

# Polysaccharides

"Glycans"  $\Rightarrow$  Polymeric molecules  
 $\Rightarrow$  long chains of monosaccharides that bound together via glycosidic bonds

Polysaccharides  $\rightarrow$  homo Polysaccharides  $\Rightarrow$  one type of monosaccharides  
 $\rightarrow$  hetero Polysaccharides  $\Rightarrow$  more than one type " "  
 branched      linear

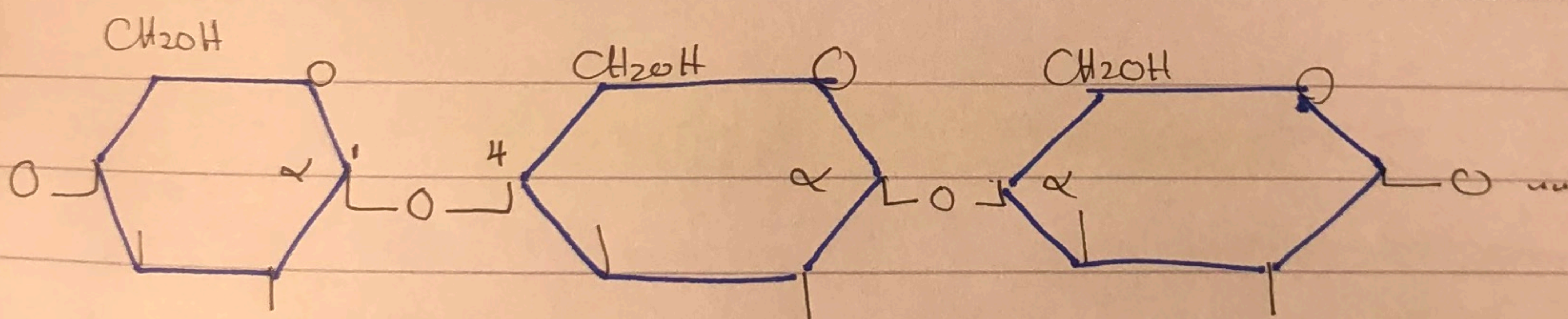
$\Rightarrow$  For storage "Starch - glycogen"

$\Rightarrow$  ~~For~~ Structural "Cellulose, chitin"  
 لحد النسيج

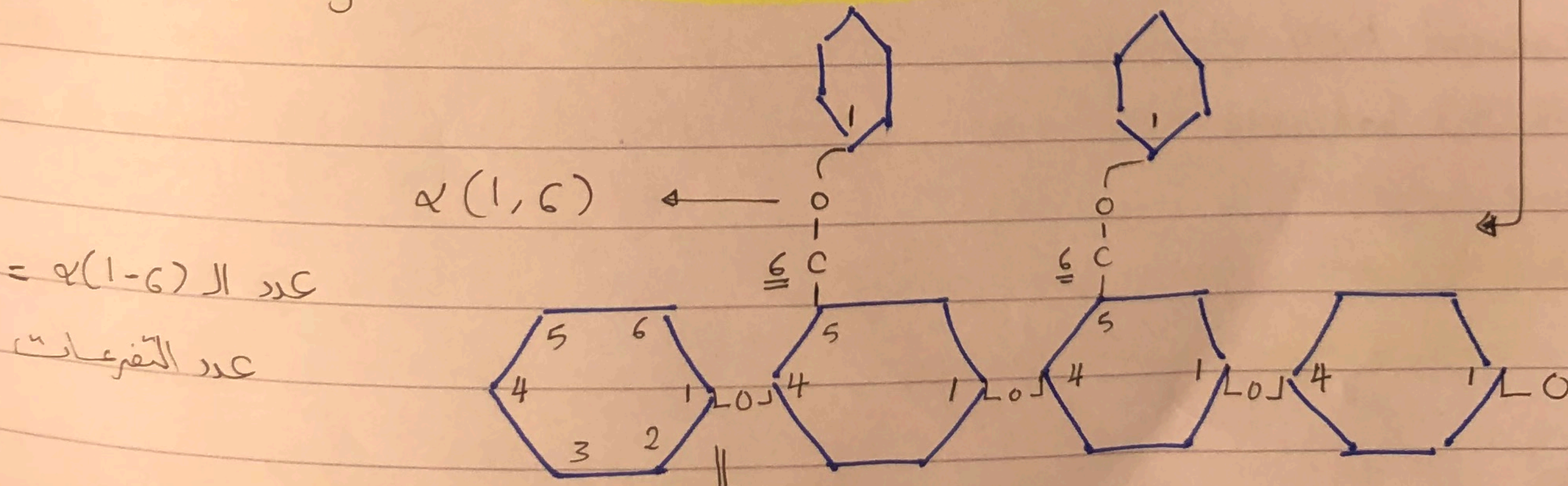
■ Starch: Storage Polysaccharide in plants "hollow helix"

$\Rightarrow$  glucose monomers "homopolysaccharide"

$\Rightarrow$  20% amylose "soluble"  
 80% amylopectin "insoluble"



Amylose =  $\alpha(1,4)$  glycosidic bonds.



