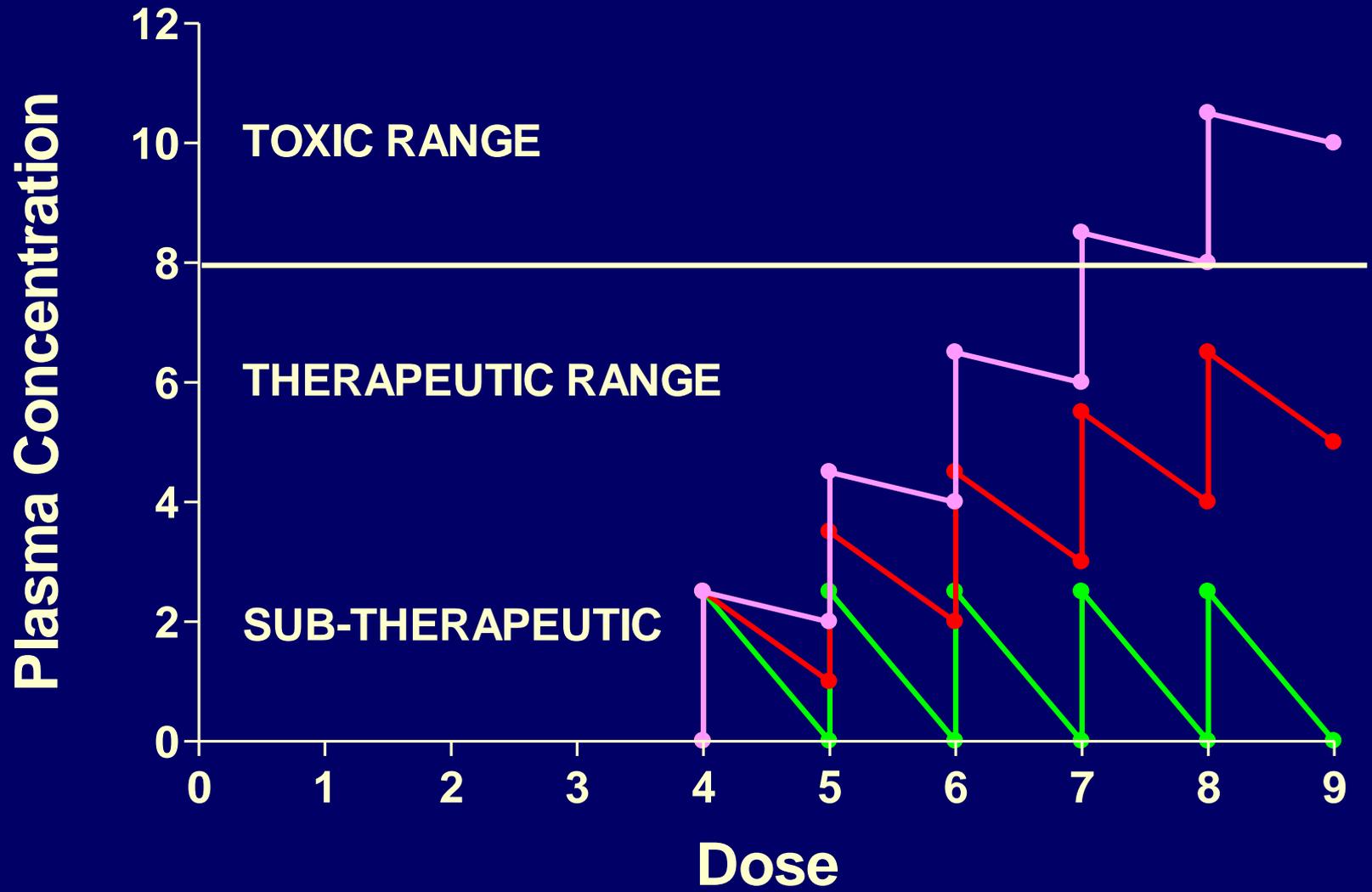
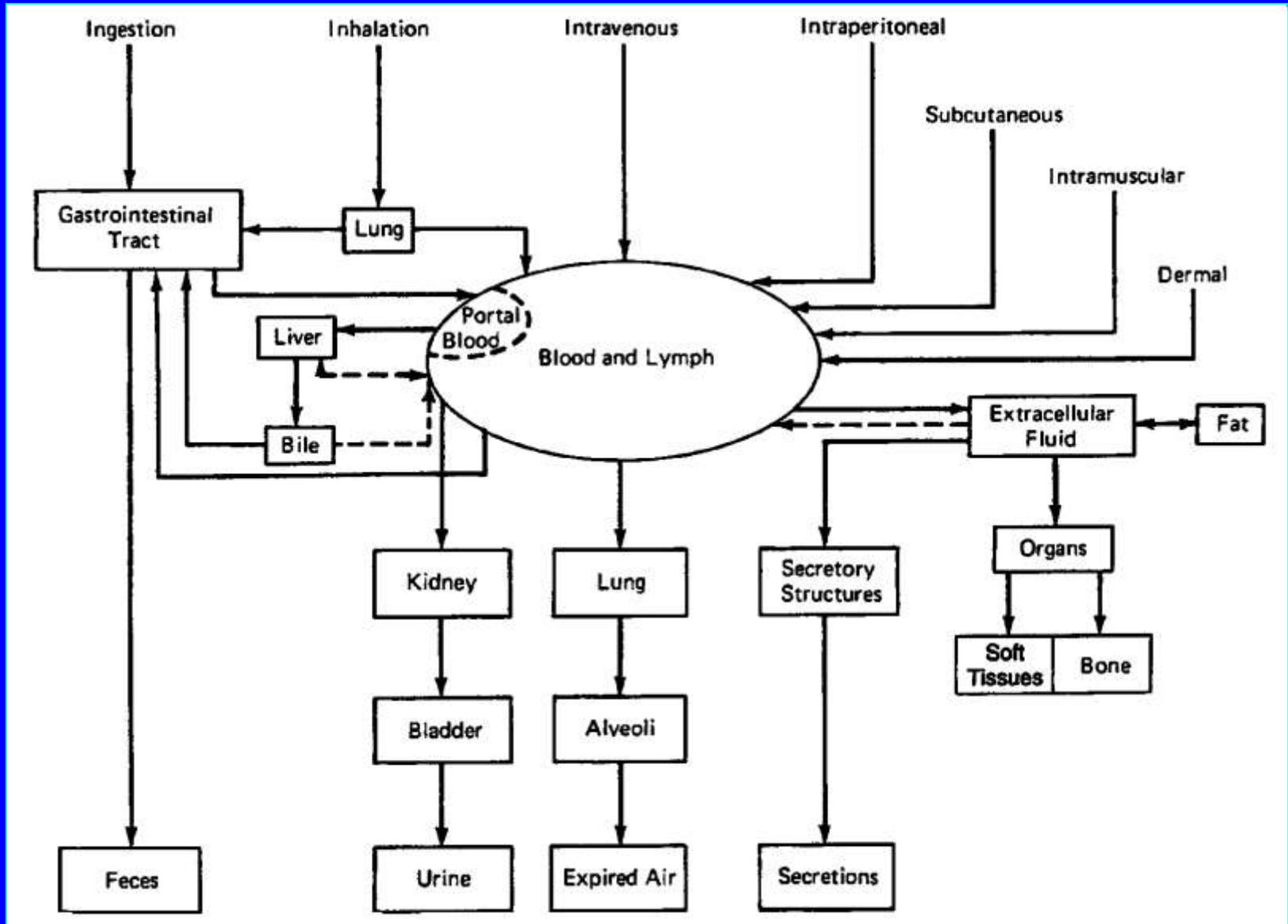


Pharmacokinetics

Pharmacodynamics



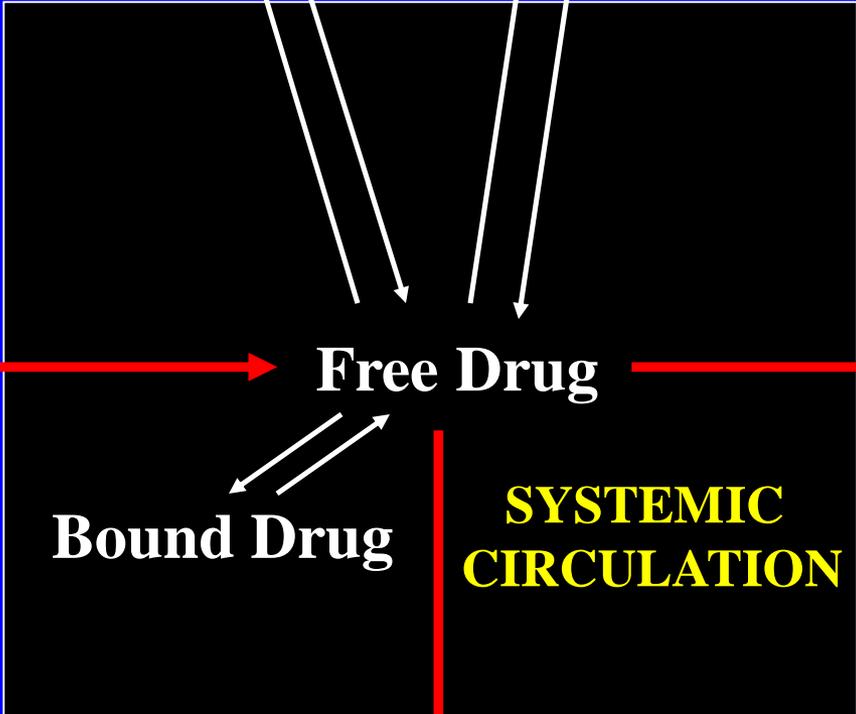
DISPOSITION OF DRUGS



The disposition of chemicals entering the body (from C.D. Klaassen, *Casarett and Doull's Toxicology*, 5th ed., New York: McGraw-Hill, 1996).

