr  | Cys  | U |
|   |                   | Phe | Ser | Tyr  | Cys  | C |
|   |                   | Leu | Ser | Stop | stop | A |
|   |                   | Leu | Ser | Stop | Trp  | G |
| C | 1st base in codon | Leu | Pro | His  | Arg  | U |
|   |                   | Leu | Pro | His  | Arg  | C |
|   |                   | Leu | Pro | Gln  | Arg  | A |
| A | 1st base in codon | Ile | Thr | Asn  | Ser  | G |
|   |                   | Ile | Thr | Asn  | Ser  | U |
|   |                   | Ile | Thr | Lys  | Arg  | C |
|   |                   | Met | Thr | Lys  | Arg  | A |
| G | 1st base in codon | Val | Ala | Asp  | Gly  | G |
|   |                   | Val | Ala | Asp  | Gly  | U |
|   |                   | Val | Ala | Glu  | Gly  | C |
|   |                   | Val | Ala | Glu  | Gly  | A |

3<sup>rd</sup> base in codon

... من الجين

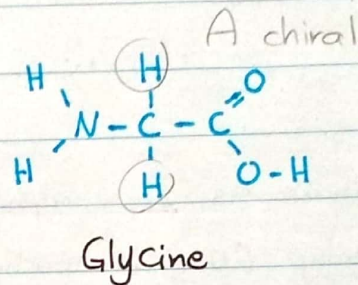
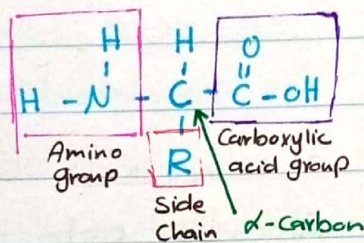
• Standard Amino Acids List •

- |                 |          |              |             |
|-----------------|----------|--------------|-------------|
| - Histidine     | - Valine | - Arginine   | - Alanine   |
| - Isoleucine    | - Lysine | - Asparagine | - Aspartate |
| - Leucine       |          | - Glutamine  | - Cysteine  |
| - Methionine    |          | - Glycine    | - Glutamate |
| - Phenylalanine |          | - Proline    |             |
| - Threonine     |          | - Serine     |             |
| - Tryptophane   |          | - Tyrosine   |             |



3. They are all chiral molecules [except glycine which has a chiral C] with L- stereochemical configuration (left-handed isomers).

- Chiral objects: cannot be superimposed.
- Achiral objects: can be superimposed.
- Chiral molecules should contain at least one chiral center (usually a carbon atom).
- Chiral carbon: asymmetric carbon atom attached to 4 different groups of atoms.



## Isomerization

- Isomers: are molecules with same molecular formula but different chemical structures.

1) Stereoisomers (spatial isomers): differ in the configuration of atoms rather than the order of atomic connectivity.

2) Constitutional (structural) isomers: atoms and functional groups bind together in different ways.

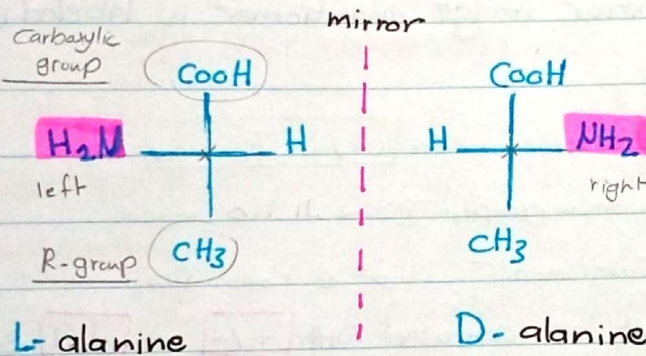
\* Enantiomers: are two stereoisomers that are mirror images to each other but not superimposable.

Enantiomers must have at least one chiral center

• D- (dexter) L-(laevus) nomenclature system commonly used to assign the configurations in sugars [carbohydrates] and amino acids.