عدد ال (1-6)

عدد التفرعات

Amylopectin  $\Rightarrow \alpha(1,4)$

~~Amylopectin~~  $\Rightarrow \alpha(1,6)$  only on the branch point



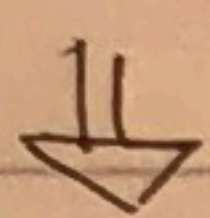
## - Digestion of starch -

[1] Salivary amylase enzyme (in the mouth) randomly hydrolyses the  $\alpha(1,4)$  bonds.

[2] Oligosacchrides continue to the small intestine by Pancreatic amylase

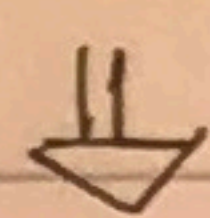
[3] Further hydrolysis by

$\alpha$ -glucosidase



Removes one  
glucose residue at time

debranching enzymes



hydrolyzes  
 $\alpha[1,6]$  bonds

[4] The produced monosacchrides are absorbed and transported to blood

## - Storage Polysacchrides -

Amylopectin

=> in plants

=> Branched every "(24 → 30)"

Glycogen

=> Animals and humans

=> Highly branched (8-14)

=> Found in skeletal muscle

up to (1.2% of muscle mass)

(10% of liver mass)



give glucose  
to organs

Same glyco bonding



## " Synthesis & Break down of Glycogen "

□ Tissues like brain cells need a constant supply of blood glucose for survival. "all time"

□ Some tissues like liver, skeletal muscles stores glucose in a form that can be easily mobilized for example "Glycogen"

□ → **glycogenesis:**

⇒ blood glucose is high

⇒ Li.

↳ **glycogenolysis:**

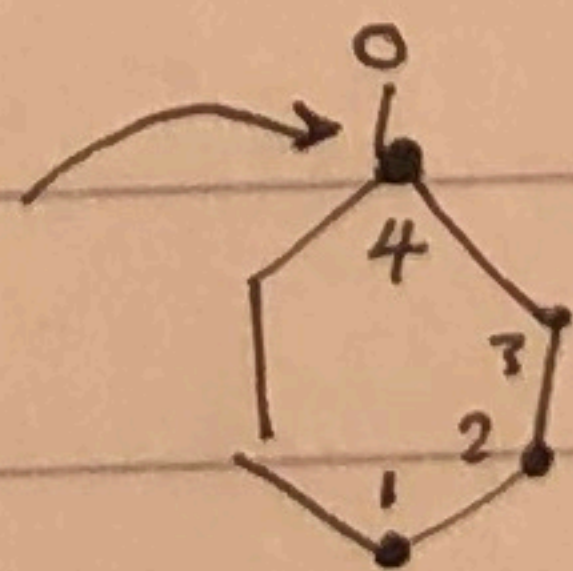
⇒ to release glucose when blood glucose sugar is low

"normal is " 80-100 mg/dl"

⇒ The balance between the need and availability is "metabolic homeostasis"

## " Storage Polysaccharides "

⇒ Branching in Glycogen & starch have non-reducing ends.

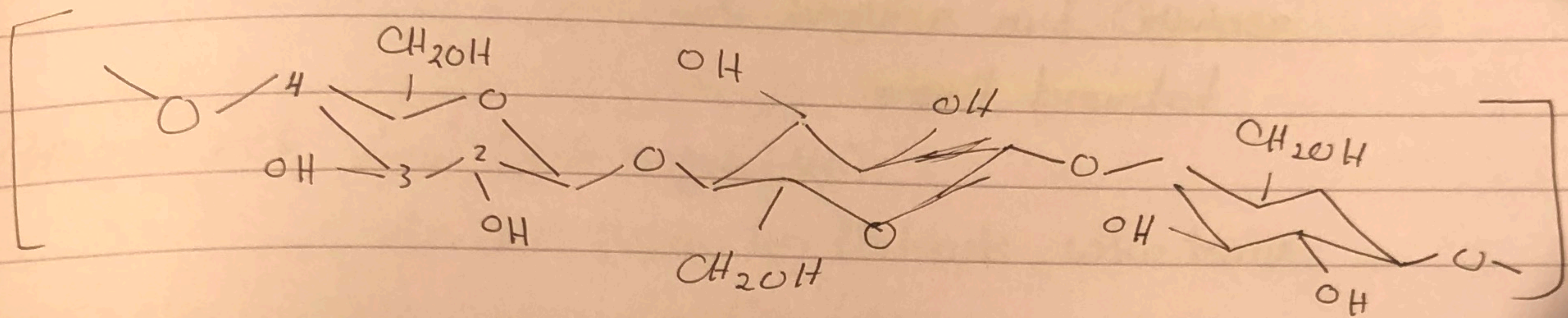


⇒ The main linear part have only one reducing ends.



