

1- The precursor amino acid of adrenaline
is?:
a. Histidine
b. Tyrosine
c. Glutamate
d. Tryptophan
e. Alanine
2- The bonding of unit molecules to produce a
polysaccharide is called _? :
a. condensation
b. translation
c. cellular respiration
d. hydrolysis
e. degradation

- 3- A solution with pH = 5 is $_$ __ than a solution with pH = 7?
- a. 2 times more basic
- b. 10 times more basic
- C. 10 times more acidic
- d. 100 times more acidic e. 1000 times more basic
- 4- The amino acid which contains a free sulfhydryl group (-SH) in its side chain "R group" is ___?
- a. Cysteine b. Leucine c. Methionine d. Tyrosinee. Cystine
- 5- For solutions and buffers, choose the WRONG statement?
- a buffer is a solution with a constant pH
- b. solutions made up of solvents and solutes
- C. solutions are non-homogenous mixtures
- d. all buffers are solutions

e. in solutions, the solvents can be gases

6- Which one of the following solutions has stronger acidity?

a.
$$H = 10^7$$

b.
$$pH = 9$$

d.
$$pH = 10^4$$

e.
$$pH = 8$$

7- In an a helix, choose the CORRECT statement?

- a. side chain residues point up and down the axis of the helix
- b. the helix is right-handed
- c. there are five residues per helical turn
- d. the oxygen of the carbonyl carbon in a peptide bond points out toward the exterior of the helix

- e. there are usually many proline residues present
- 8- In an Alpha-turn, the end residues are separated by peptide bonds?
 - a. two b. one C. five d. three e. four
- 9- Which of the following has high helix-forming propensities?
- a. glycine b. proline c. alanine d. tryptophan
- e. tyrosine
- 10- The functional group NH2 is a/an
- a. amino group b. phosphate group
- c. hydroxyl group d. carbonyl group
- e. carboxyl group
- 11- There is an overlapping between biochemistry

- and one of the following subjects of science?
- a. Pharmacology b. Toxicology c. Physiology
- d. Biology e. Pathology
- 12- Which one of the following compounds is optically active?
- a. L-glycine
- b. D-methionine
- c. B-alanine
- d. GABA
- e. cystine
- 13- In an Alpha-turn, the end residues are separated by peptide bonds?
- a. two b. one C. five d. three e. four
- 14- The bonding of unit molecules to produce a polypeptide is called?

- a. hydrolysis
- b. translation
- c. cellular respiration
- d. Dehydration synthesis
- e. degradation
- 15- The amino acids: lysine, alanine, glutamic acid and selenocysteine share one common feature from the following?
- a. Standard amino acids
- b. Non-standard amino acids
- c. Proteinogenic amino acids
- d. Non-proteinogenic amino acids with protein role
- e.Non-proteinogenic amino acids with non-protein role

- 16- The one-letter code of amino acids sequence: V-N-L-K-Y-W-W-D-A-0 matches one of the following three-letter code of amino acids sequence?
- a.Val-Asn-Leu-Lys-Tyr-Trp-Trp-Asp-Ala-Gln
- b. Val-Asp-Lys-Leu-Tyr-Trp-Trp-Asn Ala-Glu
- c.Pro-Asp-Lys-Leu-Tyr-Trp-Trp-Asn-Ala-Glu
- d.Val-Asn-Leu-Lys-Tyr-Trp-Trp-Asp-Ala-Glu
- e.Val-Asn-Leu-Lys-Tyr-Trp-Trp-Asn-Ala-Gin
- 17- The molecular formula for glycine is CaHsO2N. What would be the molecular formula for linear oligomer made by linking ten glycine molecules together by condensation synthesis?
- a. C20 H50 O20 N10
- b. C20 H32 O11 N10
- c. C20 H40 O10 N10
- d. C20 H68 O29 N10

- e. C22 H68 O29 N10
- 18- One of the followings is a calcium sensor subunit:
- a. Tnl
- b. TnC
- C. myosin head
- d. TnT
- e. tropomyosin
- 19- The sugar found in RNA is:
- a. Deoxyribose b. Ribose c. Ribulose
- d. Erythrose e. Sucrose
- 20- There are several levels of protein structure, the most complex of which is ?
 - a. primary b. secondary c. tertiary d. quaternary
 e. alpha helix