**LOCUS OF ACTION
"RECEPTORS"**
Bound \rightleftharpoons Free

**TISSUE
RESERVOIRS**
Free \rightleftharpoons Bound

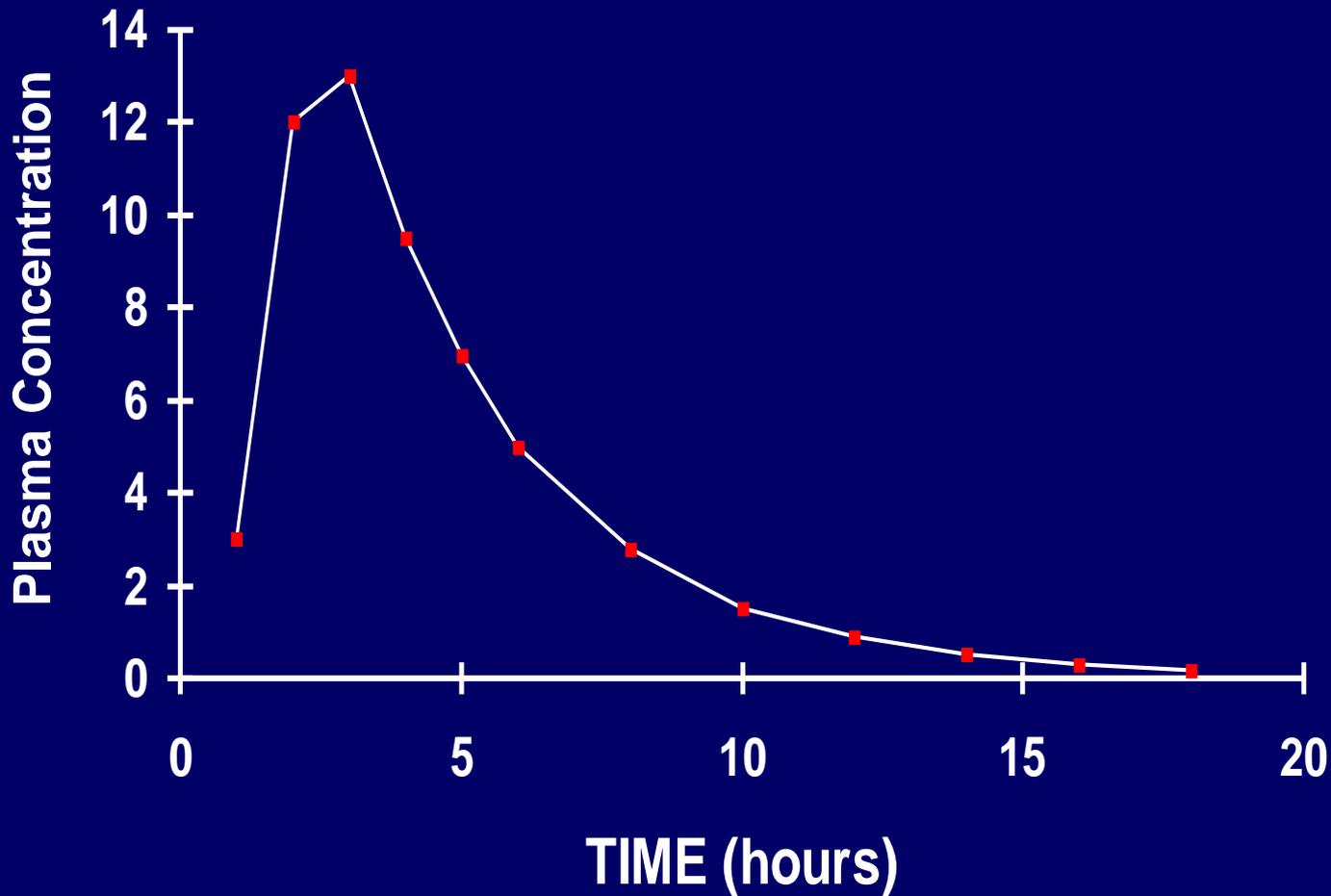


ABSORPTION

EXCRETION

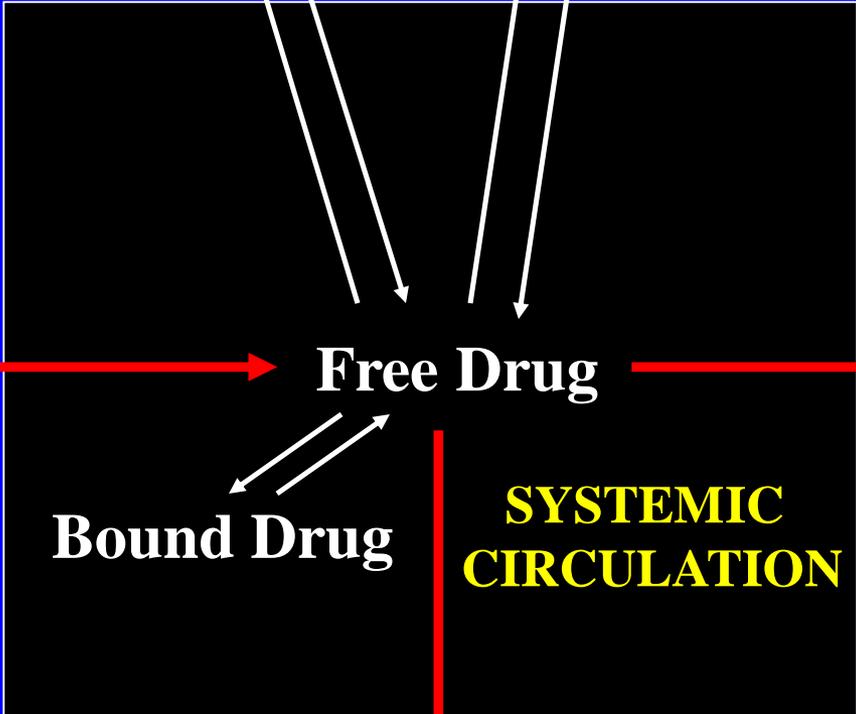
BIOTRANSFORMATION

Plasma concentration vs. time profile of a single dose of a drug ingested orally



**LOCUS OF ACTION
"RECEPTORS"**
Bound \rightleftharpoons Free

**TISSUE
RESERVOIRS**
Free \rightleftharpoons Bound



ABSORPTION

Free Drug

EXCRETION

Bound Drug

**SYSTEMIC
CIRCULATION**

BIOTRANSFORMATION

Bioavailability

Definition: the fraction of the administered dose reaching the systemic circulation