## Structural Polysaccharides

**Cellulose**: Structural component of plant cell walls  
=> linear polymer of D-glucose  
=> linked via  $\beta(1-4)$  glycosidic bonds



Cellulose

- The most abundant organic molecule  
=> over half of the Carbon in biosphere
- So different from starch architecturally.  
"hollow helix" due to its  $\beta$ .
- Forms very long straight chains  
=> Parallel chains interact with each other through H bonds

Herbivores & termites	humans
=> Can digest Cellulose easily	=> Can't digest Cellulose
=> They have the enzymes capable of hydrolyzing the $\beta(1-4)$ bonds	=> humans can digest $\alpha(1-4)$ bonds but not $\beta(1-4)$ .

الزائدة الوحيدة كانت مسؤولة  
قديمًا عن هضم  $\beta(1-4)$  ولكن  
مع تطور الإنسان توقفت عن  
العمل.



## Structural Polysaccharides

The most branched  $\Rightarrow$  Glycogen  
 $\Rightarrow$  Amylopectin  $\left. \begin{array}{l} \text{Then} \\ \text{Randomly} \end{array} \right\}$   
 $\Rightarrow$  Amylose and Cellulose  
aren't branched

Cellulose rich food (like vegetables)

$\Rightarrow$  are given to patients who have  
Constipation.

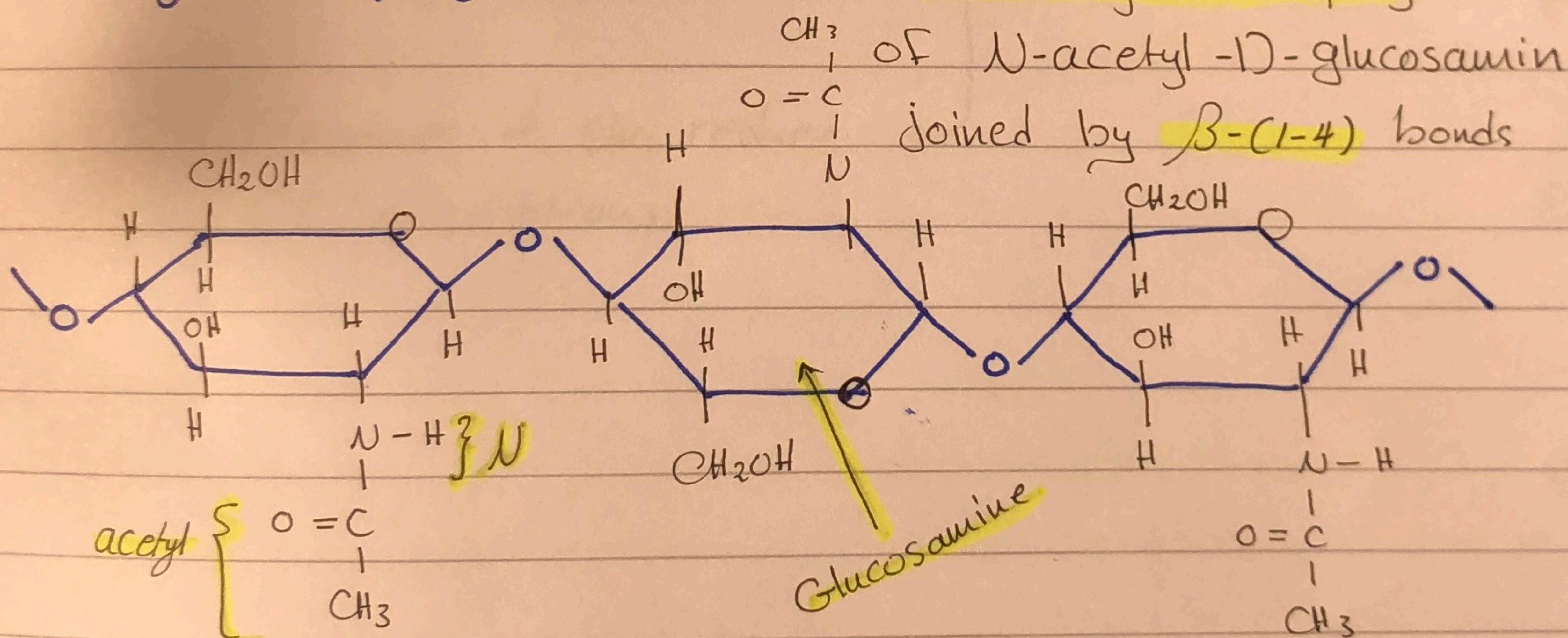
Chitin  $\Rightarrow$  homopolysaccharide

$\Rightarrow$  The structural component of the exoskeletons of invertebrates

$\Rightarrow$  Main component of the cell wall of fungi

$\Rightarrow$  long chain polymer

$\Rightarrow$  it's a long chain polymer



$\Rightarrow$  N-acetyl  
group

$\Rightarrow$  It has similar structure to cellulose  
they only differ in the replacement of  
OH at C2 of each monomer with acetyl amine group



## " Structural Polysaccharides "

### => " Chitosan "

=> linear Polysaccharide Composed of Randomly distributed  $\beta$ -(1-4)-linked D-glucosamine & N-acetyl-D-glucosamine

↓ "acetylated unit"  
↓ "deacetylated unit"

=> Produced Commercially

By deacetylation of chitin " Shrimp shells are used " with alkali sodium hydroxyl.

