



Biochemistry of Carbohydrates II

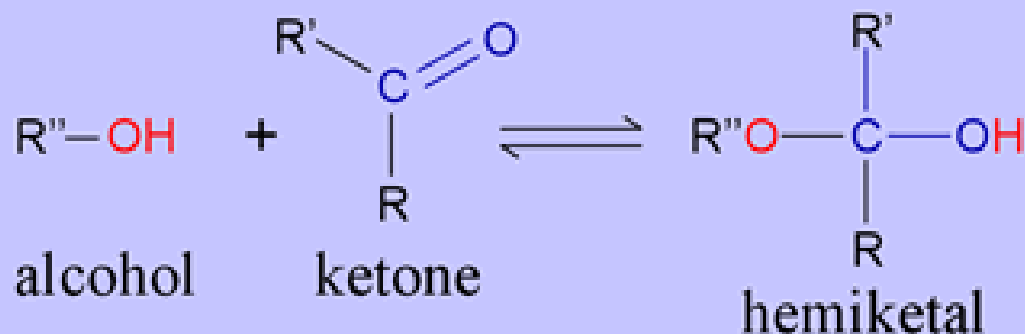
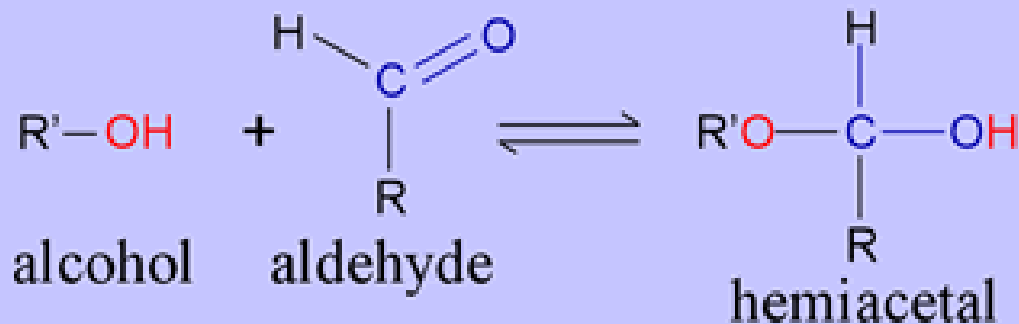


Dr. Nesrin Mwafi

Biochemistry & Molecular Biology Department

Faculty of Medicine, Mutah University

Hemiacetal & Hemiketal



Anomers

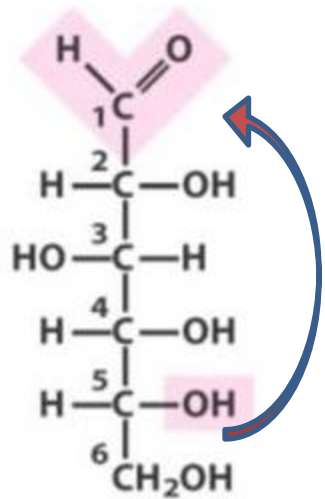


- Anomers are pair of stereoisomers that differ in spatial arrangement of atoms at the anomeric carbon

Monosaccharide cyclization

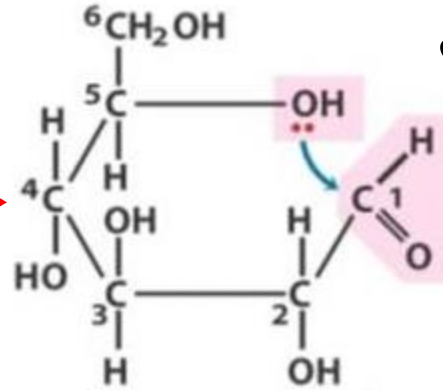
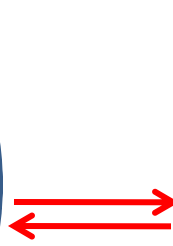