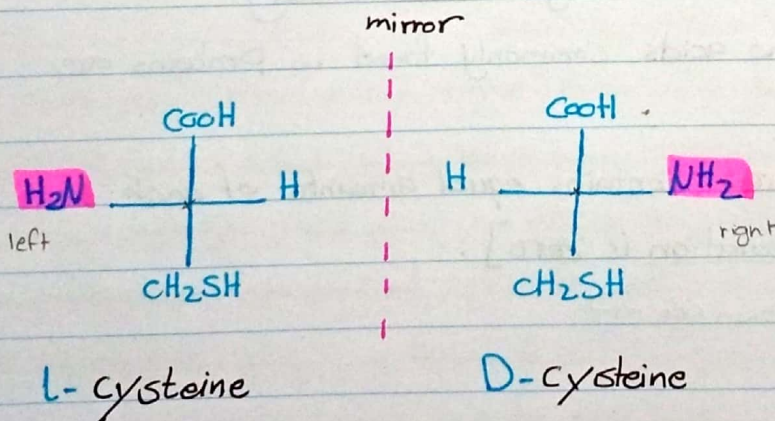
• As a rule of thumb: if the amino group is on the (right-hand side) of  $\alpha$ -carbon at fisher projection, the configuration is (D). If it is on the (left-hand side), the configuration is assigned as (L).

### ~ Fischer Projections of Amino Acids ~



### ~ FISCHER PROJECTION ~

• is one way commonly used to represent the structure of chiral molecules like carbohydrates and amino acids.



في التسمية الكيميائية  
 - [D] → اليمين / كمال  
 - [L] → اليسار / كمال

✓ Most naturally occurring sugars are D-isomers while most naturally occurring amino acids are L-isomers [amino acids of protein].

✓ D-amino acids polypeptides [right-handed isomers] are components of bacterial cell walls to resist digestion by other organisms.

### - Optical Activity -

• Enantiomers are optically active and can rotate the Polarized light plane either clockwise or counterclockwise.

✓ (+)/(-) nomenclature system:-

If one enantiomer rotates the light clockwise, it is labeled (+) or (d) (dextrorotatory). The second mirror image enantiomer is labeled (-) or (l) (levorotatory).

\* Question: What it does mean (+) L-Tyrosine?

on the left-hand side of the amino group is Tyrosine. -

هذا يدل على أن Polarimeter

✓ **D/L** system should not be confused with **+/-** or **d/l** system.

- for example: D-isomer might be levorotatory.

✓ 9 of 19 L-amino acids commonly found in proteins are dextrorotatory.

✓ **Racemic mixture**:- Contains equal amounts of each enantiomer [net rotation is zero].

Zero rotation is ←

# ~ Amino Acids [ $>300$ ] ~

• Proteinogenic Amino Acids 23

• Non-Proteinogenic Amino Acids

•  $\alpha, \beta, \gamma$  and  $\delta$  amino acids.

20

3

Standard

Non-Standard

• non-Protein role

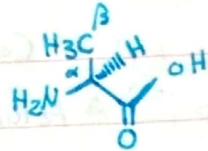
• Protein role

role [hydroxy Proline]

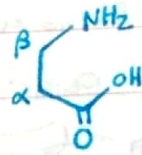
Functional group - non-proteinogenic amino acids

[Carnitine] & [GABA] etc

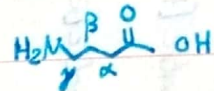
non-standard



$\alpha$ -alanine



$\beta$ -alanine



$\gamma$ -amino butyric acid [GABA]

• Protein role  
- Protein role  
المساهمة البروتينية في انشاء mRNA translation Process.

تتبع البروتين من صنع الى ال

Cytoplasm ويصنع على

في ذرى الخلية

[Post-translational system]

هنا نوع حسب اواخر الخلية

functional group

مثال يقدر اسمها و مثلاً: Proline [Standard-amino acid] بنوع Proteinogenic

على mRNA-translation، اذا هذا البروتين التي فيه بروتين راجع على connective tissue مثلاً

التي وتوثره على الاصنع هاد و جعلوا هي بروتين صيفي على (OH) في

Hydroxy Proline هاد اصنع مختلف بنوع non-proteinogenic/non-standard

with protein role

\*  $\beta$ -amino acids are non-proteinogenic with  $\beta$ -alanine is the only common naturally occurring  $\beta$ -amino acid.  $\beta$ -alanine is used in plants and microorganisms in the synthesis of pantothenic acid (vitamin B<sub>5</sub>).

