Fluid and ELECTROLYTES

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DISTRIBUTION OF FLUIDS

- Fluids = 60% of total body Weight
- Two major compartments >>
- Intracellular 2/3rd, Extra-cellular 1/3rd.
- Extra-cellular >>

interstitial fluid (in between cells) 3/4th intravascular fluid (plasma) 1/4th

Total body water 40 L, 60% body v	volume = veight	
	Extracellular fluid volume = 15 L, 20% body weight	
Intracellular fluid volume = 25 L, 40% body weight	Interstitial fluid volume = 12 L, 80% of ECF	Plasma volume = 3 L, 20% of ECF

COMPOSITION & PROTEIN

 EXTRA CELLULAR FLUID[I.V.+inters.] HAVE THE SAME COMPOSITION

BUT

- INTRAVASCULAR PROTEIN [MAINLY ALBUMIN] CONCENTRATION IS MORE THAN THE INTERSTITIAL ONE and it is the most effective intravascular osmotic pressure.
- Na IONS[ELECTRICITY] ACCOUNT FOR MOST THE OSMOLALITY IN THE EXTRACELLULAR FLUID
- WHEREVER SODIUM GOES WATER GOES WITH IT PROPORTIONALLY
- Posm(mOsm/kg) = 2× Na+ + [glucose]/18+BUN/2.8
- The normal osmolality of plasma ranges from 275 to 290 mOsm/kg.

Osmolality: is concentration of substance in osmoles in one Kg of solvent Osmolarity : is concentration of substance in osmoles in one liter of solvent

Fluid requirements/24 hours:

- Water : 30-35 ml/kg
- K : 1 mEq/kg
- Na : 1-2 mEq/kg
- Cl : 1.5 mEq/kg

Normal water loss:

- Urine 1200-1500 ml/24h
- Sweat : 200 ml
- Respiratory losses: 500-700 ml
- Feces :100-200 ml

• Insensible fluid losses : loss of fluid that is not directly measured .

Daily secretions :

- Bile :1 L/24 h
- Gastric : 2L/24 h
- Pancreatic : 600 ML/24 h
- Small intestine : 3 L/24 h
- Saliva : 1500 ml/24 h
- Most secretions are reabsorbed

GI ELECTROLYTE LOSSES

- Sweat hypotonic (Na concentration 35–65)
- Saliva K⁺ (highest concentration of K⁺ in body)
- Stomach H⁺ and Cl⁻
- Pancreas HCO₃⁻
- Bile HCO₃⁻
- Small intestine HCO₃⁻, K⁺
- Large intestine K⁺
- Dialysis can remove K, Ca, Mg, PO₄, urea, and creatinine

Third spacing !

- Fluid accumulation in the interstitial of tissues (lumen of the small bowel in case of IO, Pancreatitis, PERITONITIS, Post surgery).
- INTRVASCULAR , INTRACELLAUR SPACSES as the first two spaces.
- **POSTOPERATIVE** !
- Around post op day 3, Fluids return back to intravascular space so be aware of fluid overload at this time.

INPUT = OUTPUT = EUVOLEMIA

- Assessment of fluid status :
- Skin turgor , mucus membranes
- Vital signs
- Weight changes
- Urine output : minimum adult = 0.5 ml/kg
- Jugular vein distention , LL Edema
- Crackles
- CVP , PCWP
- Lactic acid
- CXR findings

LEG OEDEMA TESTING NECK VEINS DISTENTION





CHEST X-RAYNORMALPULMONARY OF DEMA



Electrolyte implance

Sodium (135-145 mEq/L)

- Plasma sodium levels are directly affected by fluid balance.
- Hyponatremia results from excess water intake that is not excreted, whereas hypernatremia results from an excessive loss of free water.
- Hyponatremia is defined as a sodium level less than 135 mEq/L
- ➤ mild, 130–134 mEq/L
- ➤ moderate, 120–129 mEq/L
- ➤ severe, <120 mEq/L.</p>