Linear form



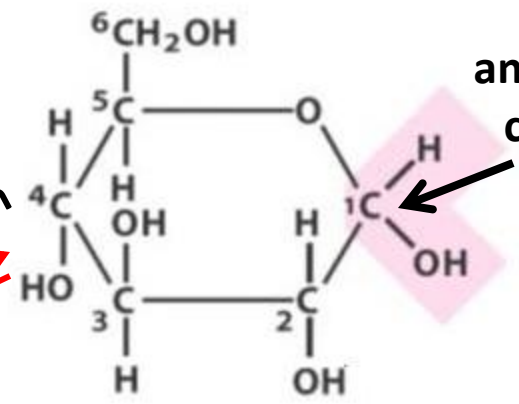
D-glucose

Fisher projection

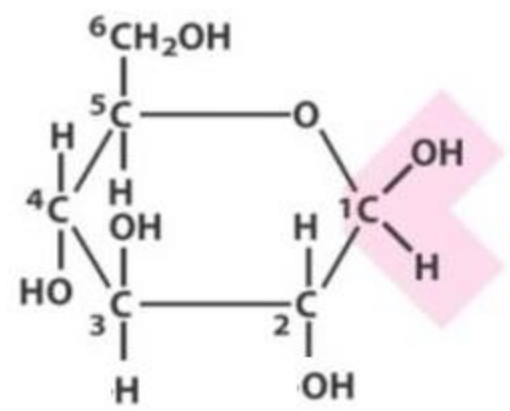


cyclization

cyclization



α -D-glucose



β -D-glucose

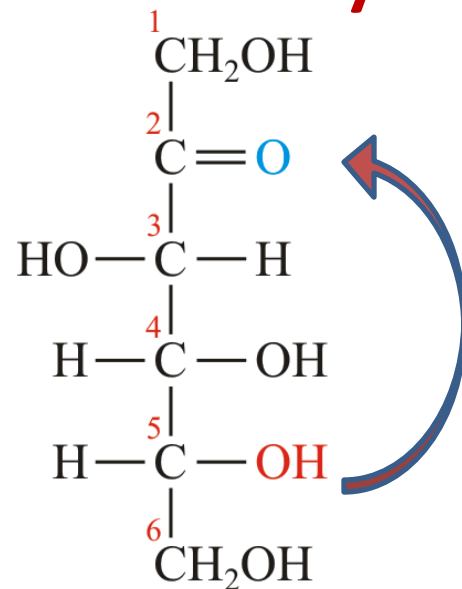
Haworth projection

Anomers

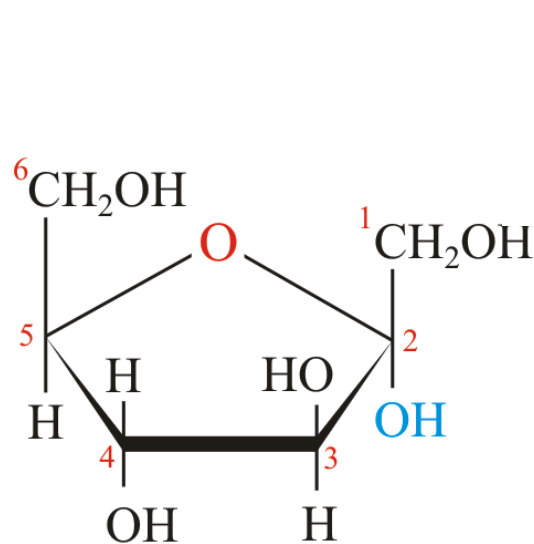


- ❑ Anomers are pair of stereoisomers that differ in spatial arrangement of atoms at the anomeric carbon
- ❑ In cyclic sugars, the carbonyl carbon becomes a chiral center (asymmetric carbon) with two possible configurations: α and β . This new carbon is called anomeric carbon.
- ❑ In α -anomer, the OH group of the anomeric carbon is projecting down the plane of the ring and on the opposite side of the terminal CH_2OH group (in Fisher projection) and vice versa in β -anomer.
- ❑ The anomers freely interconvert in aqueous solution, e.g. at equilibrium D-glucose is a mixture of β -anomer (63.6%), α -anomer (36.4%) and extremely tiny amounts of the straight chain.