Used For:

weight loss and obesity treatment } why?

Because it can reduce fat absorption.



## " Heteropolysaccharides "

=> Two or more different monosaccharides.

=> associated with lipid or protein

↓ ↓  
glycolipid glycoprotein.

=> They are naturally found in connective tissue

Cartilage tendons blood vessel walls

### Hyaluronic acid " Hyaluronate "

□ major component of joint fluid (synovial fluid)

lubricating agent

shock absorber

□ involved as tissue repair so, it's used as skin lotions containing sodium

hyaluronate to help dry skin caused by eczema

it's a linear polymer of the disaccharides

monomers

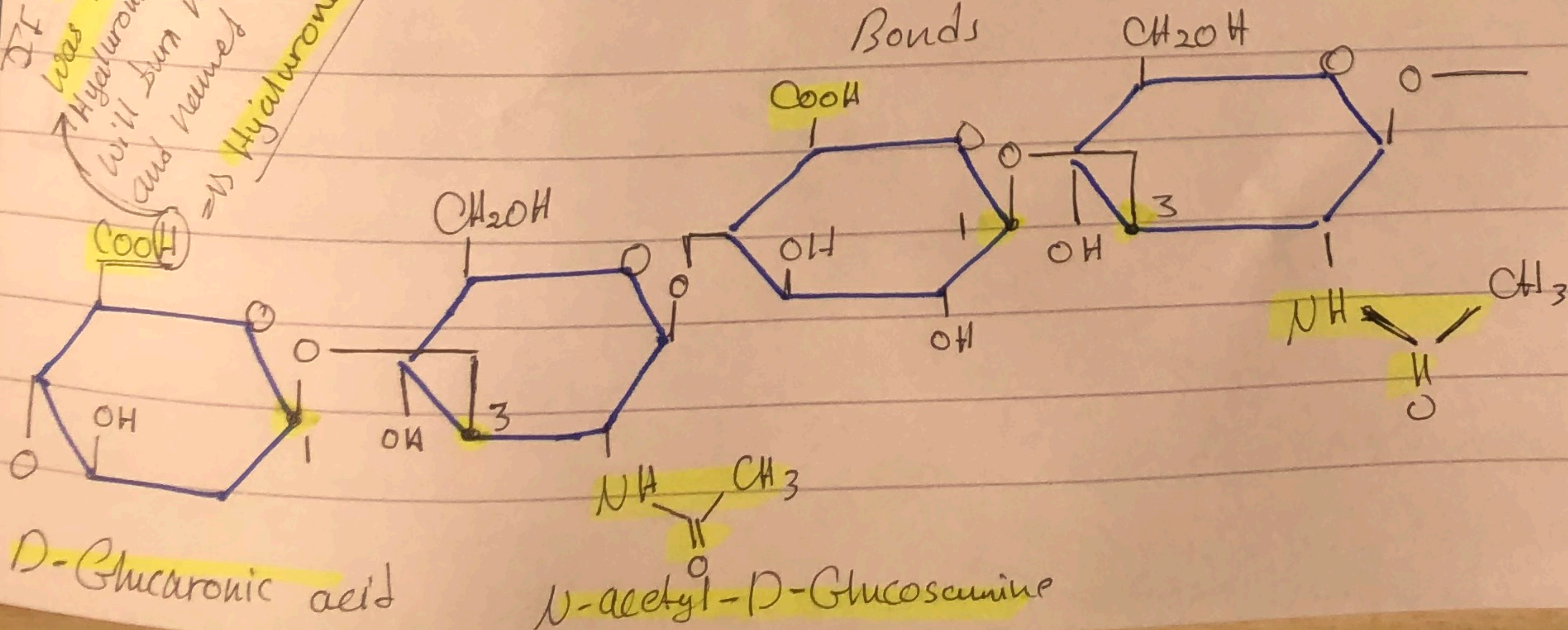
D-glucuronic acid & N-acetyl-D-glucosamine