HYPONATREMIA

- 1- measure serum osmolality
- Hypo / hyper / iso-tonic
- Hypotonic hyponatremia
- HYPO-VOLEMIC : NG suction , burns , pancreatitis , diaphoresis >> IV NS & Correct underlying cause
- EU-VOLEMIC : SIADH , CNS >>> fluid restriction
- HYPER-VOLEMIC : RF, CHF, LIVER FAILURE, DILITUIONAL, FLUID OVERLOAD >> fluid restriction and diuretics



HYPONATREMIA

Clinically : seizures , confusion lethargy , coma weakness

 Always don't forget to correct hyponatremia slowly (0.25–0.5 mEq/L per hour), to avoid central pontine myelinolysis.

HYPERNATREMIA

- Hypernatremia is defined as a sodium level greater than 145mEq/L
- (moderate, 146–159 mEq/L; severe, ≥160 Eq/L).
- <u>Causes</u> :
- In surgical pt. its mostly due to volume depletions with inadequate hydration, others : DI Vomiting diarrhea, diuretics
- <u>Clinically :</u>

seizures , confusion, tremors .respiratory paralysis .

- Slow lowering of serum sodium is very important less than 12 mEq/day TO AVOID SEZIURES . (brain edema)
- Treatment by hypotonic fluids D5W ., ½ NS , ¼ NS



В

POTASSIUM [NORMAL SERUM K 3.5-5mmol/L]

- 98% INTRACELLULAR
- 2 % EXTRACELLULAR
- 10% LOSS OF TOTAL BODY K DROPS SERUM LEVEL FROM 4 to3 meq



- RENAL EXCRETION OF K IS REGULATED BY ALDOSTERONE [Mineralocorticoid]
- RENAL FAILURE FAILS TO EXCRETE K, THIS LEADS TO HYPERKALAEMIA
- THERE IS ALWAYS EXCHANGE OF K FOR HYDROGEN IONS IN PH CHANGES TO CORRECT THE PH.
- HYPERKALEMIA LEADS TO ACIDOSIS.
- ACIDOSIS LEADS TO HYPERKALEMIA

Hyperkalaemia due to acidosis

IN ACIDOSIS [H] IONS GO INTO THE CELL TRYING TO REDUCE THE ACIDOSIS . [K] COMES OUT OF THE CELL CAUSING HYPERKALEMIA WHICH CAN BE LETHAL

HYPOKALAEMIA

 IS THE COMMONEST INTHE SURGICAL WARD

PARALYTIC ILEUS



HYPOKALAEMIA LOW SERUM POTASSIUM

LOSS THROUGH 1- GIT, 2- RENAL OR INTO THE 3- CELL

- o 1-GIT. A-[LOW INTAKE] OR
 B-[HIGH LOSS; VOMIT. N/GTUBE, DIARRHOEA]
 o 2-RENAL LOSS BY A-DIURETICS OR
 B-INCREASED ALDOSTERONE ACTIVITY.
- 3-ALKALOSIS; WHERE SERUM POTASSIUM GOES INTO THE CELLS TO REPLACE THE OUTGOING H IONS IN AN ATTEMPT TO CORRECT THE ALKALOSIS.e.g.; gastric outlet obstruction
- o 4-INSULIN THERAPY; K goes into the cell

EFFECT OF HYPOKALAEMIA

- 1- Decrease G.I. CONTRACTILITY [PARALYTIC ILEUS]
- 2- Decrease RESPIRATORY MUSCLES CONTRACTILITY
- 3- CARDIAC ARRHYTHMIA

ECG Changes in Hypokalemia :

Falt T waves , U waves , ST depression , PAC, PVC , AF.