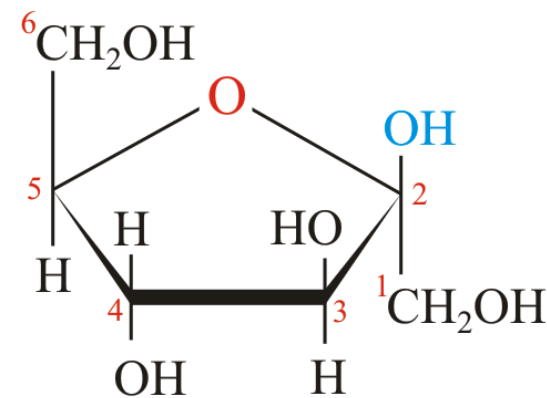
Monosaccharide cyclization



D-fructose
Linear form



α -D-fructose

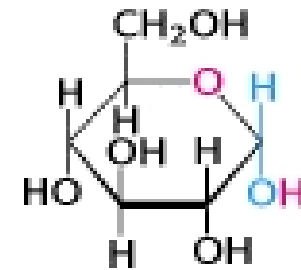
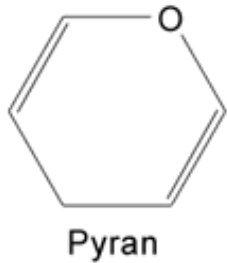


β -D-fructose

Pyranoses & Furanoses

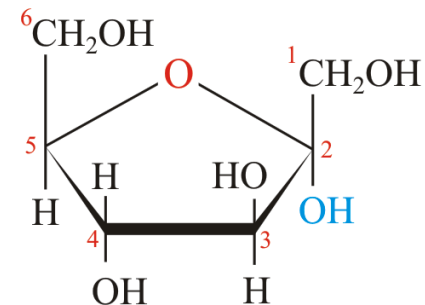
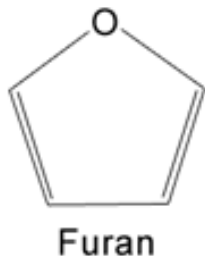


- ❑ Sugars with six-membered rings are known as pyranoses (e.g. glucopyranose) as they resemble the heterocyclic compound pyran.



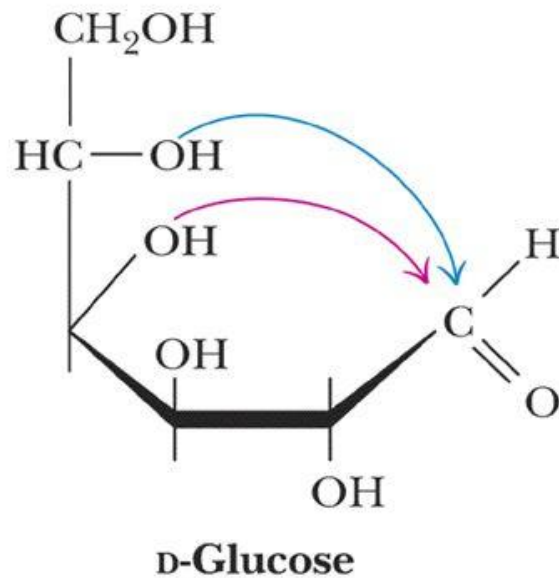
α -D-glucopyranose

- ❑ Sugars with five -membered rings are known as furanoses (e.g. fructofuranose) as they resemble the heterocyclic compound furan.