\* unlike  $\alpha$ -peptides, The  $\beta$ -peptides are artificial peptides used in some antibiotics to counter resistance as they are more stable against proteolytic degradation.

according to  $\checkmark$  Categories of standard Amino Acids  $\checkmark$   
 to  $\checkmark$  Polarities of their R groups.

1- Amino acids with non-polar [R groups]

- (6) aliphatic [R groups]
- (2) aromatic [R groups]
- (1) cyclic [R groups]

2- Amino acids with charged polar [R groups]

- (3) amino acids are positively charged (basic)
- (2) amino acids are negatively charged (acidic)

3- Amino acids with uncharged polar [R groups]

- (6) amino acids with hydroxyl, amide, or thiol groups.

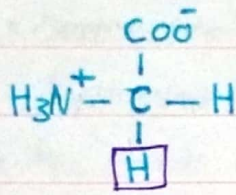
1- Amino acids with non-polar R groups

- Glycine has the simplest side chain: H atom. [achiral].
- Alanine, Valine, leucine and isoleucine have aliphatic hydrocarbon side chains.
- Methionine has a thioether side chain [sulfur atom].
- Proline has a cyclic pyrrolidine side chain.
- Phenylalanine has a phenyl moiety. تجزئة الجزيء إلى مجموعة
- Tryptophan has an indole group. مجموعة