TREATMENT OF HYPOKALAEMIA

Treat the underlying cause if possible

- **REPLACEMENT:-**
- KCL IV
- MAX RATE THROGH :
- **PERIPHERAL IV LINE : 10 mEq/hour**
- CENTRAL LINE : 20 mEq/ hour

BE CAREFUL OF HYPERKALAEMIA >>RAPID CORRECTION >>> **CARDIAC ARREST IN DIASTOLE**

PERSISTENT HYPOKALAEMIA

HYPOKALAEMIA INSPITE OF REPLACEMENT THERAPY MEANS COEXISTENT

MAGNESIUM DEFICIENCY

NB:REFRACTORY HYPOKALAEMIA & HYPOCALCAEMIA COULD BE DUE TO Mg DEFICIT

HYPERKALAEMIA

IN BRIEF;

ACIDOSIS

DESTRUCTION OF CELLS

ALDOSTERONE SHUTDOWN

- SEVERE TRAUMA
- BURNS
- CRUSH INJURY
- SEVERE CATABOLIC STATE [SEPSIS]
- RENAL FAILURE
- ADDISON`S DISEASE
- BLOOD TRANSFUION

OTHER CAUSES

- 7-ACIDOSIS
- 8-LEUCOCYTOSIS[MARKED]
- 9-THROMBOCYTOSIS[ABOVE ONE MILLION]
- 10-HAEMOLYSIS
- 11-HAEMOLYSED SPECIMEN

HYPERKALAEMIA

- Clinically :
- Decreased DTR
- Weakness
- Parasthesia
- Paralysis
- Respiratory failure

NORMAL ECG



ECG CHANGES in hyperkalemia : critical value > 6.5

- Peaked T wave
- Depressed ST segment
- Prolonged PR
- Wide QRS
- Bradycardia
- V.Fib

EMERGENCY TREATMENT OF HYPERKALAEMIA

- CALCIUM GLUCONATE I.V. REVERSES THE ACTION OF [K] ON THE HEART
- 10 UNITS OF REGULAR INSULIN IN 100 ML OF 50%DEXTROSE I.V.
 INSULIN PUSHES [K] INTO THE CELL
- SOD.BICARB.CORRECTS ACIDOSIS BUT THIS IS COTROVERSIAL

SLOW CORRECTION OF HYPERKALAEMIA

 ORAL CATION EXCHANGE RESIN [SODIUM POLYSTYRENE SULFONATE] Kayexalate

TAKES POTASSIUM IN EXCHANGE FOR SODIUM AND GETS RID OF [K] IN THE STOOL

HYPERKALAEMIA due to RENAL FAILURE NEEDS DIALYSIS

CALCIUM

- NORMAL SERUM Ca .[8.5-10.3mgm/100ml]
 [2.2-2.5 mmol/l]
- MOST OF IT STORED IN BONE
- DAILY INTAKE; 1-3 gm
- MOST OF THE INTAKE IS NOT ABSORBED
- CONTROLLED BY VIT. D, PARATHYROID, CALCITONIN
- CALCIUM IONS [40% OF THE SERUM CALCIUM] IS NECESSARY FOR NEUROMUSCULAR, ENZYMETIC FUNCTION AND BLOOD COAGULATION [IONIZED].
- [50% IS ATTACHED TO ALBUMIN-NOT IONIZED]
- [IF ALBUMIN IS LOW ; CALCIUM WILL BE LOW]
- SERUM LEVEL DOES NOT NECESSARILY INDICATE THE LEVEL OF IONIZED [FUNCTIONING] CALCIUM
- ACIDAEMIA INCREASES IONIZED CALCIUM
- ALKALAEMIA DECREASES IONIZED CALCIUM
- NB: Ca absorption needs vit.D. which is activated in the kidney, that is why in renal failure Ca drops. Low Ca leads to hyperparathyroidism. So in renal failure you may come across low or high Ca level