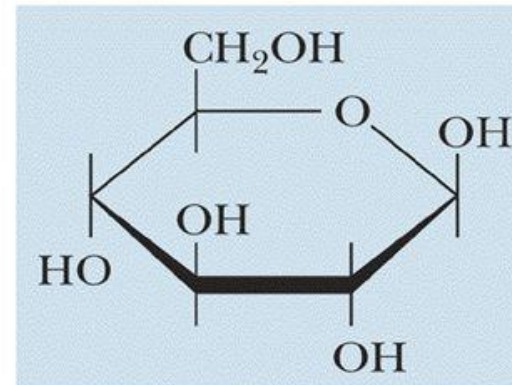


α -D-fructofuranose

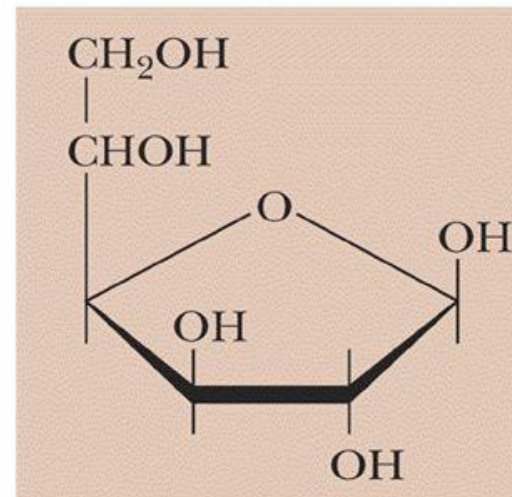
Pyranoses & Furanoses



D-glucose can cyclize in two ways forming either furanose or pyranose structures



Pyranose form
 β -D-glucopyranose

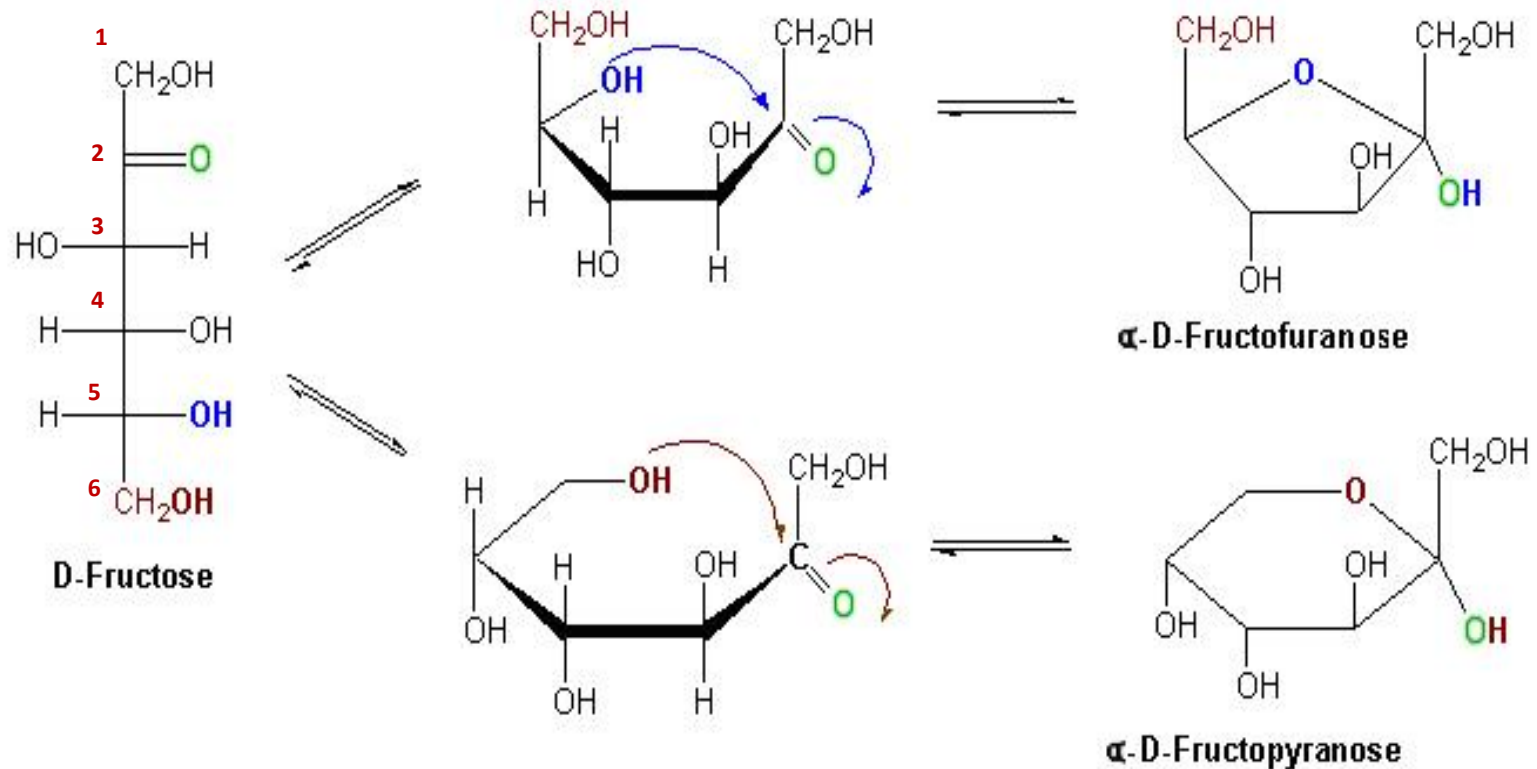


Furanose form
 β -D-glucofuranose

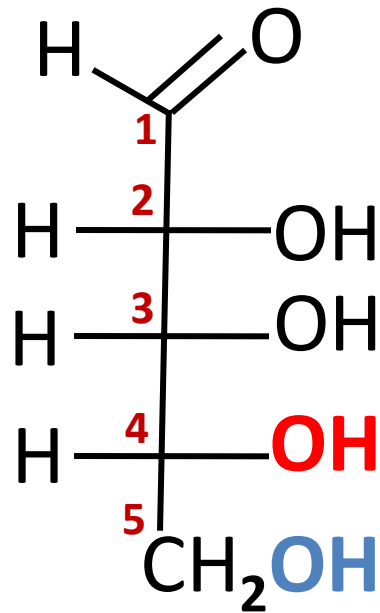
Pyranoses & Furanoses



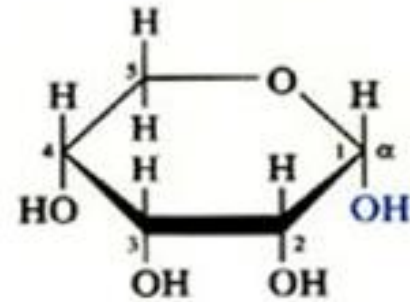
Isomeric Forms of Fructose



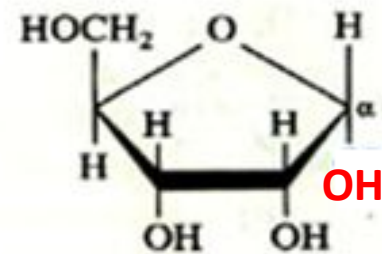