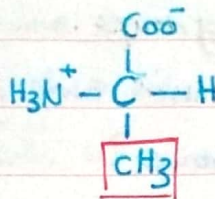
[تريبتوفان] ←



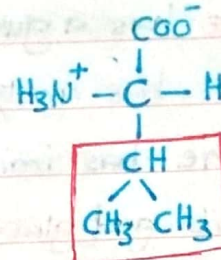
~ NonPolar, aliphatic [R groups] ~



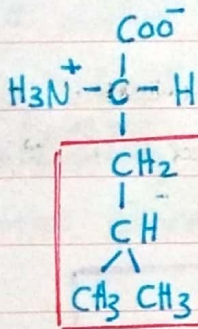
Glycine



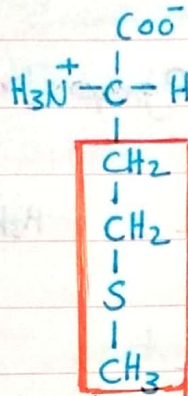
Alanine



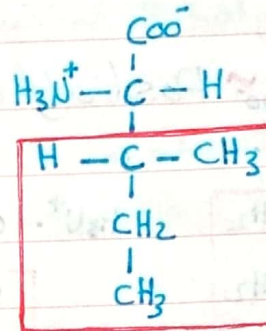
Valine



Leucine

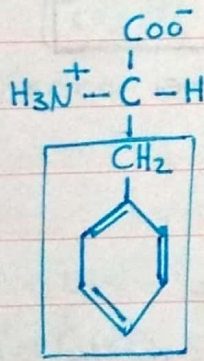


Methionine

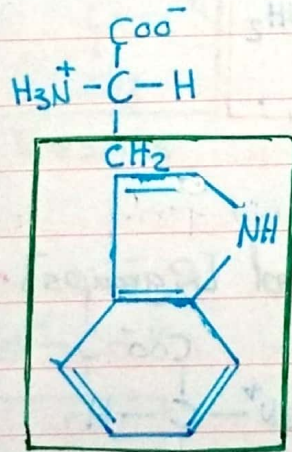


Isoleucine

~ Nonpolar, aromatic [R groups] ~

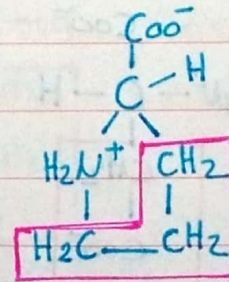


Phenylalanine



Tryptophan

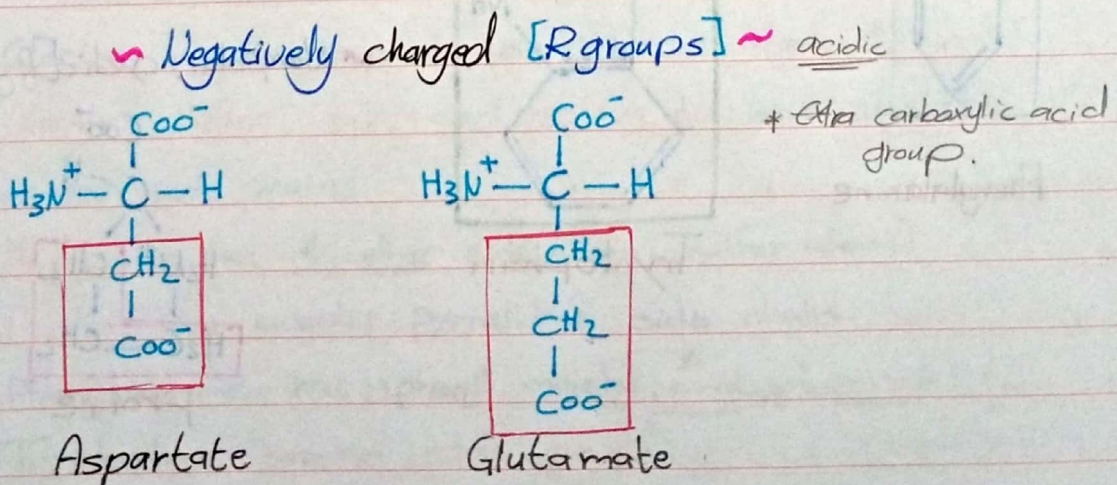
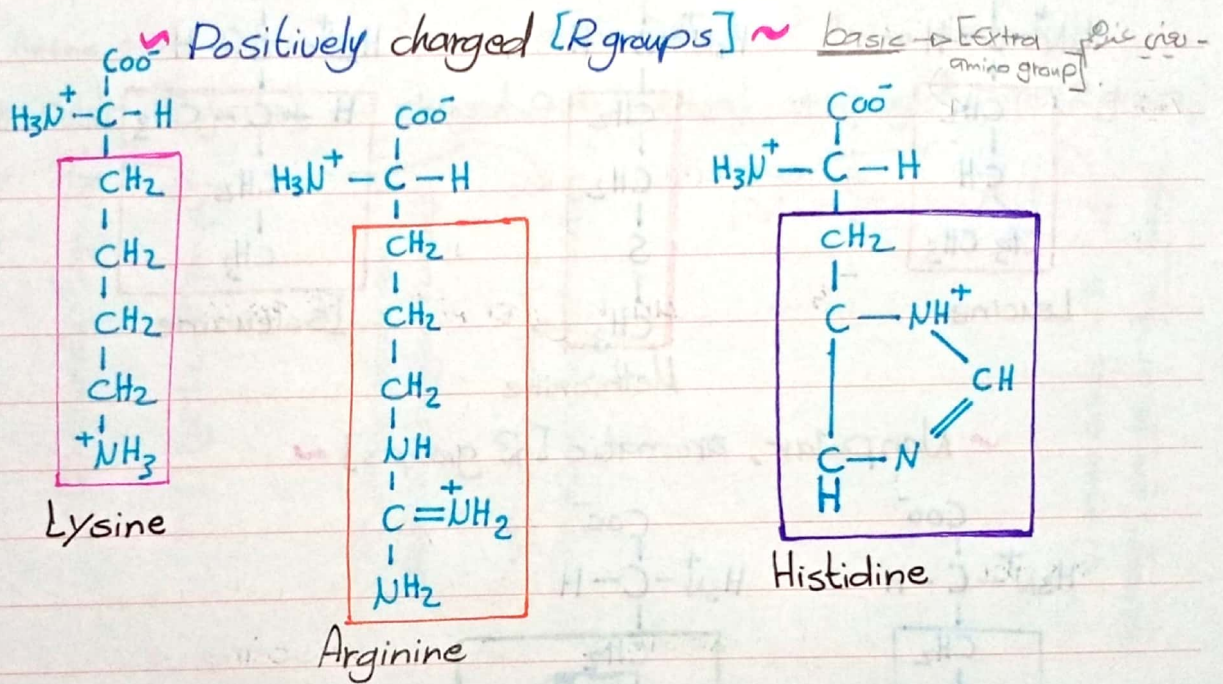
~ Nonpolar, cyclic [R group] ~



Proline

## 2. Amino acids with charged polar [R groups]

- Arginine has a guanidine group.
- Lysine has a butyl ammonium side chain.
- Histidine has an imidazole group.
- Aspartic and glutamic acids in their ionized state are called aspartate and glutamate, respectively.