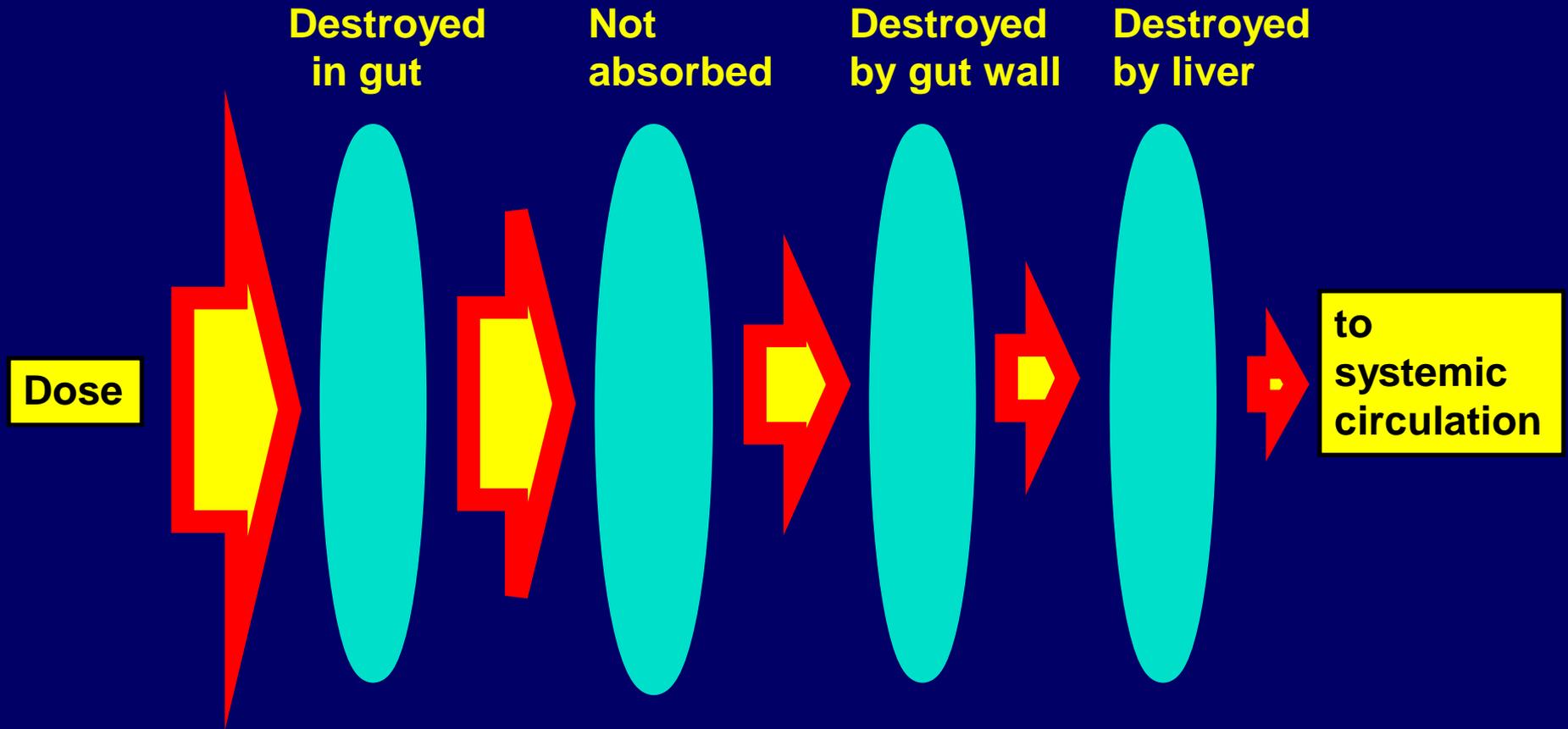
for i.v.: 100%

for non i.v.: ranges from 0 to 100%

e.g. lidocaine bioavailability 35% due to destruction in gastric acid and liver metabolism

First Pass Effect

Bioavailability



PRINCIPLE

For drugs taken by routes other than the i.v. route, the extent of absorption and the **bioavailability** must be understood in order to determine what dose will induce the desired therapeutic effect. It will also explain why the same dose may cause a therapeutic effect by one route but a toxic or no effect by another.

PRINCIPLE

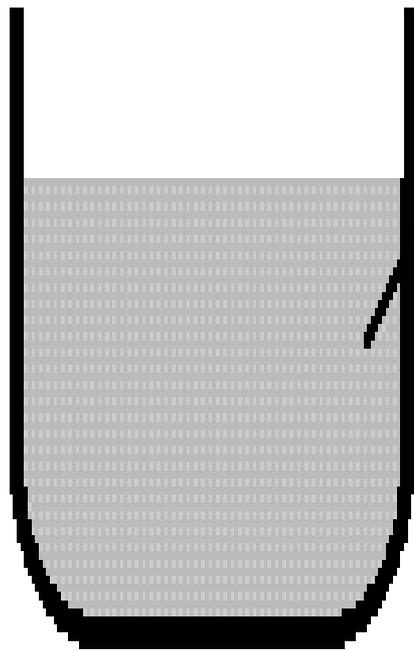
Drugs appear to distribute in the body as if it were a single compartment. The magnitude of the drug's distribution is given by the apparent volume of distribution (V_d).

$$V_d = \text{Amount of drug in body} \div \text{Concentration in Plasma}$$

(Apparent) Volume of Distribution:

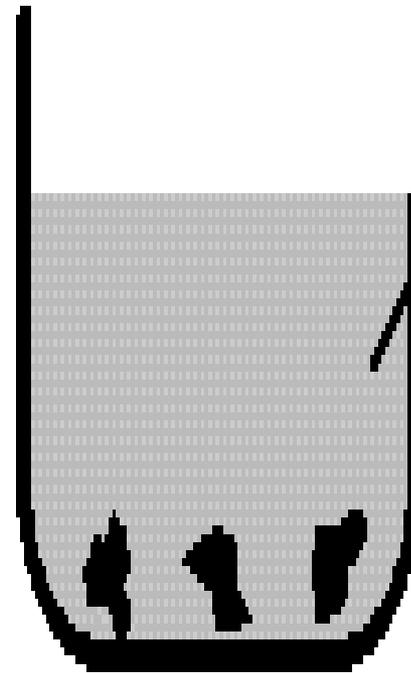
Volume into which a drug appears to distribute with a concentration equal to its plasma concentration

Drug concentration in beaker:



Dose = 10 mg
 $C_p^0 = 20 \text{ mg/L}$
Apparent
Volume = 500 ml

With charcoal in beaker:



Dose = 10 mg
 $C_p^0 = 2 \text{ mg/L}$
Apparent
Volume = 5000 ml

Examples of apparent V_d 's for some drugs

Drug	L/Kg	L/70 kg
Sulfisoxazole	0.16	11.2
Phenytoin	0.63	44.1
Phenobarbital	0.55	38.5
Diazepam	2.4	168
Digoxin	7	490

Elimination of drugs from the body

**M
A
J
O
R**

KIDNEY

filtration
secretion
(reabsorption)

LIVER

metabolism
secretion

**M
I
N
O
R**

LUNGS

exhalation

OTHERS

mother's milk
sweat, saliva etc.

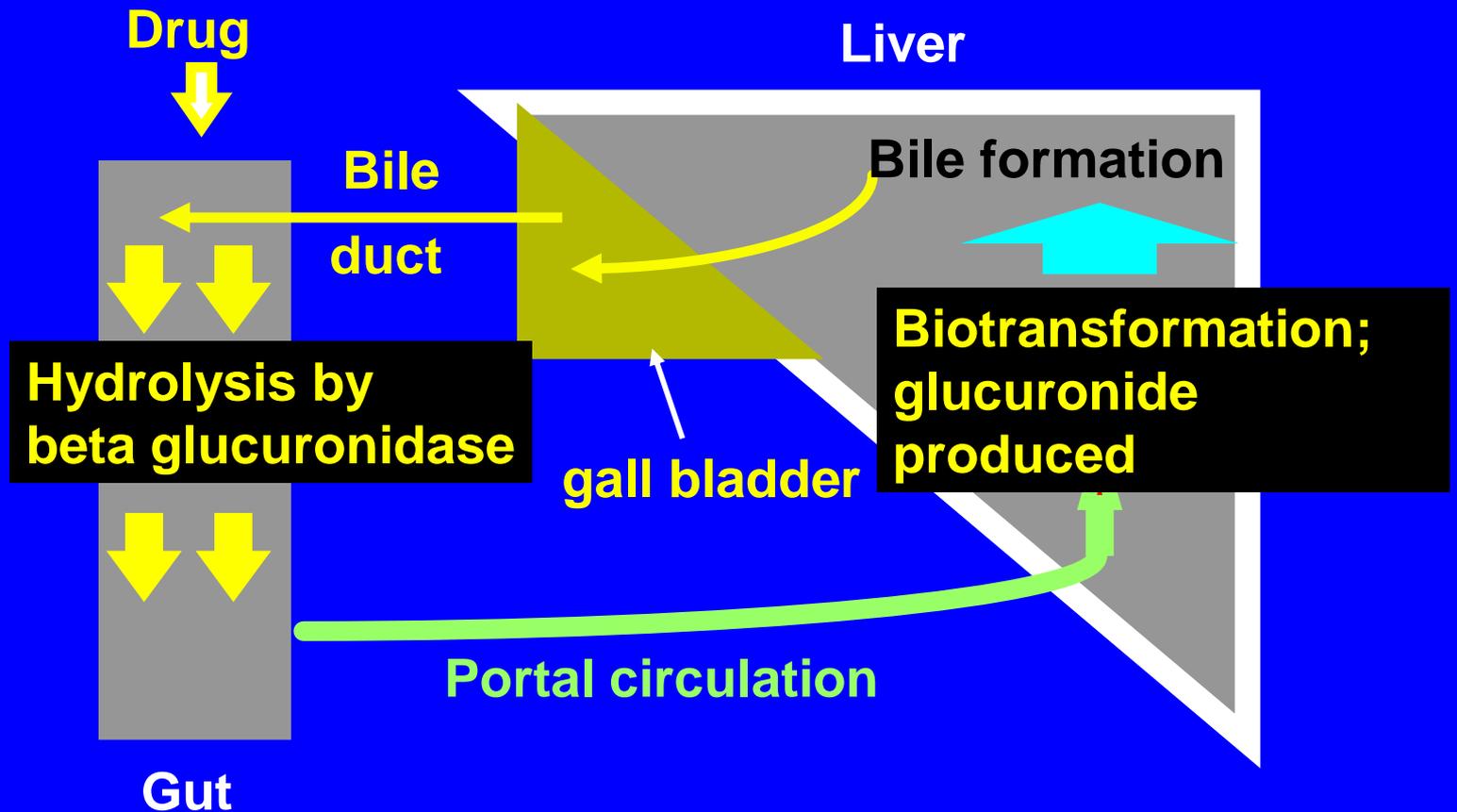
Elimination by the Kidney

- **Excretion - major**
 - 1) **glomerular filtration**
glomerular structure, size constraints, protein binding
 - 2) **tubular reabsorption/secretion**
 - **acidification/alkalinization,**
 - **active transport, competitive/saturable, organic acids/bases**
 - **protein binding**
- **Metabolism - minor**