Pyranoses & Furanoses



D-ribose
Fisher projection



α-D-Ribopyranose
(Haworth projection)



α-D-Ribofuranose
(Haworth projection)

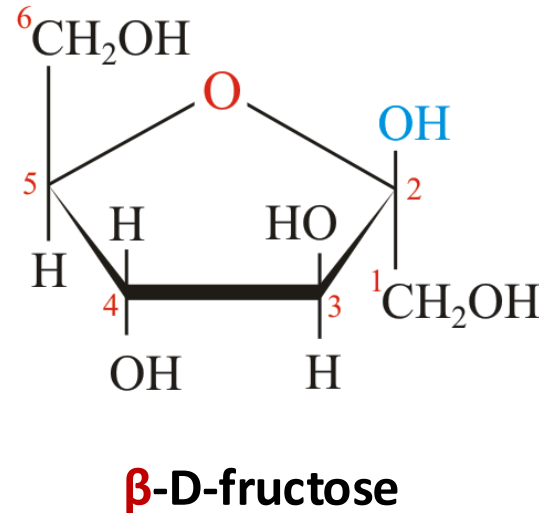
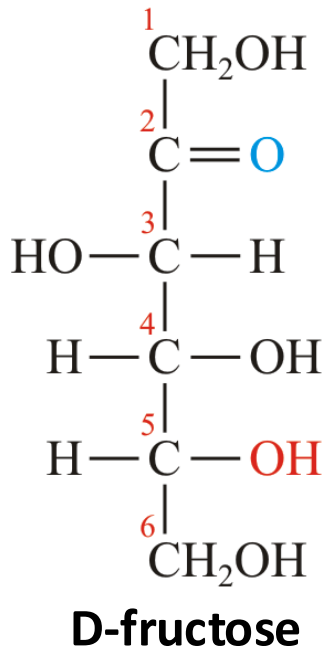
□ Hexose or pentose can exist in pyranose and furanose forms (the most stable rings). e.g. in solution, glucose and fructose are **mostly** pyranoses whereas ribose is **mostly** furanose

Haworth Projection



- Haworth projection is a simple 3D way to represent the cyclic monosaccharides.