- 21- In the formation of a macromolecule, what type of bond would join two monosaccharide's subunits?
- a. peptide bond
- b. glycosidic bond
- C. phosphodiester bond
- d. ionic bond
- e. hydrogen bond
- 22- One group of these sciences can lead to the understanding of the basis of biochemistry?
- a. Anatomy, physiology, physics and immunology
- b. Chemistry, pharmacology, biology and pathology
- c.Biology, chemistry, physiology and anatomy
- d. Biology, immunology, anatomy and microbiology

- e. Chemistry, pathology, anatomy and pharmacology
- 23- The amino acid which contains a thioether group in its side chain "R group" is:
- a. cysteine b. leucine c. methionine d. tyrosine
- e. cysteine
- 24- If you are working in the biochemistry laboratory and a chemical gets into your mouth, you should _____?
- a. Spit it out
- b. Rinse your mouth
- C. Visit a doctor
- d. All the statements are true
- e. Tell your demonstrator

- 25- If you have a 9M solution of Al(OH)3, then the normality of this solution is equal to_?
- a.3 b. 27 C. 4.5 d. 0.33 e. 1.33
- 26- Enantiomers are all of the followings EXCEPT?
- a. pair of stereoisomers
- b. optically active
- c. chiral molecules
- d. mirror images
- e. superimposable
- 27- A process by which a protein structure assumes its functional shape or conformation is?
- a. denaturing b. folding C. synthesis
- d. hydrolysis e. aggregation
- 28- Which of the following pairs of amino acids might contribute to protein conformation by

forming electrostatic interactions?

- a. Glycine and leucine
- b. Glutamate and lysine
- C. Phenylalanine and tyrosine
- d. Lysine and arginine
- e. Tyrosine and aspartiC

- 29- The pH value at which the molecule is neutral is called:
- a. zwietterion
 b. isoelectric point
 C. ampholyte
 d. anionic
 e. cationic
- 30- For 3N solution of H2S04, the Molarity of this?

- b. 1.5
- c. 0.67
- d.0.75
- **e.3**
- 31- The stronger the acid (choose the correct answer)?
- a. The higher the pKa
- b. The higher the pH
- c. The lower the Ka and pKa
- d. The higher the OH concentration
- e. The higher the Ka

- 32- Given that pk1=2.3, pk2=9.7 then the dominant form of leucine at pH=6 is?
 - a. Neutral b. Positively charged

- c. Negatively charged d. Cationic e. Anionic
- 33- Which one of the followings is the initiation codon?
- a. AUG b. GUA c. UAG d. UGA e. UUA 62-
- 34- The sugar found in DNA is?
- a. Deoxyribose b. Ribose C. Ribulose d. Erythrose
- e. Sucrose
- 35- Which of the following pairs of amino acids might contribute to protein conformation by forming disulfide bond?
- a. Cysteine and cysteine
- b. Glutamate and lysine
- C. Phenylalanine and tyrosine
- d. Lysine and arginine

e. Glycine and leucine

- 36- The.... is an essential amino acid and required for the synthesis of tyrosine?
- a. Arginine b. Lysine C. Tryptophan
- d. Phenylalanine e. Tyrosine
- 37- For 5N solution of H2s04, the Molarity of this solution is equal to?
- a. 6
- b. 1.5
- c. 0.67
- d. 0.75
- e. 2.5
- 38- The neutral form of amino acid is called _?
- a. Zwietterion b. Isoelectric point C. Ampholyte

d. Anionic e.Cationic

- 39- There are several levels of protein structure, the most simple of which is?
- a. Primary b. Secondary C. Tertiary d. Quaternarye. Beta sheet
- 40- The process of returning a denatured protein structure to its original structure and normal level of biological activity is ____?
- a. Denaturing b. Synthesis C. Renaturation
- d. Hydrolysis e. Aggregation
- 41- The one-letter code of amino acids sequence: [M-N-L-H-Y-D-K-Q-A-R] matches one of the following three-letter code of

amino acids sequence?