linked via

$\beta$ -1-4 &  $\beta$ -1-3 bonds

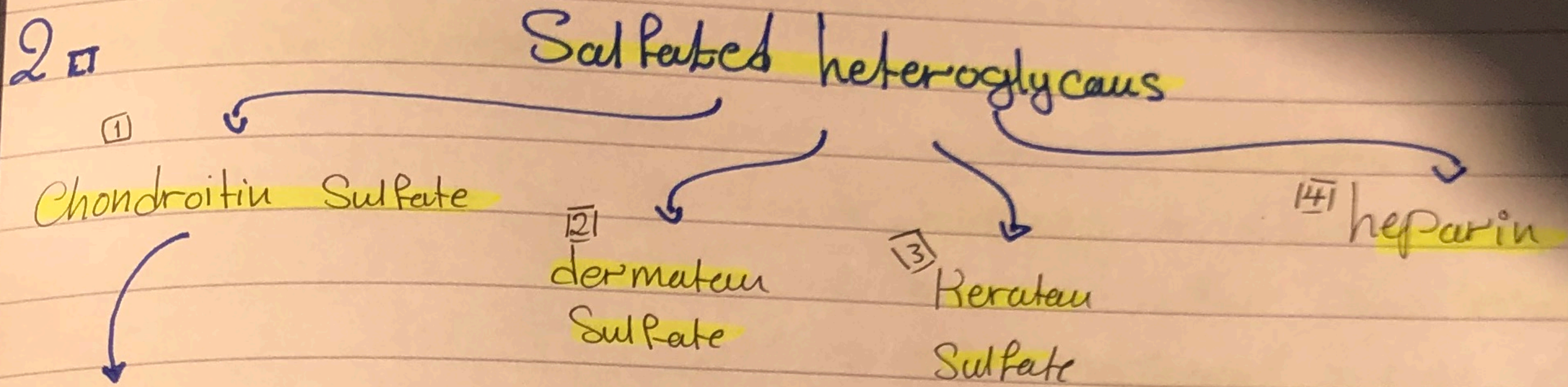
Bonds

If this # was lost the hyaluronic acid would be named => Hyaluronate





# "Heteropolysaccharides"



Chondroitin-4 Sulfate & Chondroitin-6-Sulfate (linear)  
 => Unbranched Polymers

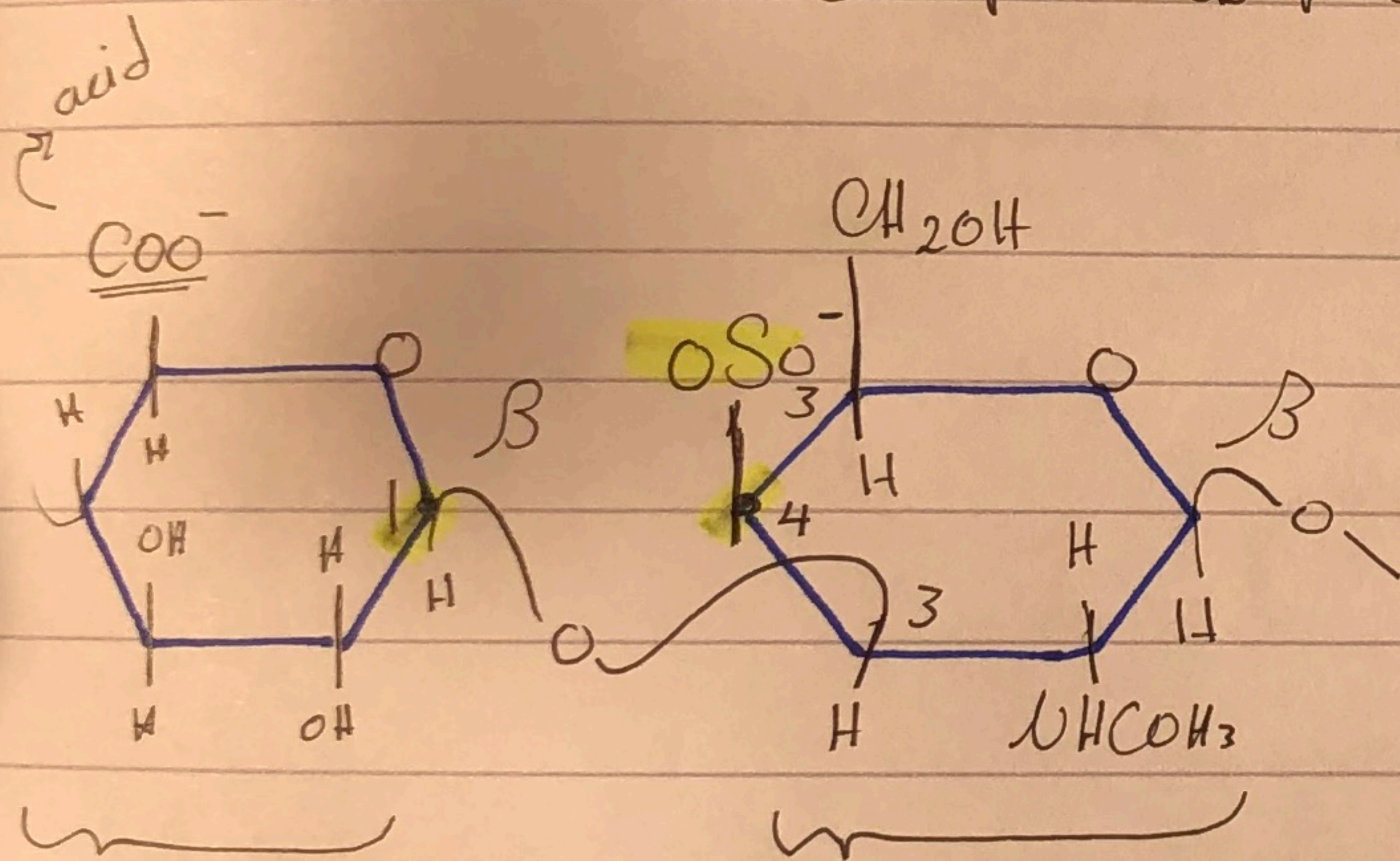
=> Contains

D-glucuronic acid  
 "D-Glucuronate"