Fisher projection



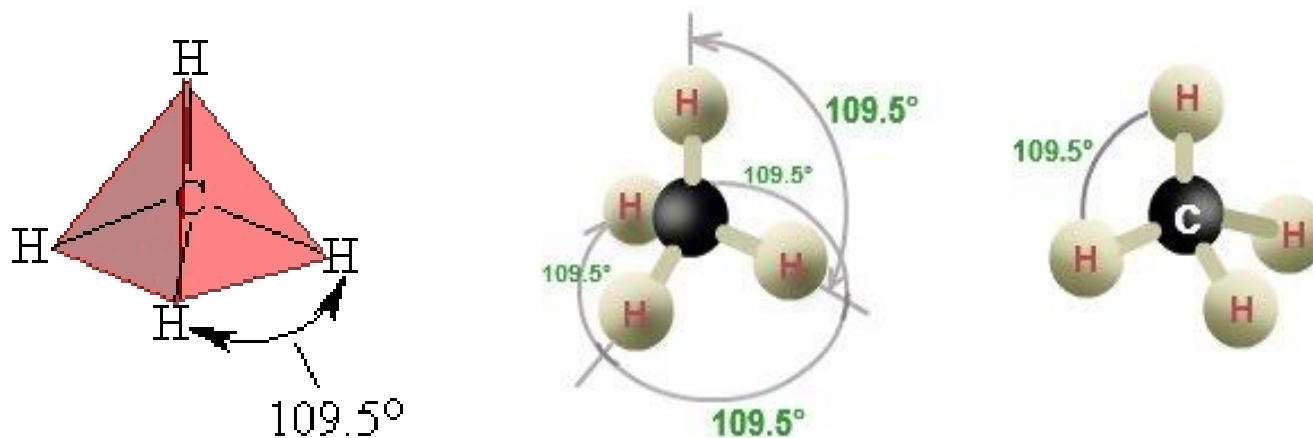
Haworth projection

Conformers

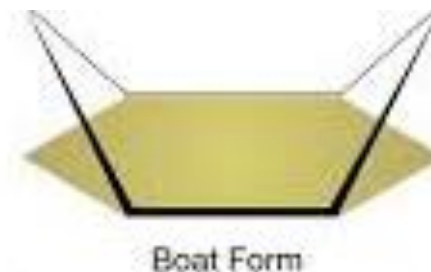
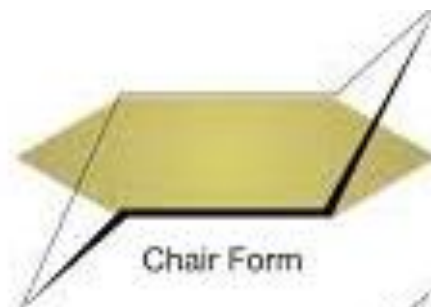


- The geometry of the carbon atoms of monosaccharide ring is tetrahedral (bond angles are close to 109.5°), so sugar rings are not actually planar. For example, pyranoses take on either Chair or Boat conformations (conformational isomers or conformers).

Conformers



Carbon atoms are tetrahedral

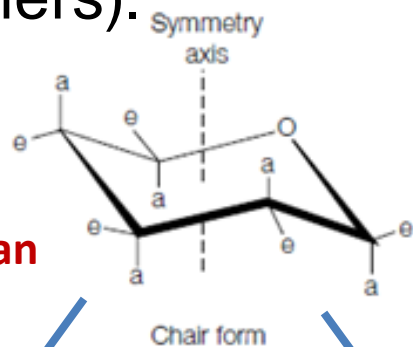


Conformers are stereoisomers with different rotations about single bonds

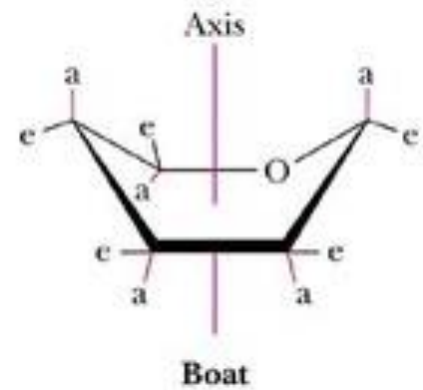
Conformers



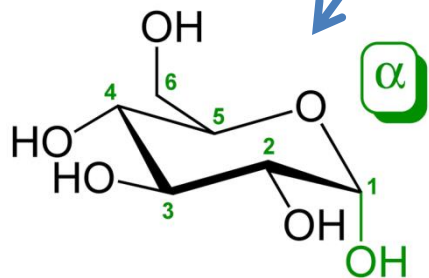
□ The geometry of the carbon atoms of monosaccharide ring is tetrahedral (bond angles are close to 109.5°), so sugar rings are not actually planar. For example, pyranoses take on either **Chair** or **Boat** conformations (conformational isomers or conformers).