a. Met-Asp-Leu-His-Tyr-Asp-Lys-Gin-Ala-Arg

- b. Met-Asn-Leu-His-Tyr-Asn-Lys-Gln-Ala-Arg
- c. Met-Asn-Lys-His-Tyr-Asp-Leu-Gln-Ala-Arg
- d. Met-Asn-Leu-His-Tyr-Asp-Lys-Gln-Ala-Arg
- e. Met-Asn-Leu-His-Trp-Asp-Lys-Gln-Ala-Arg

- 42- Suppose that the acid (CH3COOH) has a pKa = 7.76 was placed in a solution that has a pH=4.25, the dominant form of this acid in the solution will be?
- a. CH3COOH
- b. CH3COOH2
- C. CH3CH+
- d. CH3CO0-
- e. CH3CO-
- 43- What should you maintain in your lab work

area?

- a. A messy work space
- b. A clean and tidy work space
- C. Chemicals spread all around you
- d. Lots of food and drink
- e. Use cell and head phones
- 44- Regarding pH, pKa and Ka, choose the correct answer?
- a. The higher the pH the stronger the acid
- b. The higher the pKa the stronger the acid
- c. For acids, at pH higher than pKa more base than acid
- d. At pH lower than pKa more dissociation to acids
- e. No relation between strength of acid and Ka
- 45- The molecular formula for glycine is CoH5OpN.

What would be the molecular formula for a linear oligomer made by linking five glycine molecules together by condensation synthesis?

- a. C10 H25 O10 N5
- b. C10 H17 O8 N5
- C. C10 H18 O10 N5
- d. C10 H29 O19 N5
- e. C10 H17 O6 Ns
- 46- The amino acids: pyrrolysine, n-formyl methionine and selenocysteine share one common feature from the followings?
 - a. Standard amino acids and proteinogenic
- b. Non-standard amino acids and nonproteinogenic
- C. Non-proteinogenic amino acids with protein role

- d. Non-proteinogenic amino acids with nonprotein role.
- e. Proteinogenic amino acids added to polypeptide chain by unique mechanism
- 47- If you have 8M solution of Ca(OH)2, the Normality of this solution is equal to...?
- a. 4 b. 16 c.0.25 d. 1.25 e.
- 48- For the peptide bond, choose the INCORRECT answer?
- a. It is a Trans bond
- b. It is a Cis bond
- c. It is rigid and planar
- d. It is a covalent bond
- e. It has a partial double bond character
- 49- For Myoglobin, choose the WRONG statement?
 - a. Myoglobin Can't carry C02

- b. Myoglobin 02 affinity is higher than hemoglobin02 affinity
- C. Myoglobin has cooperativity of 02 binding
- d. Myoglobin has No quaternary structure
- e. Myoglobin is found in muscles
- 50- In an alpha helix, choose the CORRECT statement?
- a. There are usually many Glycine residues present
 b.Side chain residues point up and down the axis
 of the helix
- C. The helix is always left-handed
- d. There are 3.6 residues per helical turn
- e. The oxygen of the carbonyl carbon in a peptide bond points out toward the exterior of the helix

- 51- One of the followings is a tropomyosin binding subunit?
 - a. Elastin b. TnC C. Myosin head d. TnT
 - b. kl e. Collagen
- 52- Which one of the followings prefers to adopt B -strand conformations?
- a. Leucine b. Proline C. Alanine d. Methionine
- e. Tyrosine
- 53- In a delta-turn, the end residues are separated by ___ peptide bonds?
 - a. Two b. One c. Five d. Three e. Four
 - 54-75% of Myoglobin structure is a-helix in ____ regions?
 - a. Seven b. Five c. Six d. Nine e. Eight

- 55- Given that pk1=2.3, pk2-9.7 then the dominant form of leucine at pH=1 is?
- a. Zwitterion b. Neutral C. Negatively charged
- d. Cationic e. Anionic
- 56- The functional group OH is a /an?
- a. amino group b. phosphate group
- C. hydroxyl group d. carbonyl group e. carboxyl group
- 57 In the formation of a macromolecule, what type of bond would join two amino acids subunits? a. Peptide bond
- b. Glycosidic bond
- C. Phosphodiester bond
- d. lonic bond e. Hydrogen bond
- 58- For solutions and buffers, choose the WRONG statement?
- a. A buffer is a solution with a constant pH