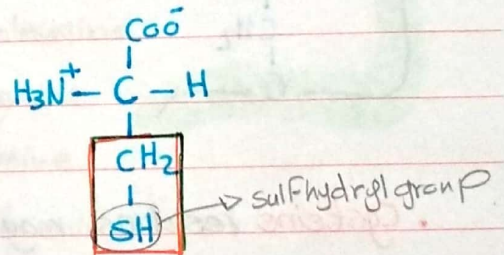
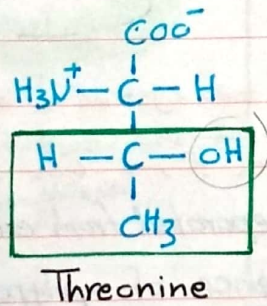
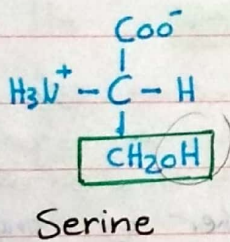


### 3- Amino acids with uncharged polar [R groups]

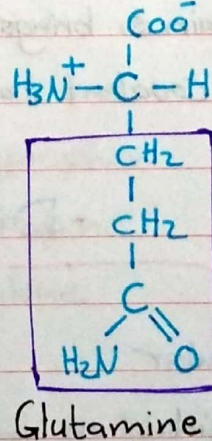
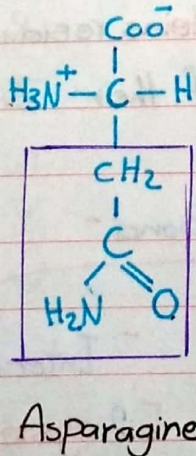
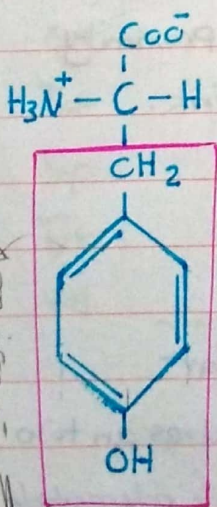
- Serine and threonine bear hydroxyl (-OH) R group.
- Asparagine and glutamine have amide bearing side chains. They are the amide derivatives of aspartic and glutamic acids.
- Tyrosine is aromatic and has a phenolic group.
- Cysteine has a thiol group that can form a disulfide bond (-S-S) with another cysteine through the oxidation of 2 thiol groups ([Cystine] is the oxidized dimeric form).

The disulfide bridge in proteins contributes to the stability and overall shape of a protein.