N-acetyl-D-Galactosamine

=> OH Groups at position 4 or 6 being Sulfated

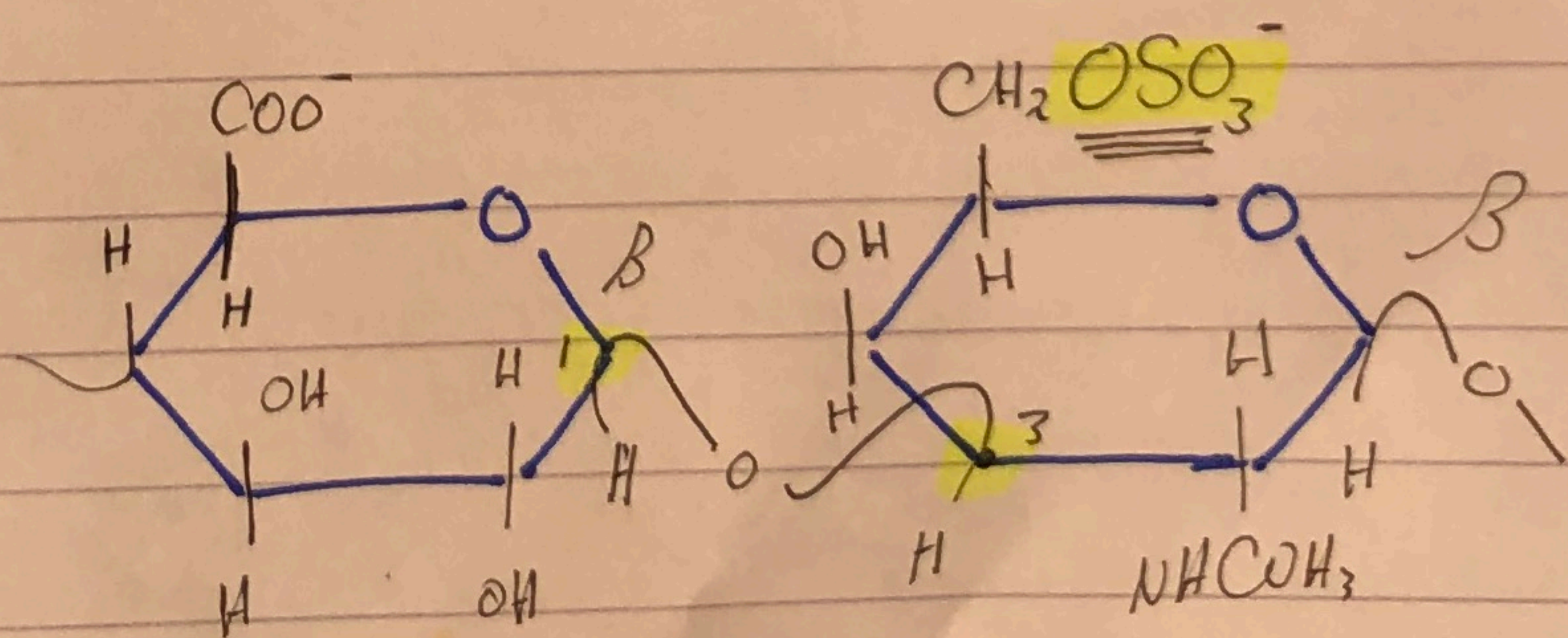
L>  $OSO_3^-$



D-Glucuronate

N-acetyl-D-Galactosamine  
 4-Sulfate

Note that the  
 the D-Glucuronate  
 isn't Sulfated



D-Glucuronate

N-acetyl-D-galactosamine  
 -6-Sulfate.