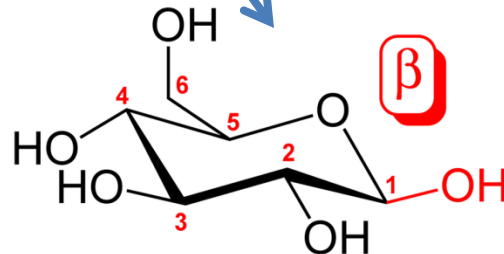
a: axial
e: equatorial



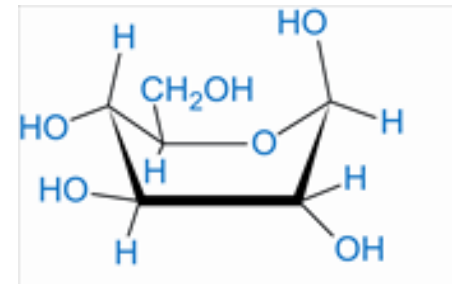
α is Less stable than β due to steric repulsion



Chair form of α -D-glucose



Chair form of β -D-glucose



Boat form of β -D-glucose

Sugar Modification

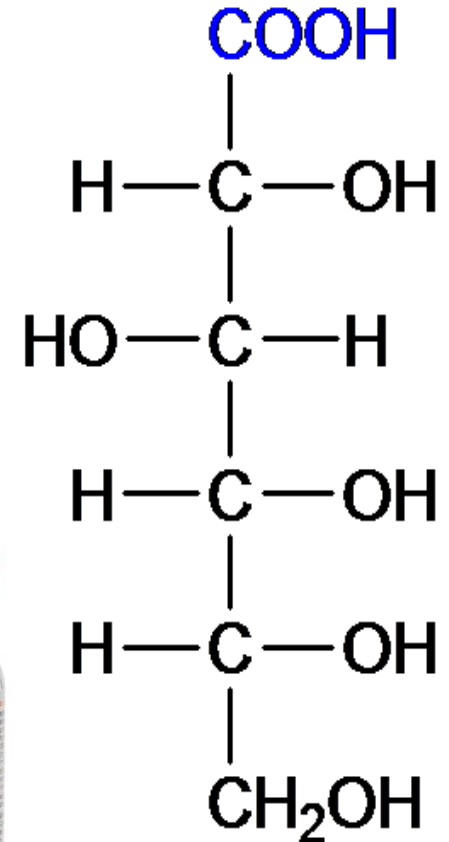


1. **Aldonic acids** : oxidation of aldehyde (C1) to carboxylic acid; e.g. D-gluconic acid

- **Uses:**

- Some drugs are injected in the form Of **gluconate** (the salt of gluconic acid)

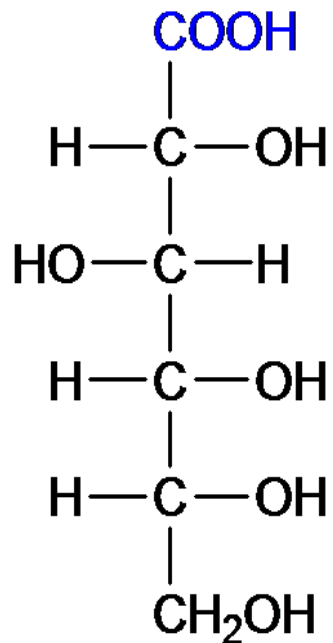
- **Calcium gluconate solution (I.V)** as cardioprotective agent in patients with high blood level of K^+ (6.5 mmol/L) occurring due to kidney failure