### ~ Polar, uncharged polar [R groups] ~

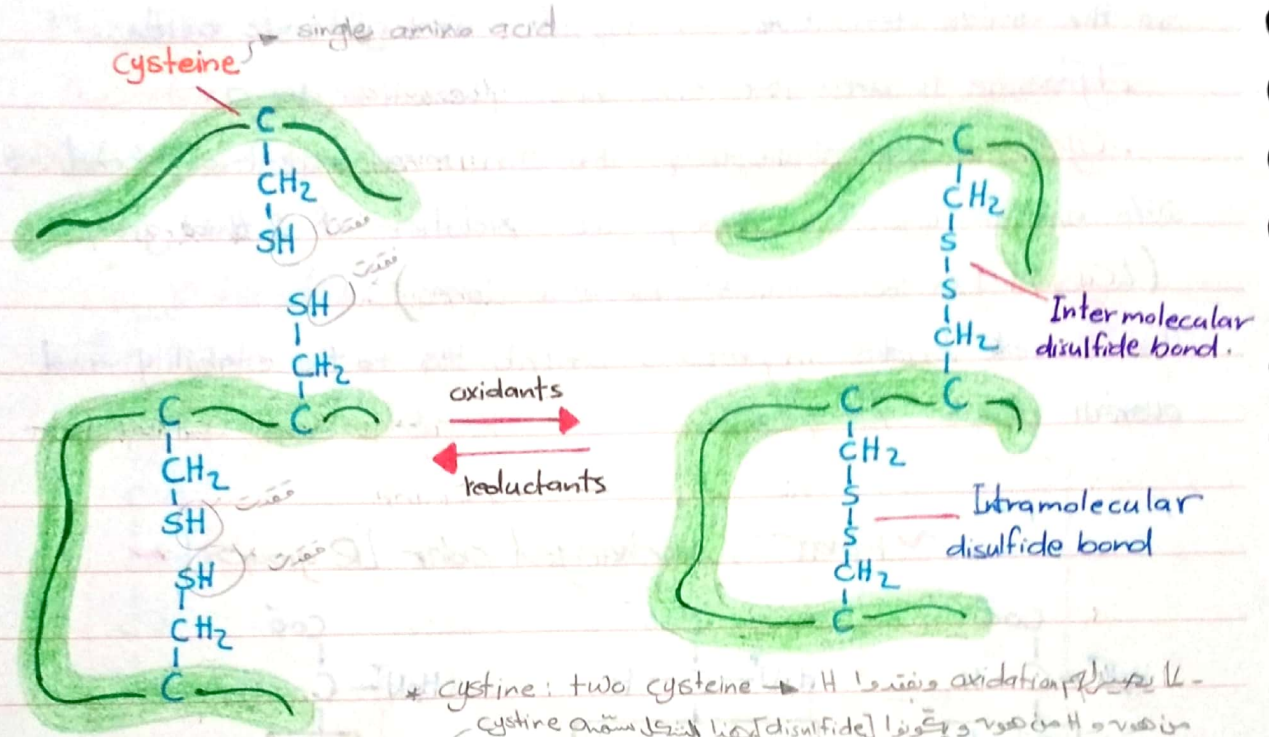


Cysteine: the most important amino acid.



+ carbonyl group + nitrogen in side chain which amide bond.

\* Disulfide bond is a covalent linkage formed between the sulfhydryl groups (SH) of two cysteine residues (after oxidation) to produce a cystine residue.



\* Cystine: two cysteine → H<sub>2</sub>O oxidation  
 cystine dimer (disulfide) → 2 cysteine + H<sub>2</sub>O  
 dimer (disulfide) → 2 cysteine + H<sub>2</sub>O  
 2 cysteine + H<sub>2</sub>O → dimer (disulfide) + 2H<sup>+</sup>

\* Cysteine residues may be separated from each other by many amino acids in the primary sequence of a polypeptide or may even be located on two different polypeptides. The folding of the polypeptide chain(s) brings the cysteine residues into proximity and permits covalent bonding of their side chains.

### - Disulfide Bond -

- Intramolecular  
 [2 cysteine residues on the same polypeptide chain].

- Intermolecular  
 [2 cysteine residues on two separate / different polypeptide chains].

## ~ Amino Acids Abbreviations ~

more common

\* راجع اسلاید 30 + 31 فی المشافہہ

3-letters

1-letter

Amino acid

|     |   |                                     |
|-----|---|-------------------------------------|
| Ala | A | <u>A</u> lanine                     |
| Arg | R | <u>A</u> rginine                    |
| Asn | N | Asparagine                          |
| Asp | D | Aspartic acid [ <u>A</u> spartate]. |
| Cys | C | <u>C</u> ysteine                    |
| Gln | Q | Glutamine                           |
| Glu | E | Glutamic acid [ <u>G</u> lutamate]. |
| Gly | G | <u>G</u> lycine                     |
| His | H | <u>H</u> istidine                   |
| Ile | I | <u>I</u> soleucine                  |
| Leu | L | <u>L</u> eucine                     |
| Lys | k | Lysine                              |
| Met | M | <u>M</u> ethionine                  |
| Phe | F | Phenylalanine                       |
| Pro | P | <u>P</u> roline                     |
| Ser | S | Serine                              |
| Thr | T | <u>T</u> hreonine                   |
| Trp | W | Tryptophan                          |
| Tyr | Y | <u>T</u> yrosine                    |
| Val | V | <u>V</u> aline                      |

\* راجع اسلاید 30 + 31 فی المشافہہ Example میں مطلوب مندرجہ