- b. Solutions made up of solvents and solutes
- C. Solutions are homogenous mixtures
- d. All solutions are buffers
- e. In solutions, the solvents can be gases
- 59- Enantiomers are one of the followings?
- a. Pair of constitutional isomers
- b. Optically active
- C. Achiral molecules
- d. Not always mirror images
- e. Superimposable
- 60- The precursor amino acid of adrenaline is ___?
- a. Histidine b. Tyrosine C. Glutamate d. Tryptophan
- e. Alanine

- 61- The bonding of unit molecules to produce a polysaccharide is called _?
- a. condensation
- b. translation
- C. cellular respiration
- d. hydrolysis
- e. degradation
- 62- Choose the correct statement Select one:
- a. myosin is a tetramer protein
- b. tropomyosin protein binds head to tail with actin thin filament
- C. troponin complex contains five subunits
- d. myoglobin protein has the quaternary structure level
- e. the heavy chain of myosin contains four

domains

- 63-The glucose is the main energy substrate in? Select one:
- a. Brain cells
- b. Skeletal muscles
- C. Liver cells
- d. Heart cells
- e. Kidney cells

64-If the OH group at ONLY one chiral carbon in the linear structure of monosaccharides is located on the right or the left, then the resulting stereoisomer is assigned as? Select one:

- a. L-sugar
- b. D-sugar

c. Epimer d. a-sugar e. B-sugar 65 - C6H1206 is the molecular formula of all of the following sugars EXCEPT? Select one: a. Glucose b. Fructose c. Galactose d. Mannose e. Ribose 66- After the cyclic sugar formation, the carbonyl carbon will be converted to? Select one:

a. anomeric carbon

b. achiral carbon

c. alcoholic carbon

- d. carboxylic acid
- e. ketonic group
- 67- One of the following is not correct regarding cellobiose? Select one
- a. It is a reducing sugar
- b. This disaccharide results from degradation of celulose
- C. It consists of two B-glucose units
- d. The monomers in cellobiose are found in the configuration and as cyclic pyranose rings
- e. The glucose units are joined together by a-1,4 glycosidic bond
- 68-Many factors affect calcium absorption, Which of the following carbohydrate is effective in promoting the calcium absorption? Select one:
- a. Sucrose

b. Maltose C. Lactose d. Xylose e. Galactose 69- One of the following compounds is not considered as modified sugar? Select one: a. glucuronic acid b.glycerol C. deoxyribose d. glyceraldehyde e. glucosamine 70-This polysaccharide is hetero, natural, linear and mainly found in mast cells? Select one:

- a. Glycogen
- b. Hyaluronic acid
- c. Dermatan sulphate
- d. Heparin
- e. Chitin

71- Regarding Lactose Intolerance, one of the following is correct?

Select one:

- a. It is caused by deficiency of the sugar lactose in milk
- b. It has symptoms like constipation and fever
- c. due to deficiency of lactase enzyme, the lactose found in milk will be absorbed from the wall of small intestine intact
- d. GIT disturbances are resulted from undegraded lactose reaching the colon intact

e. Small babies are given the milk formula AR

- 72-Enantiomers are all of the followings EXCEPT?
- a. pair of stereoisomers
- b. optically active
- c. chiral molecules
- d. mirror images
- e, superimposable
- 73- The bonding of unit molecules to produce a polysaccharide is called
- a a, condensation
- b. b. translation
- c. cellular respiration
- d. hydrolysis
- e. degradation

1	В	11	Α	21	В
2	Α	12	E	22	
3	D	13	E	23	Α
4	Α	14	D	24	D
5		15	С	25	
6		16	Α	26	E
7	В	17		27	В
8	E	18	В	28	
9	С	19	В	29	В
10	9	20	D	30	

31	E	42		53	В
32	Α	43	В	54	E
33	Α	44		55	D
34	Α	45		56	С
35	Α	46	E	57	Α
36	Α	47		58	
37		48	В	59	В
38	Α	49	В	60	В
39	Α	50	D	61	Α
40	С	51	D	62	В
41	D	52	E	63	Α

64	С
65	E
66	Α
67	E
68	С
69	D
70	D
71	D
72	E
73	A5