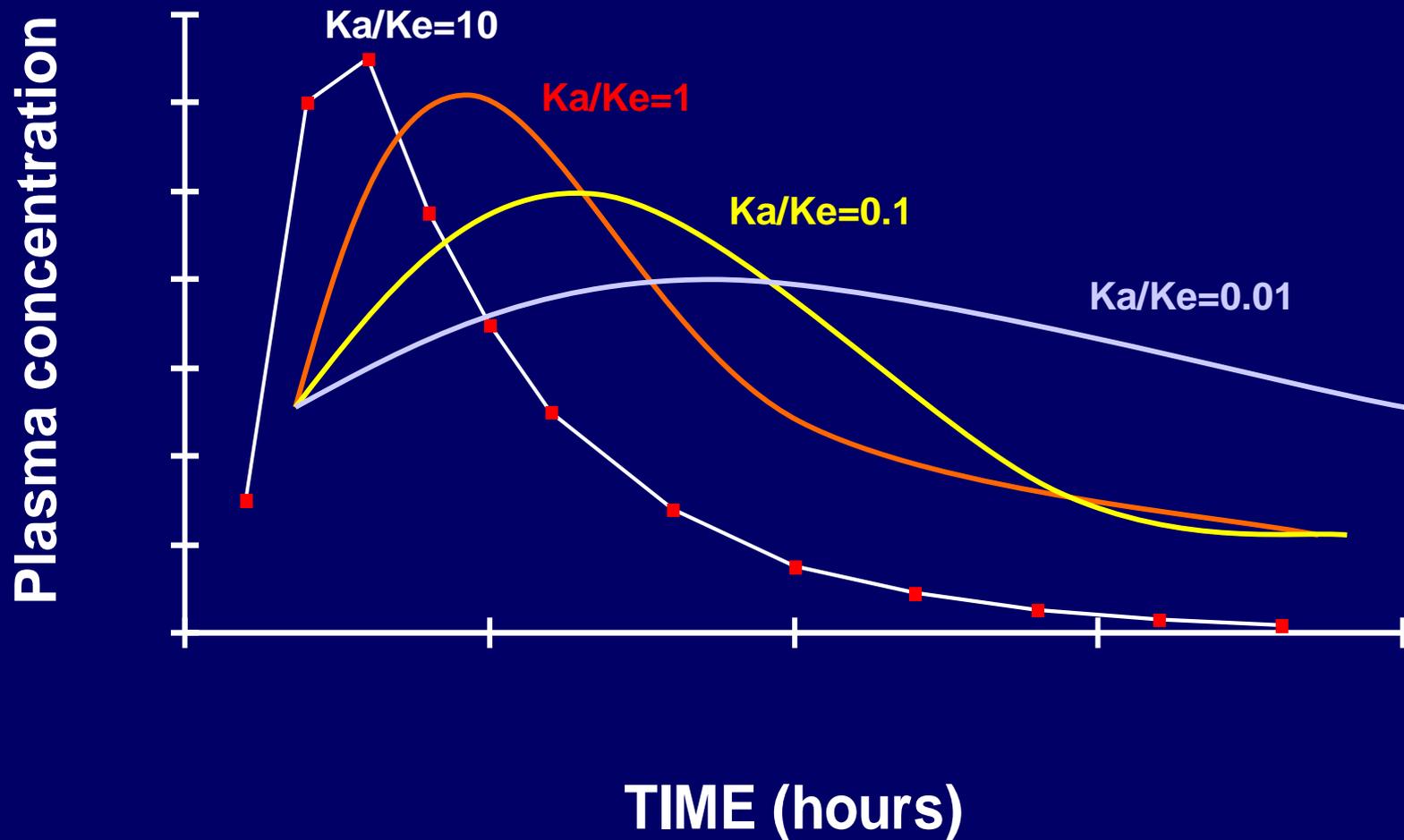
Elimination by the Liver

- Metabolism - major
 - 1) Phase I and II reactions
 - 2) Function: change a lipid soluble to more water soluble molecule to excrete in kidney
 - 3) Possibility of active metabolites with same or different properties as parent molecule
- Biliary Secretion – active transport, 4 categories

The enterohepatic shunt



Influence of Variations in Relative Rates of Absorption and Elimination on Plasma Concentration of an Orally Administered Drug



Elimination

- **Zero order:** constant rate of elimination irrespective of plasma concentration.
- **First order:** rate of elimination proportional to plasma concentration. Constant *Fraction* of drug eliminated per unit time.

Rate of elimination \propto Amount

Rate of elimination = $K \times$ Amount

Zero Order Elimination

Pharmacokinetics of Ethanol

- Ethanol is distributed in total body water.
- Mild intoxication at 1 mg/ml in plasma.
- How much should be ingested to reach it?

Answer: 42 g or 56 ml of pure ethanol ($V_d \times C$)

Or 120 ml of a strong alcoholic drink like whiskey

- Ethanol has a constant elimination rate = **10 ml/h**
- To maintain mild intoxication, at what rate must ethanol be taken now?

at 10 ml/h of pure ethanol, or 20 ml/h of drink.