## " Hetero Polysaccharides "

### Chondroitin Sulfate :

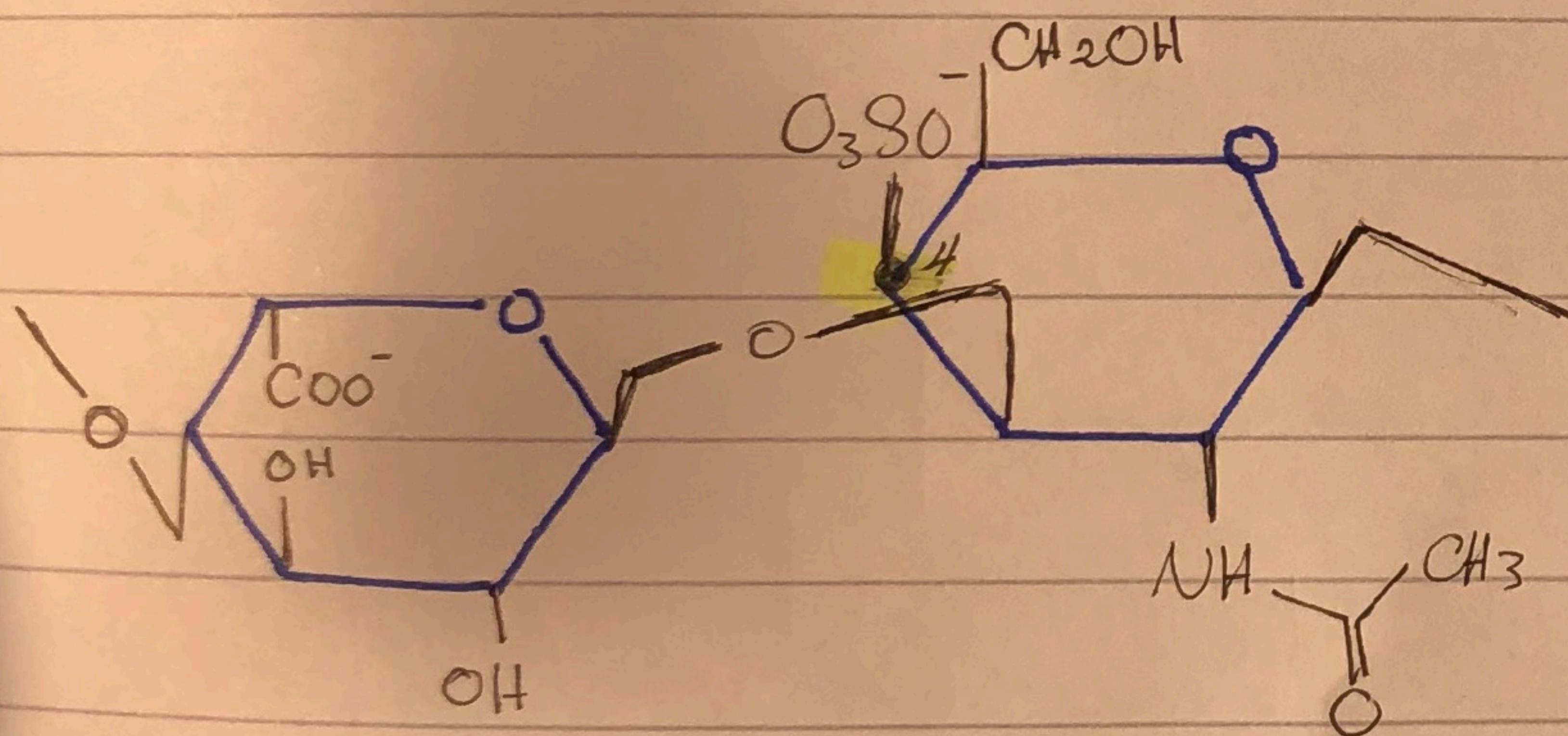
- => Major Component of Cartilages
- => Provide Resistance to Compression
- => loss of Chondroitin Sulfate Causes osteoarthritis
- => Used as dietary Supplement to treat  $\downarrow$   
Commonly Sold with glucosamine.