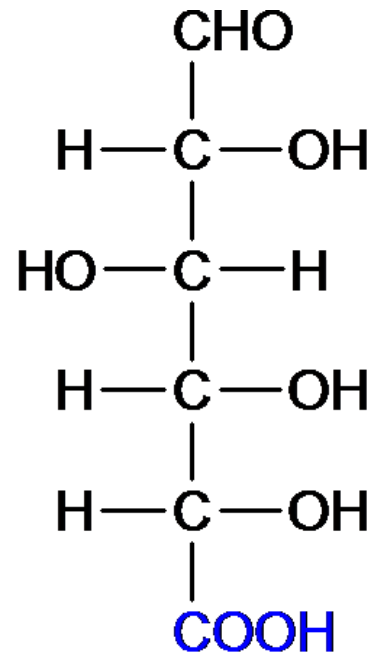
Sugar Modification



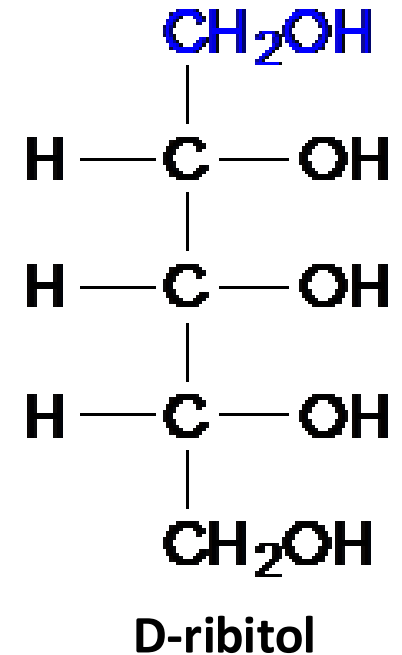
1. Aldonic acids : oxidation of aldehyde (C1) to carboxylic acid; e.g. D-gluconic acid



2. Uronic acids : oxidation of OH at (C6) to carboxylic acid; e.g. D-glucuronic acid



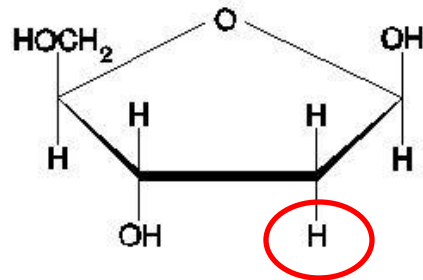
3. Alditols : reduction of carbonyl group to alcohol; e.g. D-ribitol, D-glycerol and D-sorbitol (sweetener)



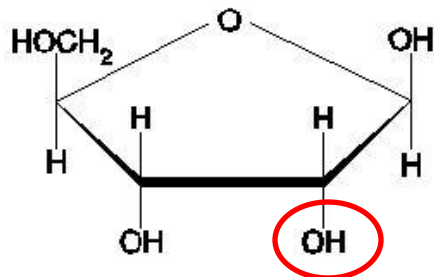
Sugar Modification



4. Deoxy sugars : OH group is replaced by H; e.g. β -D-2-deoxyribose



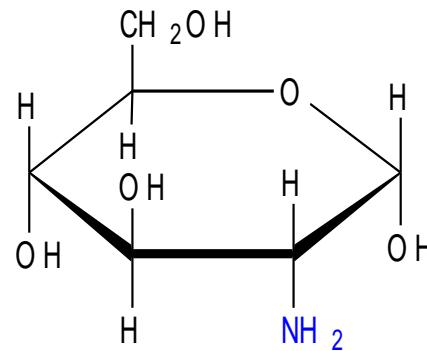
Deoxyribose



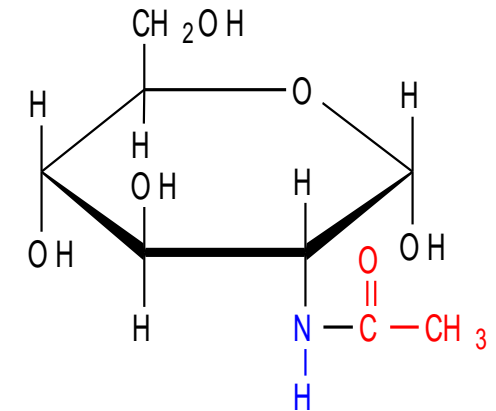
Ribose

4. Amino sugars : one or more OH groups are replaced by **amino group** which is often **acetylated**;

- α -D-glucosamine (rebuild cartilage in osteoarthritis & osteoporosis) and α -D-N-acetylglucosamine (both are derivatives of α -D-glucose)



α -D-glucosamine



α -D-N-acetylglucosamine