Rarely Done → DRUNKENNES → Coma → Death

S

First Order Elimination

$$dA/dt \propto A$$

$$dA/dt = -k \cdot A$$

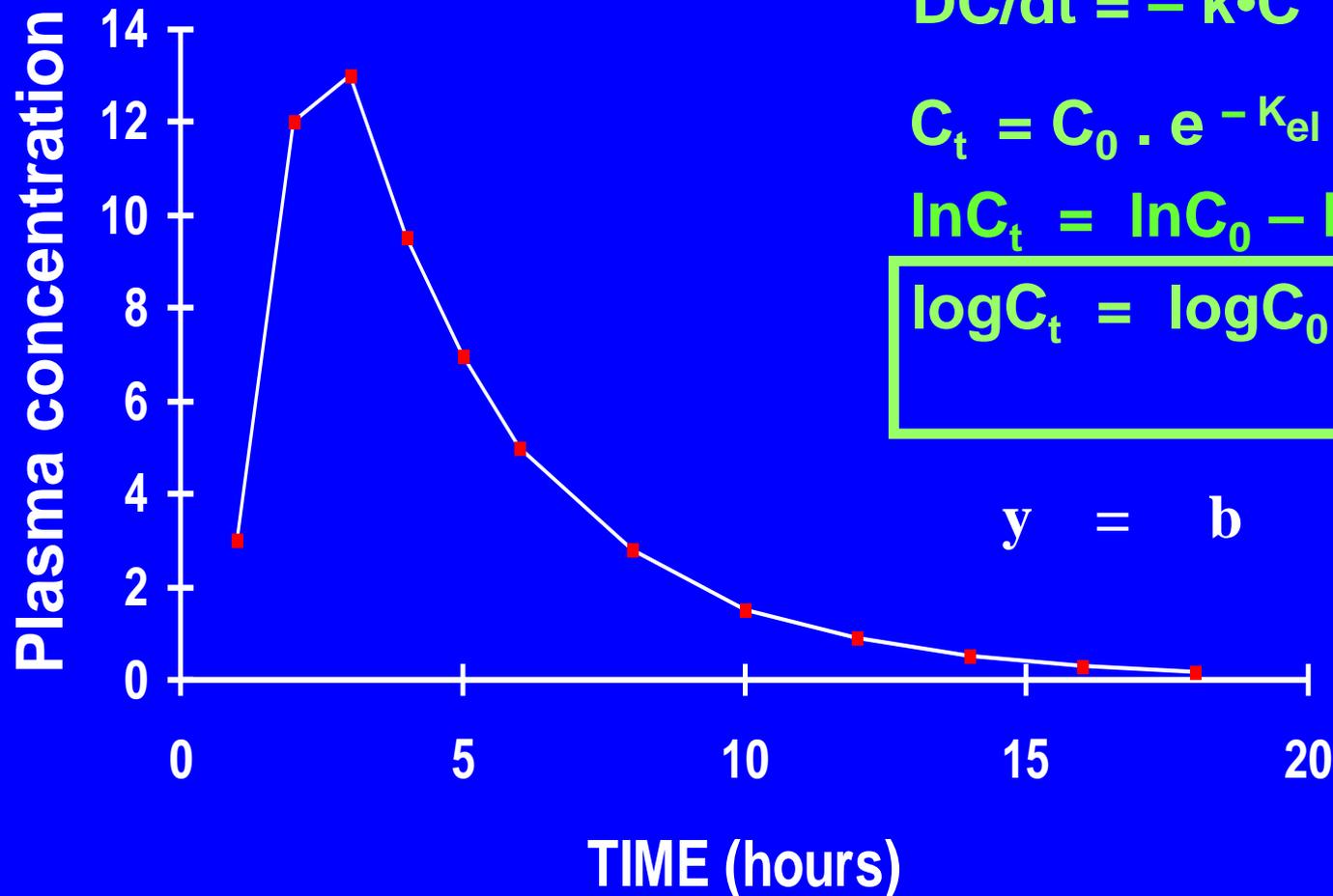
$$dC/dt = -k \cdot C$$

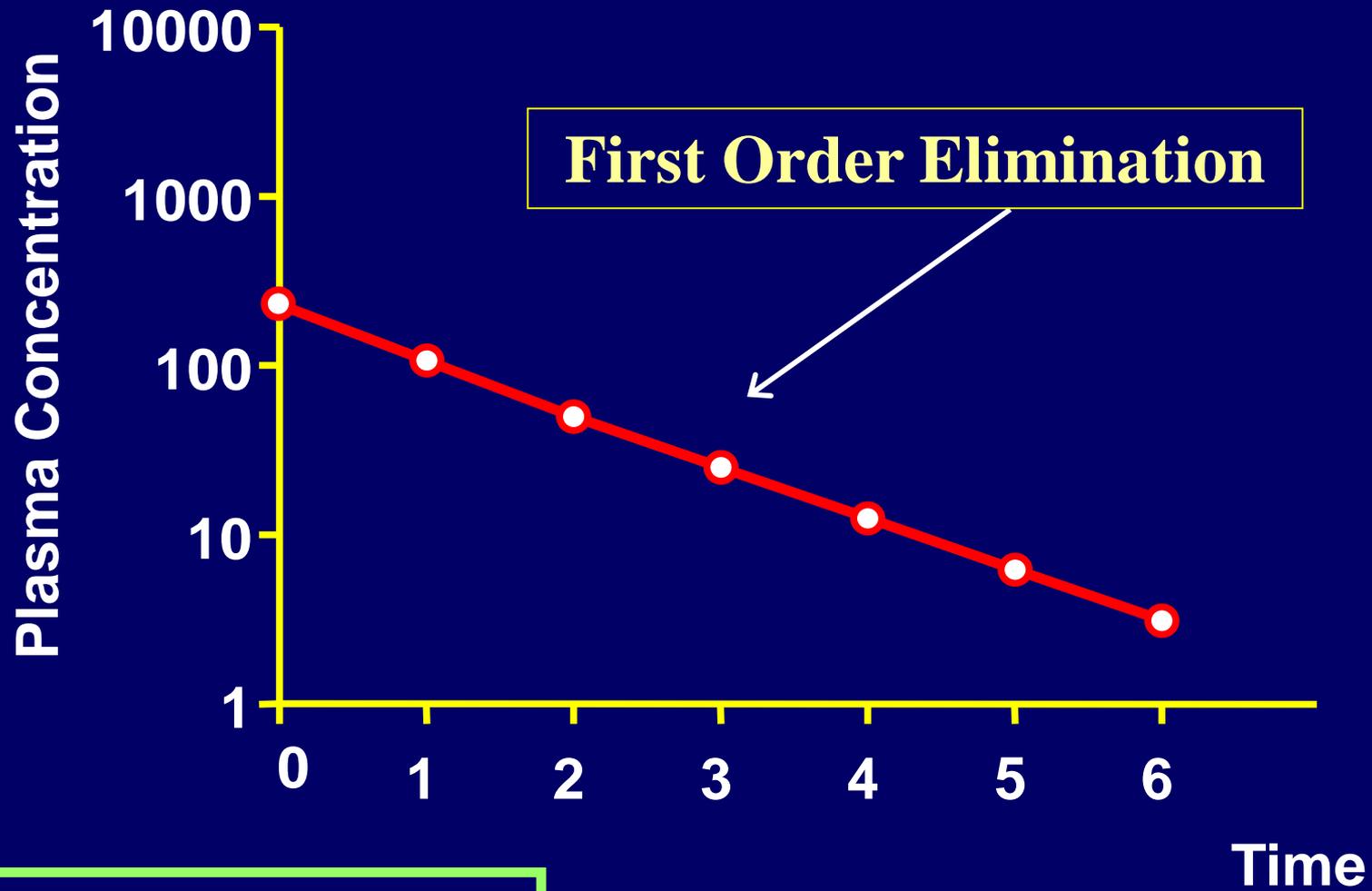
$$C_t = C_0 \cdot e^{-K_{el} \cdot t}$$

$$\ln C_t = \ln C_0 - K_{el} \cdot t$$

$$\log C_t = \log C_0 - \frac{K_{el}}{2.3} \cdot t$$

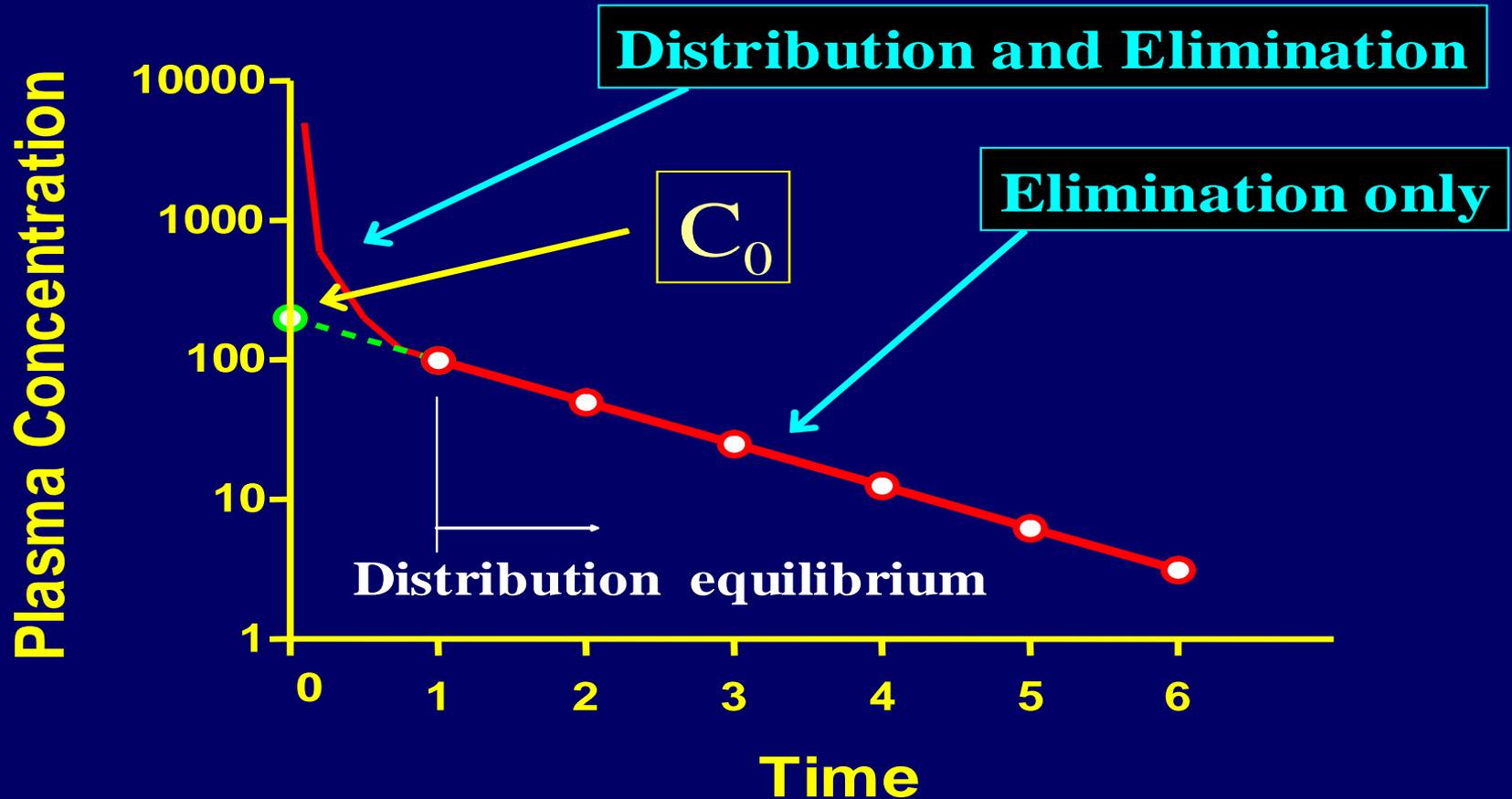
$$y = b - a \cdot x$$





$$\log C_t = \log C_0 - \frac{K_{el} \cdot t}{2.303}$$

Plasma Concentration Profile after a Single I.V. Injection



PRINCIPLE

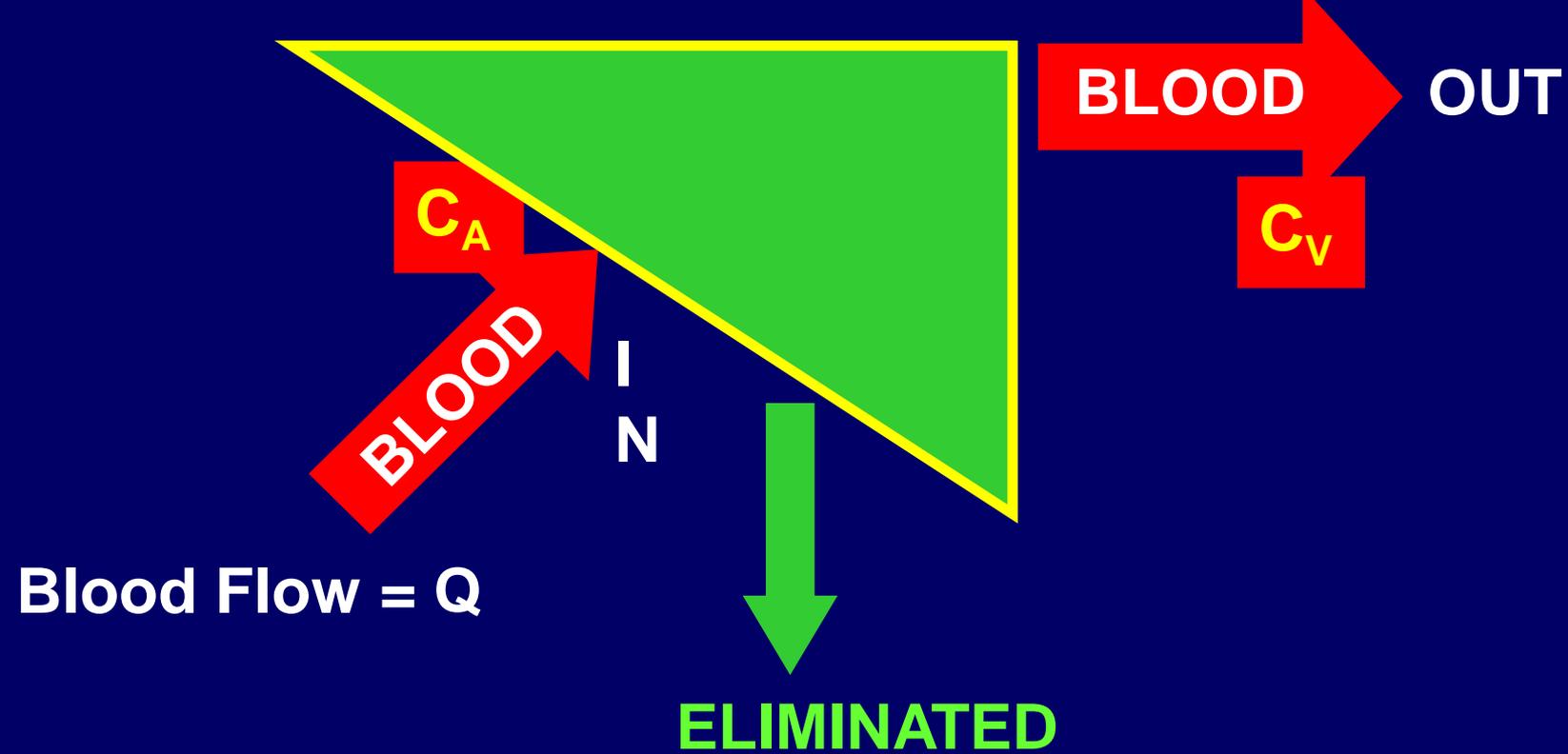
Elimination of drugs from the body usually follows first order kinetics with a characteristic half-life ($t_{1/2}$) and fractional rate constant (K_{el}).

First Order Elimination

- **Clearance:** volume of plasma cleared of drug per unit time.
Clearance = Rate of elimination ÷ plasma conc.
- **Half-life of elimination:** time for plasma conc. to decrease by half.

Useful in estimating:

- time to reach steady state concentration.