CALCIUM AND ALBUMIN

IF SERUM ALBUMIN DROPS SERUM CALCIUM WILL DROP TOO, BECAUSE CALCIUM IS ATTACHED TO ALBUMIN

HYPOCALCEMIA < 1.8 mmol/l

THE COMMONEST; AFTER THYROID SURGERY, RENAL FAILURE AND PANCREATITIS

1-HYPOPARATHYROIDISM
2-HYPOMAGNESEMIA[PATIENTS ON IV.FEEDING FOR LONG TIME]
3-SEVERE PANCREATITIS
4-RENAL FAILURE; ACUTE AND CHRONIC
Ca ABSORPTION IS HELPED BY ACTIVE FORM OF VIT.D WHICH IS DONE IN THE KIDNEY. IN RENAL FAILURE THAT DOES NOT HAPPEN. HOWEVER LOW Ca CAN LEAD TO

HYPERPARATHYROIDISM CAUSING HYPERCALCEMIA WITH BONE CHANGES

5-SEVERE TRAUMA [BLOOD LOSS] [ALBUMIN LOSS] 6-MASSIVE BLOOD TRASFUSION 7-CRUSH INJURY [renal failure] 8-NECROTIZING FASCIITIS 9-RHABDOMMYOLYSIS

ALL SOFT TISSUE INJURY CAN CAUSE HYPOCALCEMIA

HYPOCALCEMIA CLINICAL MANEFESTATION

1-CIRCUMORAL NUMBNESS, TINGLING TIPS OF FINGERS AND TOES

2-NEUROMUCULAR HYPERACTIVITY A-EXAGERATED DEEP REFLEXES B-POSITIVE CHVOSTEK SIGN C-CARPOPEDAL SPASM D-MUSCLE CRAMPS E-ABDOMINAL CRAMPS F-CONVULSIONS [RARE]

CARPOPEDAL SPASM







ASSESSMENT TIP

Eliciting Chvostek's sign

Begin by telling the patient to relax his facial muscles. Then stand directly in front of him, and tap the facial nerve either just anterior to the earlobe and below the zygomatic arch or between the zygomatic arch and the corner of his mouth. A positive response varies from twitching of the lip at the corner of the mouth to spasm of all facial muscles, depending on the severity of hypocalcemia.



HYPO-CALCAEMIA MANAGEMENT

ACUTE 1-CHECK FIRST BLOOD [PH]

2-ALKALOSIS SHOULD BE TREATED [AS IT REDUCES IONIZED Ca)

3-I.V CALCIUM [Ca GLUCONATE OR CHLORIDE] IN REAL HYPOCALCEMIA

CHRONIC : CHECK Mg level , CHECK ALBUMIN

- 1. ORAL CALCIUM +
- 2. VIT.D[1-ALFA-CHOLE-CALCIFEROL]
- 3. ALUM.HYDROXIDE TO BIND DIETARY PHOSPHATE

HYPERCALCEMIA

1. HYPERPARATHYROIDISM

[Adenoma 90%, Hyperplasia 9%, carcinoma<1%]

2-BONE METASTASIS OF CERTAIN CANCER [OSTEOCLASTIC] ; BREAST [The commonest cause], PROSTATE, BRONCHUS, KIDNEY AND THYROID]

3-NEOPLASM WITH ENDOCRINE SECRETION ; BRONCHUS, KIDNEY AND OVARY
[1+2+3 account for 90% of hypercalcemia]
4-VIT.D INTOXICATION
5-SARCOIDOSIS
6-HYPERTHYROIDISM
7-PROLONGED IMMOBILIZATION
8-MILK ALKALI SYNDROME[Ca +milk used to be ttt of p.u]

HYPERCALCEMIA CLINICAL PICTURE

NEUROMUSCULAR LOW ACTIVITY

EASY FATIGABILITY

- MUSCLE WEAKNESS
- CONSTIPATION
- ANOREXIA & VOMITING
- DEPRESSION
- POLYUREA THEN OLIGUREA DUE TO DEHYDRATION
- POLYDIPSIA AND VOMITING
- CALCINOSIS [ANYWHERE BUT RENAL IN PARTICULAR]
- SEVERE FORMS CAUSE COMA AND DEATH