### 2) Dermatan Sulfate:

- => Natural Poly Saccharide found in skin
- => linear Polymer of disaccharides

L-Iduronic acid

N-acetyl-D-galactosamine-4-Sulfate



Dermatan Sulfate

L-Iduronate

N-acetyl-D-galactosamine-4-Sulfate



# Heteropolysaccharides

## [4] Keratan Sulfate:

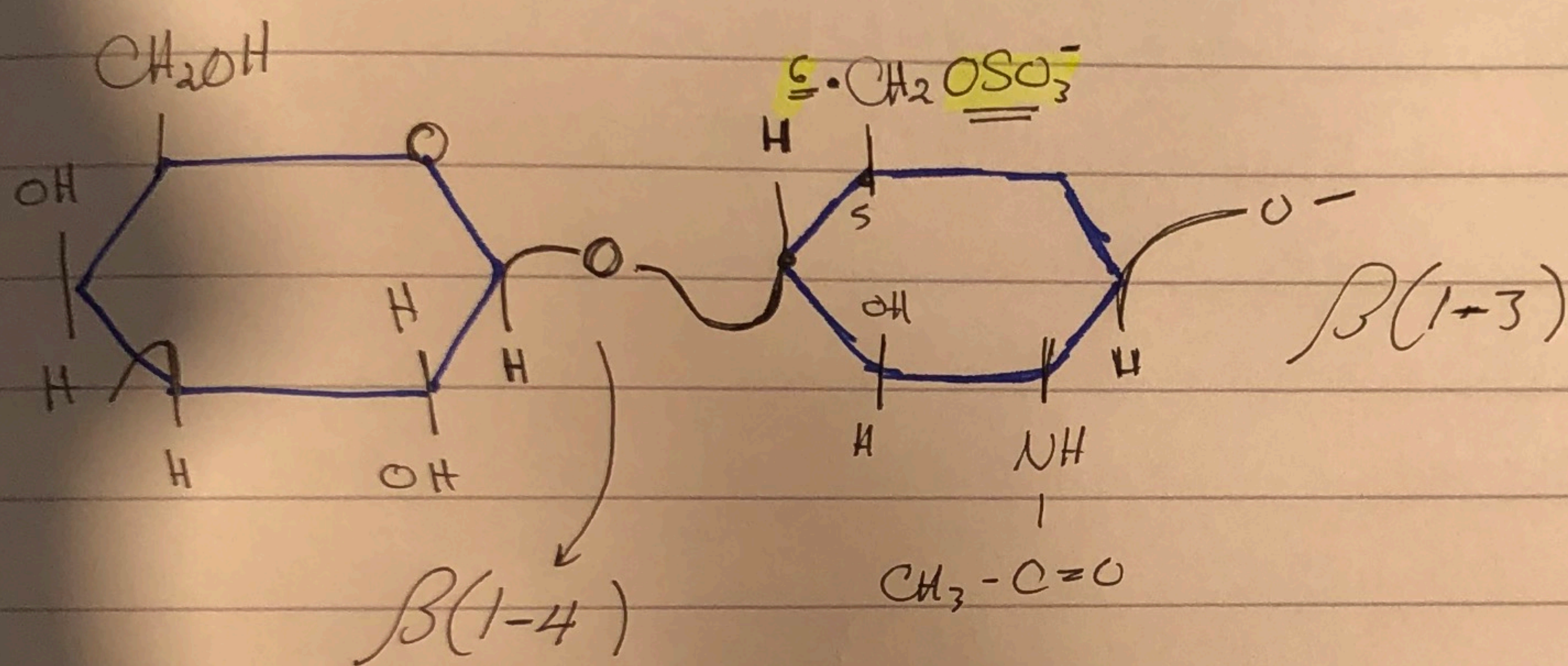
Keratan

- => Natural polysaccharide found in bones & cartilage
- => Highly hydrated molecules
- => linear polymer repeating disaccharide

Keratin = Protein

D-galactose

N-acetyl-D-glucosamine  
-6-Sulfate



D-galactose

N-acetyl-D-galactosamine-6-Sulfate

## [5] Heparin:

- => The most highly charged polymer of any biological molecule
- => Complex mixture of linear polysaccharides

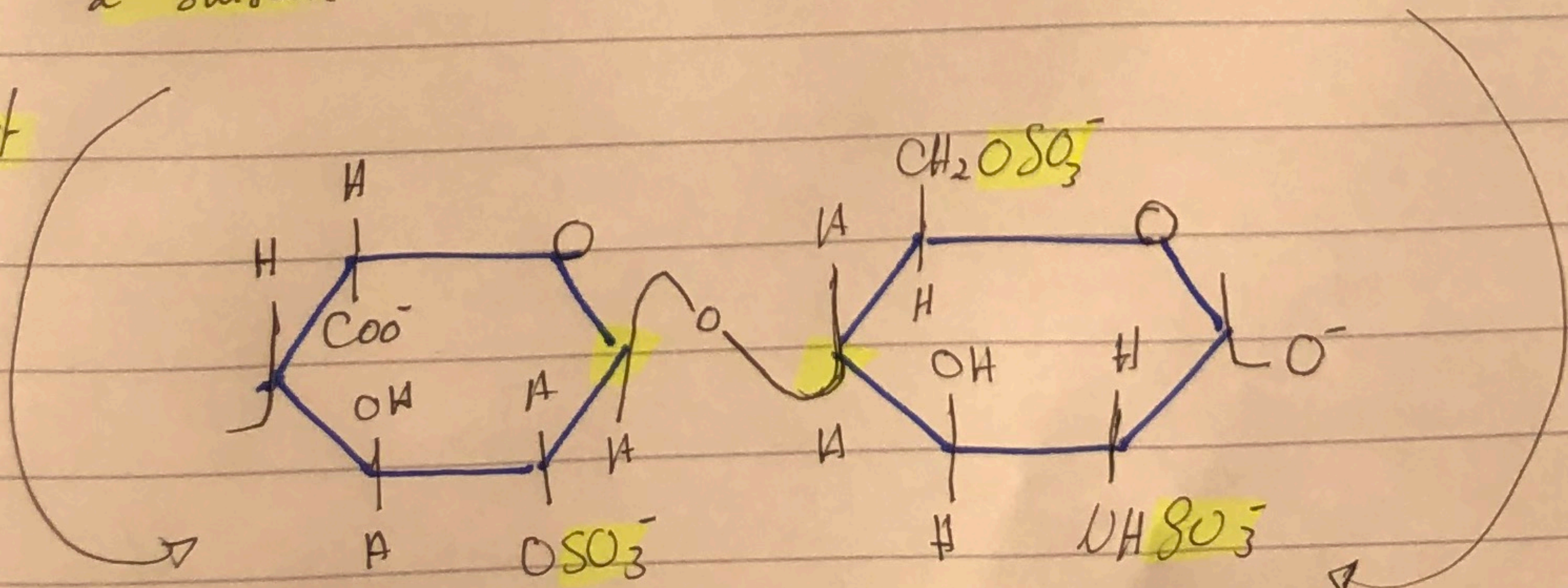
one example

is the  
sulfated

disaccharide unit

L-Iduronate  
2-Sulfate

N-Sulfo-D-glucosamine-6-Sulfate



Varies in  
the degree  
of sulfation  
of its sugar  
units.

- => Stored within the secretory granules of mast cells
- => Inhibits blood clotting => injectable anticoagulant