- time for plasma concentration to fall after dosing is stopped.



$$\text{Total Body Clearance} = CL_{\text{liver}} + CL_{\text{kidney}} + CL_{\text{lungs}} + CL_x$$

Rate of elimination = K_{el} x Amount in body
Rate of elimination = CL x Plasma Concentration

Therefore,

$$K_{el} \times \text{Amount} = CL \times \text{Concentration}$$



$$K_{el} = CL/V_d$$



$$0.693/t_{1/2} = CL/V_d$$



$$t_{1/2} = 0.693 \times V_d/CL$$

PRINCIPLE

The half-life of elimination of a drug (and its residence in the body) depends on its **clearance** and its **volume of distribution**

$t_{1/2}$ is proportional to V_d

$t_{1/2}$ is inversely proportional to CL

$$t_{1/2} = 0.693 \times V_d / CL$$

Multiple dosing

- On continuous steady administration of a drug, plasma concentration will rise fast at first then more slowly and reach a plateau, where:

rate of administration = rate of elimination
ie. steady state is reached.

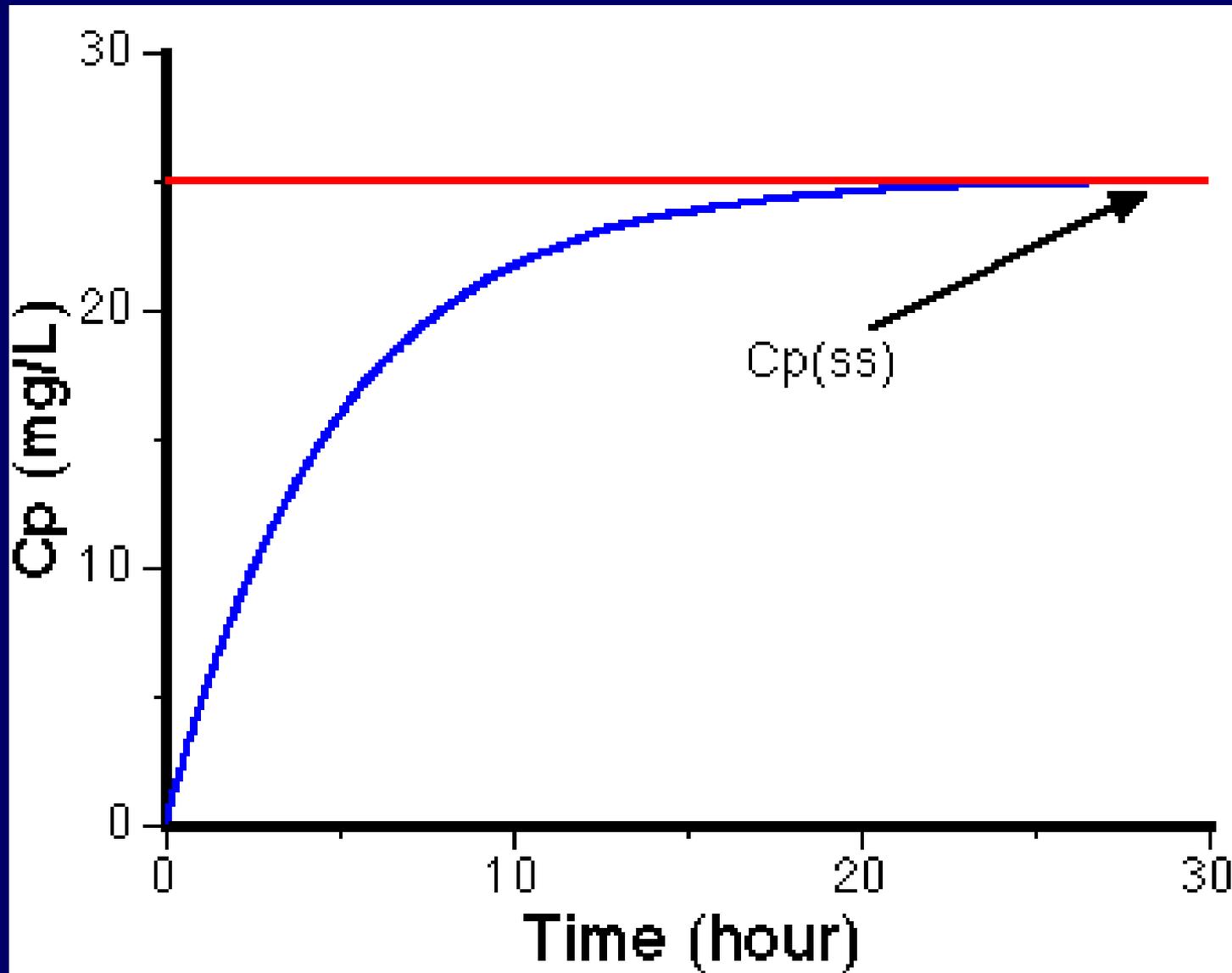
- Therefore, at steady state:

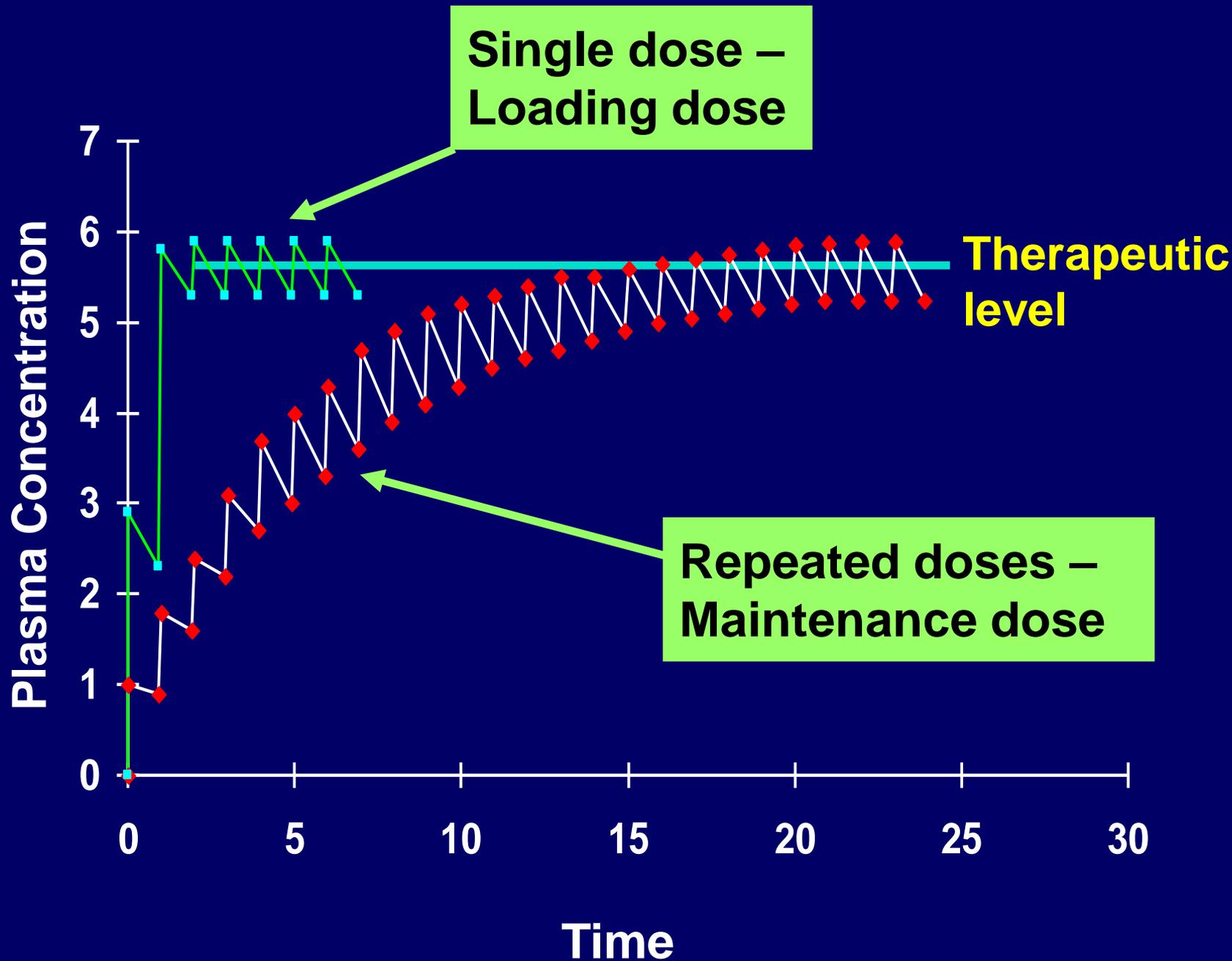
Dose (Rate of Administration) = clearance x plasma conc.

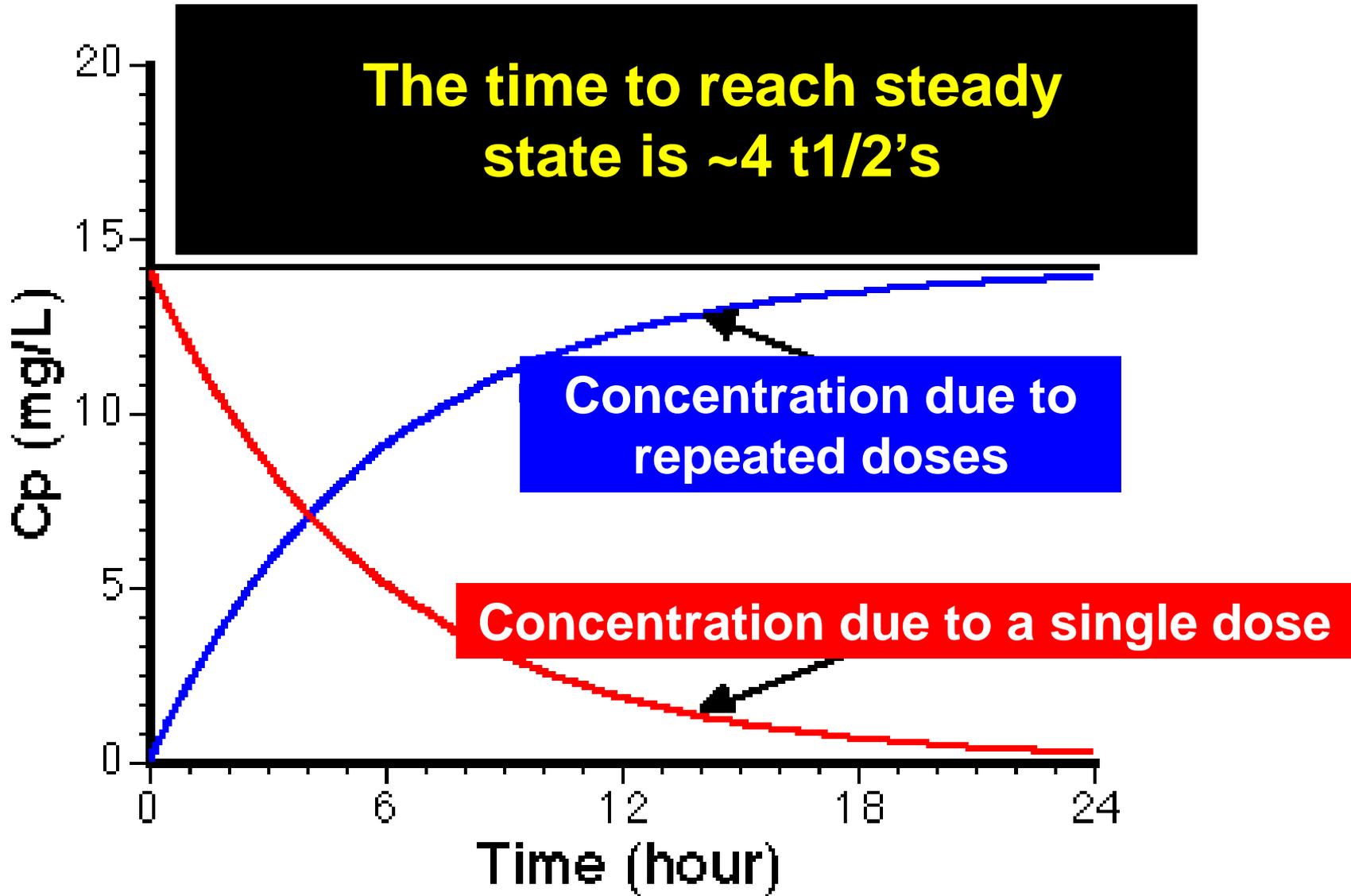
Or

If you aim at a target plasma level and you know the clearance, you can calculate the dose required.

Constant Rate of Administration (i.v.)



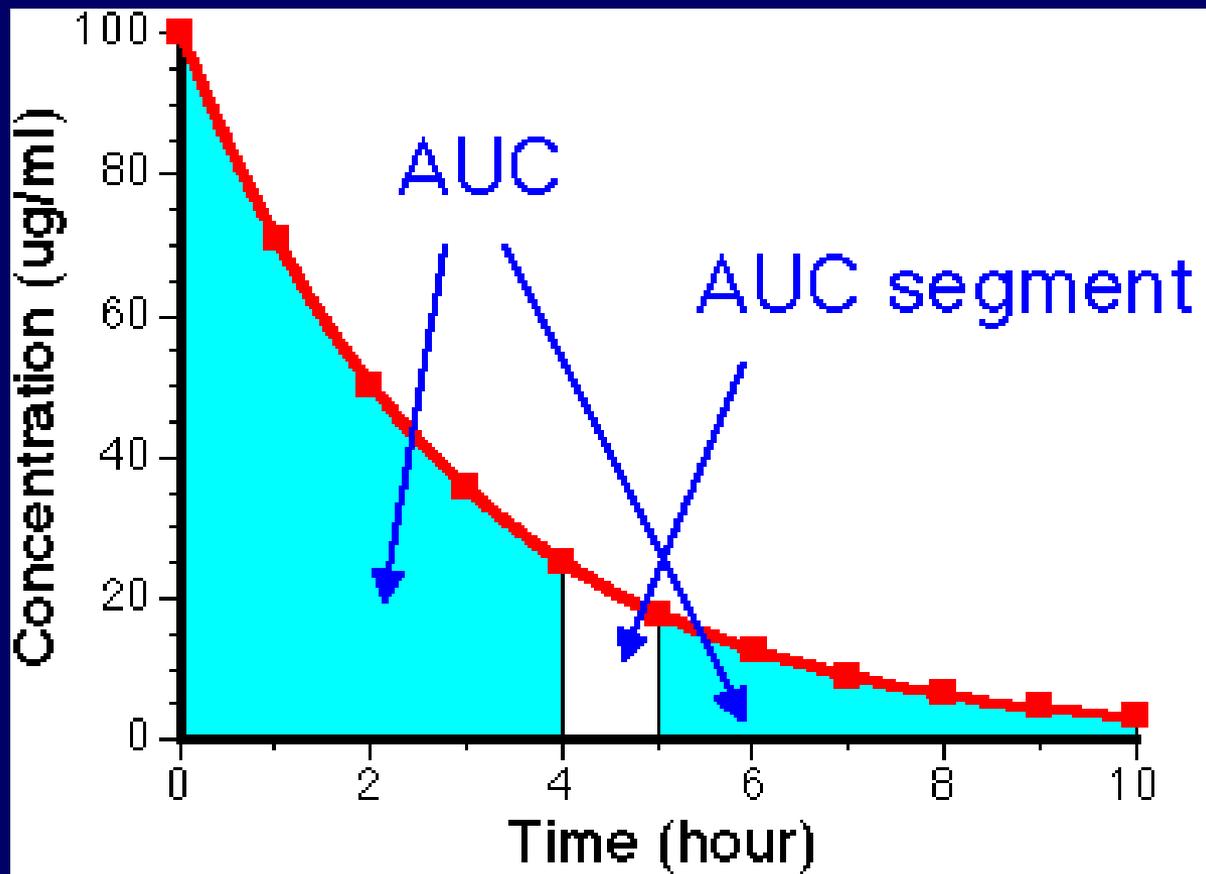




Pharmacokinetic parameters

Get equation of regression line; from it get K_{el} , C_0 , and AUC

- Volume of distribution $V_d = DOSE / C_0$
- Plasma clearance $Cl = K_{el} \cdot V_d$
- plasma half-life $t_{1/2} = 0.693 / K_{el}$
- Bioavailability $(AUC)_x / (AUC)_{iv}$