BONES, STONES, ABDOMINAL GROANS AND PSYCHIC MOANS IN HYPERPARATHYROIDISM

MEDICAL EMERGENCY:>15 mg/100ml

Bone cysts and fractures



SEVERE HYPERCALCEMIA [>14.5mg/100ml]

1.CORRECT DEHYDRATION AND WASH OUT CALCIUM THROUGHTHE KIDNEY BY N/S INFUSION

2. FUROSEMIDE [LASIX] AND SOD.SULFATE INCREASE URINARY EXCRETION OF CALCIUM

3. I.V INORGANIC PHOSPHATES. INHIBIT BONE RESORPTION

3. PLICAMYCIN ; FOR BONE METS

4. STEROIDS; FOR 1- SARCOIDOSIS 2- VIT.D INTOXICATION 3- ADDISON'S DIS

5. CALCITONIN; FOR RENAL & CARDIOVAS.DIS.

6. HEMODIALYSIS; FOR RENAL FAILURE

MAGNESIUM

• Normal range :1.5-2.5 meq/l.

- MAINLY INTRACELLULAR LIKE POTASSIUM
- DEFICIENCY SIMILAR TO CALCIUM DEFICIENCY OR POTASSIUM EXCESS [NEUROMUSCULAR HYPERACTIVITY]
- NECESSARY FOR ENZYME FUNCTIONS

Hypomagnesaemia: LOW MAGNESIUM CLINICAL PICTURE LIKE LOW CALCIUM

- 1. EXAGERATED TENDON REFLEXES
- 2. CHVOSTEK SIGN
- 3. TETANY
- causes :TPN ,Renal failure , diarrhea , vomiting
 IT IS ALWAYS ASSOCIATED WITH Ca AND K DEFICIENCY
 REPLACED BY Mg sulfate or chloride solution , IV MgSO4 .

Always remember that its impossible to correct hypokalemia without correction of the Mg.

HYPERMAGNESEIMA : HIGH MAGNESIUM LEVEL CLINICALY; [LIKE HIGH Ca] ECG CHANGES [LIKE HIGH K]

- RENAL FAILURE : IN ASSOCIATION WITH HIGH K, HIGH Na, LOW Ca.
- CLINICALY;[LIKE HIGH Ca] WEAKNESS,LOSS OF REFLEXES, PARALYSIS, COMA, DEATH.
- ECG CHANGES LIKE HIGH K.
- TRETMENT : CACIUM GLUCONATE IV , INSULIN +GLUCAOSE , DIALYSIS.

PHOSPHATE (2.5-4.5 mg/dl)

- HYPERPHOSHATEMIA :Mostly due to Renal failure.
- Majority are asymptomatic , may have symptoms associated with hypocalcaemia.

• Treatment : sevelamer chloride, a phosphate binder , low phosphate diet , dialysis

HYPOPHOSPHATEMIA Critical value is less then 1 mg/dl

- Mostly due to re-feeding syndrome, ICU patients.
- Clinically : muscle weakness , failure to wean from the ventilator , infection risk due to poor leukocyte chemo-taxis from low ATP , encephalopathy.
- Treatment : potassium phosphate.

ACID – BASE BALANCE

- RESPIRATORY + METABOLIC
- NORMAL VALUES :
- PH = 7.35-7.45
- PCO2 = 35-45
- HCO3 = 22-26
- LUNG : CO2 REGULATION , RAPID EFFECT .
- **KIDNEY** : HCO3 REGULATION , SLOW EFFECT .

RESPIRATORY

• ALKALOSIS : HYPERVENTILATION , low pCO2.

• ACIDOSIS : POOR MINUTE VENTILATION (COPD, CNS DEPRESSION PTX , PLEURAL EFFESION, MORHPINE), high pCO2.

METABOLIC

METABOLIC ACIDSOSIS

• Loss of bicarbonate : diarrhea , ileus , fistulas, high output ileostomy.