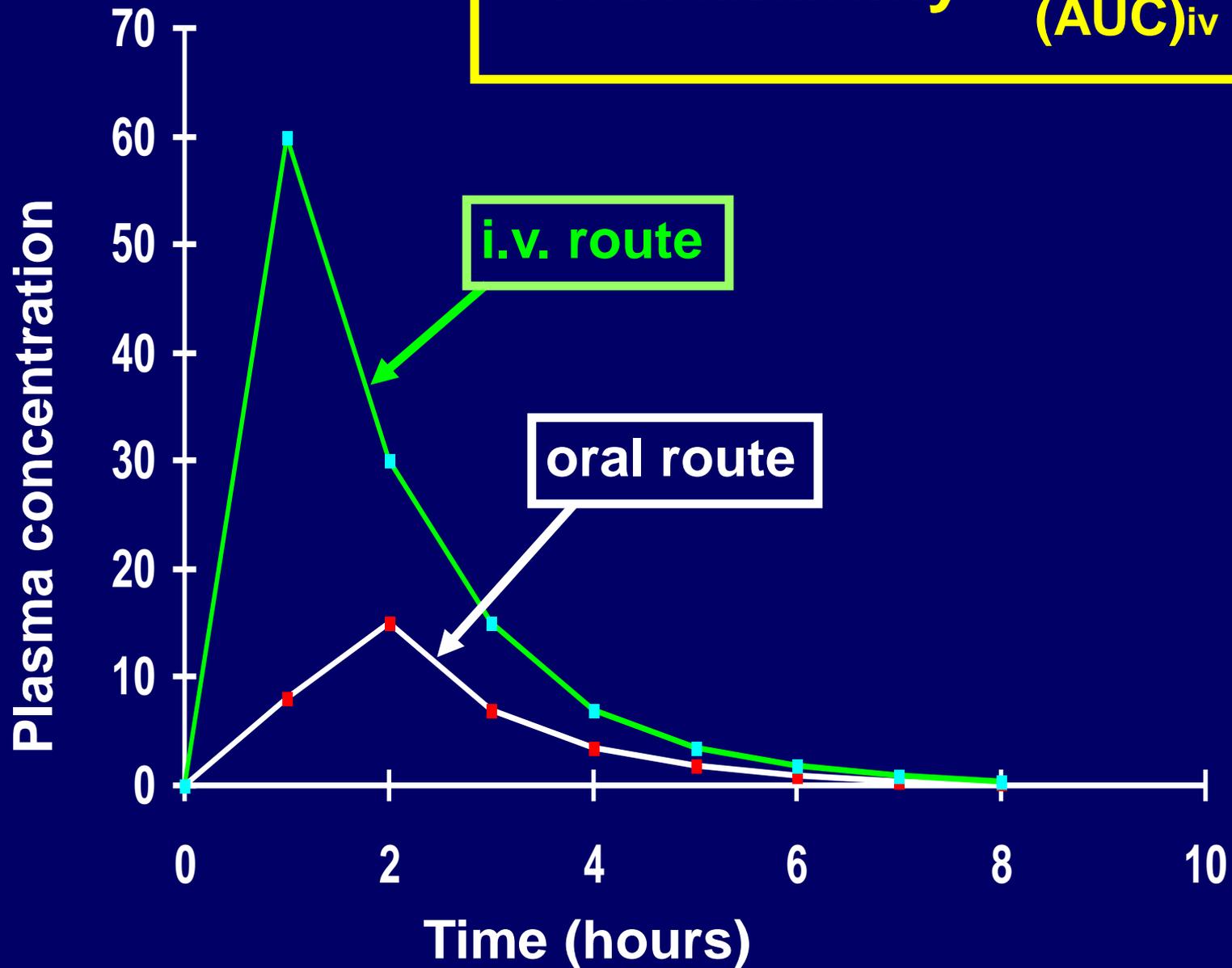
$$dC/dt = CL \times C$$

$$dC = CL \times C \times dt$$

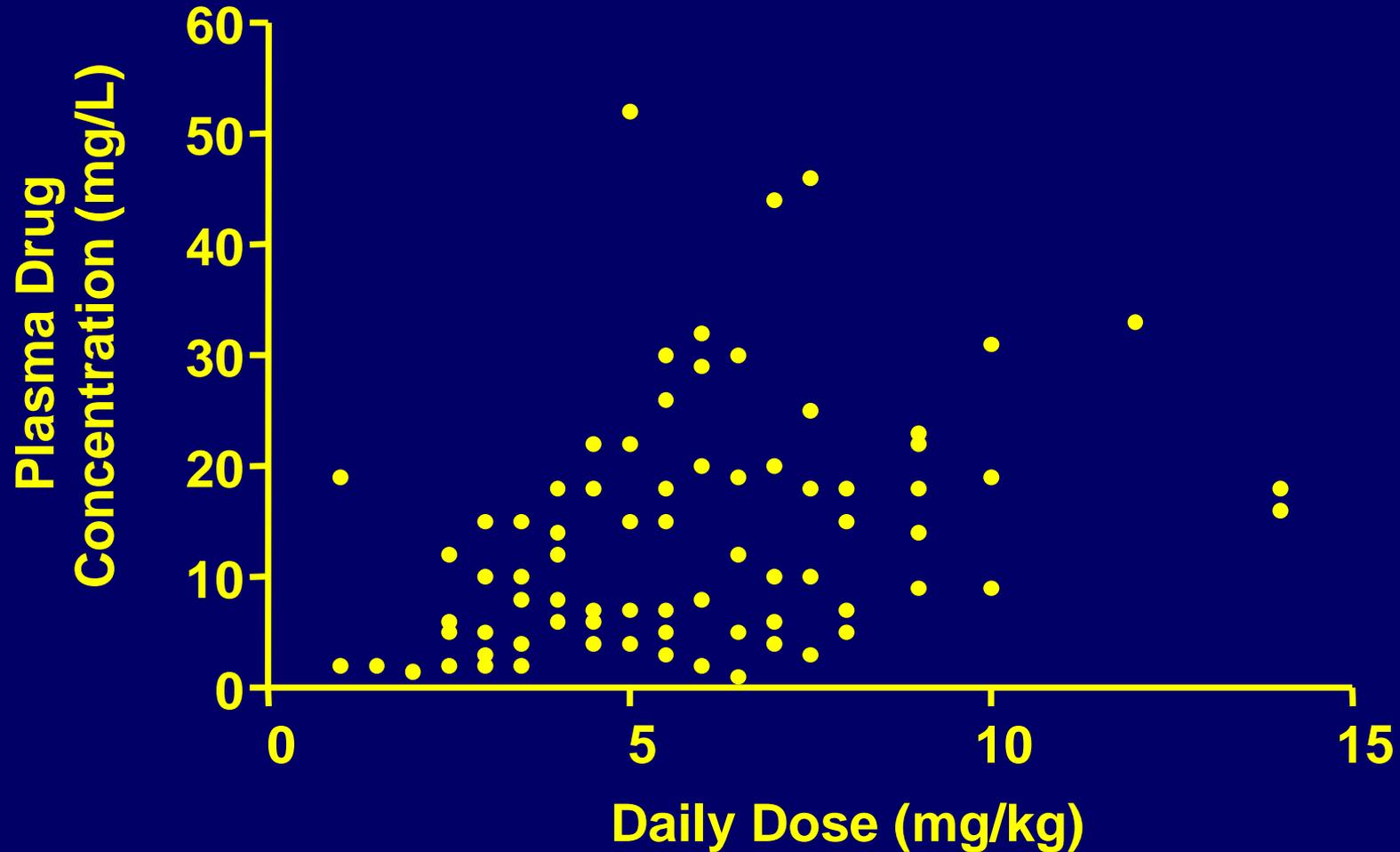
But $C \times dt =$ small area under the curve. For total amount eliminated (which is the total given, or the dose, if i.v.), add all the small areas = AUC.

$$\text{Dose} = CL \times \text{AUC}$$

$$\text{Bioavailability} = \frac{(AUC)_o}{(AUC)_{iv}}$$



Variability in Pharmacokinetics



PRINCIPLE

The absorption, distribution and elimination of a drug are qualitatively similar in all individuals. However, for several reasons, the quantitative aspects may differ considerably. Each person must be considered individually and doses adjusted accordingly.