Increase in acid : lactic acidosis , DKA , Renal failure .

METABOLIC ALKALOSIS

- Vomiting , NG suction , diuretics , Gastric Outlet Obstruction , pyloric stenosis.
- Loss of CL & H from stomach >> alkalosis , hypocholermic .
- Water loss > >kidney >> NA/K exchange >> hypokalemia
- Hypochloermic hypokalemic metabolic alkalosis
- KEY to correct CL deficit >> give N.S

FLUID MANAGEMENT

- IV fluid replacement includes maintenance, deficit, and ongoing losses.
- o Calculation of maintenance:
- 100/50/20 rule:
- Ist 10 kg: 100 mL/kg/day
- 2nd 10 kg: 50 mL/kg/day
- Rest: 20 mL/kg/day
- 4/2/1 rule (per hour)
- > 1st 10 kg: 4mL/kg/hour
- 2nd 10 kg 2 mL/kg/hour
- Rest: 1mL/kg/hour

Solutions

- o Crystalloids:
- Isotonic
- Hypertonic
- Hypotonic
- o Colloids:
- Albumin
- Dextran
- Hydroxyl starch

Isotonic Crystalloids

- 0.9% normal saline and ringer's lactate.
- Distribute uniformly through the extracellular compartment.
- After one hour, only 25% of the total volume remains in the intravascular space.
- The lactate in Ringer's lactate will be converted to bicarbonate >Patients will become alkalotic.
- Ringer's lactate is designed to mimic the extracellular fluid; it is called a balanced salt solution.
- o The most common trauma resuscitation fluid is Ringer's lactate

Hypertonic solutions

- Examples include 7.5% normal saline and 3% normal saline.
- o Indications:
- Used in shock/burns (usually in combination with colloids like dextran)

• o Effects:

- ✓ Studies have shown that it causes significant blunting of neutrophil activation with a transient increase in serum sodium that normalizes within 24 hours.
- ✓ This effect may help in decreasing widespread tissue damage and multiorgan dysfunction seen after a traumatic injury.
- o Side effects:
- Hypernatremia (hyperosmolarity) and hyperchloremia
- Hypokalemia
- Central pontine demylenation

Hypotonic solutions

• Examples include D5W and 0.45% NaCl.

 Should not be sued for volume expansion, because they only expand 10% of the infused volume.

• Indicated to replace free water deficits.

FLUID MANAGEMENT

IF ELECTROLYTES ARE NORMAL REPLACEMENT IS BY 1-N/S WHICH CONTAINS: Na 154 meq Cl 154 meq that may reduce the

Cl 154 meq that may reduce the ph.value WHICH IS GOOD IN STOMACH OUTLET OBSTRUCTION [FOR CORRECTION OF ALKALOSIS] but might cause acidosis in normal PH

OR 2- RINGER LACTATE WHICH CONTAINS: Na 130 meq , k 4 meq , Ca 3 meq Cl 109 meq Lactate 28 meq that changes to bicarbonate in the liver Suitable for metabolic acidosis

Fluid loss associated with alkalosis: Replace with N/S Fluid loss associated with acidosis: Replace with Ringer lactate

Colloid solutions

- They contain **high molecular weight substances** that remain in the intravascular space
- More **expansive** than crystalloids.

• INDICATIONS:

When crystalloids fail to sustain plasma volume. This is due to the low colloid osmotic pressure in <u>burn</u> patients and in cases of <u>peritonitis</u>.

• SIDE EFFECTS:

- Pulmonary edema
- Renal failure
- Bleeding disorders
- Early use of colloids in the resuscitation regimen may result in more prompt resuscitation of tissue perfusion. Moreover, it might decrease the total volume of required fluids.
- Albumin preparations: 5% or 25% albumin. Indicated for volume expansion.
- However, they are not indicated for patients with adequate colloid oncotic pressure (albumin >2.5)

Thank you